

*TYPHA LATIFOLIA* RESPONSE TO OLIGOTROPHIC AND EUTROPHIC NITROGEN  
AND PHOSPHORUS LOADING RATES UNDER LABORATORY CONDITIONS

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*TYPHA LATIFOLIA* RESPONSE TO OLIGOTROPHIC AND EUTROPHIC NITROGEN AND  
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M.A.Sc., 2013.

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## **Abstract**

*Typha latifolia* is an aggressive rhizomatous emergent wetland plant that can invade wetlands resulting in near monotypic *Typha* stands. *T. latifolia* is also one of the most commonly used macrophyte species in constructed wetlands. The hypothesis that elevated nitrogen and phosphorus concentrations observed in nonpoint source runoff increases *T. latifolia* fitness and potentially *T. latifolia* invasiveness was tested under semi-controlled laboratory conditions. A protocol was developed to propagate *T. latifolia* from seed in low P sediment to simulate an oligotrophic pre-impact reference treatment. Microcosms provided with hypereutrophic levels of P combined with oligotrophic or eutrophic levels of N had significantly greater shoot biomass and maximum leaf height compared to oligotrophic N and P treatment microcosms. These results indicated that high P often found in runoff may contribute to *T. latifolia* invasion. We recommend that noninvasive species of macrophytes be used in constructed wetlands to prevent impact to ecologically sensitive areas.

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## **Dedications**

I would like to dedicate this thesis to Manfred Praus who reveled in the outdoors and, in his own way, spiritually connected with all around him. Had he been given the opportunity to find his path, protection for the natural world would have been one of his pursuits. I would also like to dedicate this thesis to my mother who bestowed in me the confidence to prove my true capability.

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# 1 INTRODUCTION

## 1.1 Overview

The shift from diverse wetland ecosystem communities to monotypic stands of *Typha latifolia* and other *Typha* species has been documented since at least since the 1940's (Addy and MacNamara, 1948). *Typha* spp. invasions have often been attributed, in whole or in part, to cultural eutrophication (e.g., Moore *et al.*, 1989; McJannet *et al.*, 1995; Galatowitsch *et al.* 1999; Chow-Fraser, 2005; Shih and Finkelstein, 2008). Galatowitsch *et al.* (1999) however noted that the scientific evidence supporting the eutrophication hypothesis to explain *Typha* spp. invasions in the northern US and Canada was “scant”. The establishment and dominance of natural wetlands high in nutrients by *T. latifolia* (and other *Typha* species) is well documented (e.g., Grace and Wetzel, 1981a; Steward *et al.*, 1997; Chow-Fraser, 2005). However, as the contribution of other factors such as climate, habitat disturbance, water level and hydroperiod, competition and herbivory may contribute significantly to *T. latifolia* establishment and invasion, field observations generally provide suggestive evidence that *T. latifolia* establishment and invasions occur under relatively high nutrient availability. Furthermore, Grace (1987) observed *T. latifolia* dominance under moderate (mesotrophic) surface water nutrient conditions in a reservoir suggesting that high nutrient loading may not be a requisite for *T. latifolia* invasion. The controlled and semi-controlled experiments conducted to determine *Typha* spp. response to N and P treatments (Shiple and Keddy, 1988; Wetzel and van der Valk, 1998; Keddy *et al.*, 2000; Woo and Zedler, 2002) used N and P treatments that exceeded levels observed in even the most polluted natural wetlands. There is therefore uncertainty regarding the contributions of elevated N and P concentrations characteristic of environmentally relevant nonpoint source effluent.

Despite a general awareness of the benefits natural and constructed wetlands provide summarized later, there is a lack of awareness or consideration regarding the potential impacts of intentionally planting *Typha latifolia* and other invasive species commonly used in constructed

wetlands for wastewater and stormwater treatment. Few authors have identified the risks of planting constructed wetlands with invasive species. Livingston (1989) stated in the oft-cited Hammer (1989) that the use of *T. latifolia* and other *Typha* species in constructed wetlands was “unacceptable”. Similarly, Taylor (1992) raised concerns regarding the potential impacts of utilizing *T. latifolia* and other invasive species in constructed wetlands. Despite these warnings *Typha* has continued to be the most commonly used macrophyte species in constructed wetlands in North America and second only to *Phragmites australis*, also an invasive in North America, in European constructed wetlands despite the availability of noninvasive species that in some cases have been shown to be as or more effective in removing N and/or P (e.g. Neil and Graham, 1989; White *et al.*, 2000).

There is a real disconnect between the objectives of constructed wetland practitioners, ecologists and land managers. The first criterion stated by Tanner (1996) for identifying species of vegetation for use in constructed wetlands was as follows:

“No significant weed or disease risks or danger to the ecological or genetic integrity of surrounding natural ecosystems”.

Scholz (2006) recommended in his book *Wetland Systems to Control Urban Runoff* , published in the UK, that vegetation used in constructed wetlands should have “ecological adaptability” and not pose a “weed risk”. Scholz (2006) then described the merits of utilizing *T. latifolia* and *P. australis* in constructed wetlands, two of the most invasive wetland species in North America. Kadlec and Wallace (2009) stated that the most suitable species of vegetation for constructed wetlands are aggressive rhizomatous species that have rapid grow rates, can rapidly infill newly built wetlands and can tolerate water level fluctuations and pollutants - traits often characteristic of invasive species.

In continuing with this disconnect, up to 80 % of total annual budgets available for wildlife and nature preserve management in the US has been allocated to the control of invasive species, particularly invasive plants (D’Antonio and Meyerson, 2002). In 1998, over 90 % of the \$32 to \$42 million US Federal annual budget available for the protection of species listed under the

*Federal Endangered Species Act* was allocated to invasive species control (Wilcove & Chen 1998).

Yet there has been recent interest in expanding the use of constructed wetlands to the watershed scale to remove N and P (Boesch *et al.*, 2001; Arheimer and Wittgren, 2002; Zedler, 2003; Woo, 2009; Gren, 2010; Kim, 2010) and pilot studies have been conducted to evaluate constructed wetlands using *T. latifolia* as a restoration technique for recovering eutrophicated aquatic ecosystems (Li *et al.*, 2008; Ham *et al.*, 2010; Özkundack *et al.*, 2010). Additionally, the constructed wetland N and P removal rates reported in Vymazal (2007) are too low to prevent ecological impact to receiving aquatic systems and can also fail to meet regulatory water quality standards (Vymazal, 2007; Lee *et al.*, 2009). The ecological risk of using invasive species in constructed wetlands combined with high N and P in both influent and effluent therefore needs to be adequately addressed in constructed wetland design.

Lastly, research conducted to identify plant species most suitable for use in constructed wetlands, including native noninvasive species (*e.g.* Tanner, 1996), has been inconclusive due, at least in part, to low levels of experimental unit replication, inconsistent methodologies and the use of plants of variable age and fitness (Brisson and Chazarenc, 2009). The propagation of plants from seed is therefore a requirement to ensure consistent age and fitness for effective hypothesis testing. However, little information is provided in published research on plant propagation protocols. The majority of published research on *Typha spp.* response to N and P have relied on shoots produced from rhizome (root stalk), often of unknown history (*e.g.* Weng *et al.*, 2006). The three studies located in the primary literature that propagated *T. latifolia* from seed (Ye *et al.*, 1997; Wetzel and van der Valk, 1998; Kercher and Zedler, 2004) provided incomplete details on germination and propagation protocols. In order to identify native noninvasive species suitable for use in constructed wetlands and to identify nitrogen (N) and phosphorus (P) loading rates which significantly increase *T. latifolia* biomass and potentially invasiveness, a standardized protocol for propagating *T. latifolia* from seed under low nutrient reference conditions is needed.



## 1.2 Objectives

1. Develop a protocol for rearing *Typha latifolia* from seed to adult stages under laboratory conditions to enable hypothesis testing based on data collected from plants of known history, uniform age and comparable fitness;
2. Conduct controlled, rigorous experiments with sufficient replication to further existing knowledge on the effects of environmentally relevant N and P loading rates on *T. latifolia* survival, growth and biomass;

## 1.3 Sub-objectives

1. Determine the feasibility of conducting future competition and nutrient removal experiments between *T. latifolia* and the native non-invasive species such as *Schoenoplectus acutus* and *Schoenoplectus tabernaemontani* under laboratory conditions;
2. Based on literature review, present supporting evidence for the hypothesis that native, non-invasive wetland macrophytes would provide N and P removal rates comparable to currently used invasive species and raise the issue of invasive species use in constructed wetlands to encourage research to determine the feasibility of utilizing non-invasive species in constructed wetlands for N and P removal.

## 1.4 Background

It should be emphasized here that it is not the intention of the author to impress upon the reader that *T. latifolia* is inherently invasive. The evidence overwhelmingly suggests that *T. latifolia* becomes invasive due to anthropogenic habitat or environmental alteration and/or degradation of wetland ecosystems and should therefore be viewed as an indicator of ecological health. *T. latifolia* may play an essential role in ecosystems that have remained relatively unimpacted and under such conditions should be viewed as a vital component of the ecosystem. The ecological significance of *T. latifolia* as well as that of *Typha angustifolia* and *Typha x glauca* should also be carefully assessed in impacted areas. *T. x glauca* is a dominant emergent species of vegetation in the severely degraded Cootes Paradise wetland (Chow-Fraser *et al.*, 1998) likely because they are one of the few species capable of surviving in that heavily polluted and hydrologically altered system (T. Theysmeyer, Royal Botanical Gardens, personal communication). The habitat they provide may be critical to a variety of waterbirds and wildlife species. Despite the severe habitat degradation, Cootes Paradise still supports species of conservation concern (listed and rare species) (Court and Bowman, 2011) and has been designated as the second important waterfowl staging area in Lake Ontario (Painter *et al.*, 1988) with *Typha spp.* likely contributing significantly to the persistence of a number of wildlife species in Cootes Paradise.

*T. latifolia* (common cattail, broadleaf cattail) is an aggressive rhizomatous wetland plant that is native to parts of Central and North America, Europe, Asia and Africa (Grace and Harrison, 1986) and has been introduced to many other parts of the world (Invasive Species Specialist Group, 2013). *T. latifolia* (along with other *Typha* species (*spp.*)) is one of the most productive plants known (Wetzel, 1983). Capable of rapid growth and vegetative reproduction from rhizome (underground stem), *T. latifolia* can achieve heights of 1 to 3 meters, very high biomass and produce 20,000 – 200,000 seeds from a single inflorescence (flower) (*e.g.* Yeo, 1964; Grace and Harrison, 1986).

These traits under certain conditions appear to enable *T. latifolia* to outcompete other plants and form dense monotypic stands with impoverished biodiversity (*e.g.* Curtis, 1959;

McNaughton, 1966; Linde *et al.*, 1976; Grace and Harrison, 1986; Moore *et al.*, 1989; Wetzel and van der Valk, 1998; Drohan *et al.*, 2006; Sullivan *et al.*, 2010; Asomoah and Bork, 2010; Invasive Species Specialist Group, 2013) (Figure 1). Pollen records indicate that the range of *T. latifolia* has been expanding in North America since the mid to late 20th century likely in response to habitat disturbance and habitat degradation associated with human development and land use (*e.g.* Shih and Finkelstein, 2008; Olson *et al.*, 2009). Over the last two to three decades, *T. latifolia* has often been defined as an invasive species in North America where it is native (*e.g.* Grace and Harrison, 1986; Livingston, 1989; Taylor, 1992; Kercher and Zedler, 2004; Bourgeois *et al.*, 2012). Kercher and Zedler (2004) defined invasive plants as follows:

“Species or strains that rapidly increase their spatial distribution by expanding into existing plant communities”.



**Figure 1. *Typha spp.* Monospecific stand, Highway 28, Kawartha Lakes Region, Southern Ontario, June 2012.**

*Photo by Mark Tiley.*

Under this definition, invasive species include native species such as *T. latifolia* that have displayed invasive characteristics in response to habitat disturbance or environmental change. Invasive species are of major environmental concern globally. Organizations and programs such as the World Conservation Union's Global Invasive Species Program, the U.S. National Invasive Species Council, the Ontario Phragmites Working Group and the recently terminated Alien Species Partnership Program have been formed in efforts to combat invasive species (Houlihan and Findlay, 2004).

The mechanisms that contribute to *Typha spp.* invasiveness, discussed later, have been studied for nearly three decades (*e.g.* Wilcox *et al.*, 1984; Wilcox *et al.*, 2008). However, disentangling the confounding mechanisms contributing to *Typha* invasion has proven illusive and a consistent challenge to researchers that have attempted to identify the role of invasive species in native species decline (Gurevitch and Padilla, 2004).

There is strong evidence indicating that P loading is a major contributor to *Typha domingensis* invasions in the Florida Everglades (*e.g.* Davis, 1991; Reddy *et al.*, 1993; Newman *et al.*, 1996; Noe *et al.*, 2001). Most wetlands of all types are generally N-limited and reduced wetland species diversity in Europe has commonly been associated with increases in N and both elevated N and P have been associated with low plant species diversity in North American wetlands (Bedford *et al.*, 1999). Similarly James *et al.* (2005) found an inverse relationship between winter nitrate ( $\text{NO}_3^-$ ) levels and submergent plant species diversity in United Kingdom lakes.

The effects of high P and N loading in degraded aquatic ecosystems are however often confounded by altered hydroperiod (*e.g.* Davis, 1991; Drohan *et al.* 2006; Boers *et al.*, 2007) and elevated levels of other nutrients and stressors such as potassium (K) and calcium (Ca), organic matter, suspended sediments, metals, industrial chemicals and other invasive species (*e.g.* *Cyprinus carpio* (common carp)), (*e.g.* Painter *et al.*, 1988; Martín and Fernández, 1992; Chow-Fraser *et al.*, 1998; Smolders *et al.*, 2006). Other contributing factors to *Typha* invasions may also include increased levels of carbon dioxide ( $\text{CO}_2$ ) (Sullivan *et al.*, 2010), global warming, reduced grazing pressure and fire suppression. How ecosystem changes associated with

eutrophication, such as altered nutrient cycling, pH, redox conditions, microbial communities and species composition and interaction, may contribute to *Typha* invasions have also not been addressed.

The relative importance of N and P to *T. latifolia* biomass production and invasion is also unclear. The need to address the uncertainty regarding the importance of environmentally relevant N and P concentrations on *T. latifolia* productivity for the protection and restoration of natural wetlands is obvious. There is a timely need to quantify *T. latifolia* response to environmentally relevant N and P levels as the building of remedial wastewater constructed wetlands (hence forth referred to as constructed wetlands) at the local scale has been suggested and/or assessed as a means of restoring eutrophicated inland freshwater ecosystems (e.g. Li *et al.*, 2008, Ham *et al.*, 2010; Özkundack *et al.*, 2010). Boesch *et al.*(2001); Arheimer and Wittgren, (2002); Zedler, 2003; Woo, (2009); Gren, (2010) and Kim, (2010) have recommended or investigated the use of constructed wetlands at the watershed scale for the protection and restoration of coastal marine ecosystems. *T. latifolia* and other invasive species (e.g. *Phragmites australis*, *Phalaris arundinacea*, *Typha angustifolia*) are commonly used in constructed wastewater and stormwater remedial wetlands, all of which have been shown to increase biomass in response to elevated nutrient levels (e.g., Grace and Wetzel, 1981c; Weisner, 1993; Mal and Narine, 2004; Drohan *et al.*, 2006; Martina and von Ende, 2008). *Phragmites australis australis*, the European genotype of *Phragmites australis* (*P. australis americanus* is the native genotype), is also highly invasive in North America (Galatowitsch *et al.*, 1999; Mal and Narine, 2004; Gilbert, 2012). The use *T. latifolia* and other invasive species in constructed wetlands may provide the high nutrient conditions often associated with *Typha* invasions.

To date, there is no evidence to indicate a macrophyte species effect on the pollutant removal performance of natural wetlands (reviewed by Nichols, 1983) and operational wastewater and stormwater remedial constructed wetlands (reviewed by Brix, 1994; Brix and Schierup, 1989; Brix, 1997; Brisson and Chazarenc, 2009; Marchand *et al.*, 2010). Brisson and Chazarenc (2009) did not observe a consistently superior species based on their literature review and were therefore unable to provide generalizations in regards to nutrient removal efficiencies of particular wetland plant species. Studies comparing the relative metals removal capabilities of

individual species in small-scale experiments have also been inconclusive (Marchand *et al.*, 2010). Non-invasive species of vegetation have been used successfully in both constructed wastewater treatment wetlands (*e.g.* Kadlec and Wallace, 2009) and natural wetlands used to treat wastewater (*e.g.* White *et al.*, 2000). Neil and Graham (1989) observed the non-invasive native bulrush *Schoenoplectus tabernaemontani* (formerly *S. validus*) to be generally more effective in the overall removal of all water quality parameters tested, including P and N, than the invasive *T. angustifolia* although the experimental mesocosm wetlands planted with *S. tabernaemontani* had a 5 day (d) - 7.5 d longer retention time. Taylor (1992) provided a list of species considered suitable for constructed wetlands in Ontario Canada which included both invasive and non-invasive species (Table 1).

The results from several studies involving the same species comparisons conducted to identify the relative pollutant removal capabilities for a given macrophyte species were sometimes conflicting (Brisson and Chazarenc, 2009). Brisson and Chazarenc (2009) noted that major differences in methodology between studies including the size and type of vessels, the age and condition of the plants used and a low level of replication were common limitations. Several authors have propagated *T. latifolia* from seed for research purposes (*e.g.* Ye *et al.*, 1997; Wetzel and van der Valk, 1998; Kercher and Zedler, 2004). However, a detailed protocol for rearing *T. latifolia* and noninvasive native wetland plants used in constructed wetlands (*e.g.* *Schoenoplectus spp.*) from seed was not found in the peer-reviewed literature. In order to ensure plants of the same age and fitness are available for hypothesis testing, a standardized protocol for rearing wetland plants from seed is needed given that nurseries may not have the means or capacity to provide disease-free and pest-free plants of consistent age and fitness.

In order to provide the reader with background information on how constructed wetlands remove N and P, N and P removal pathways and the relative importance of direct plant uptake are provided in later sections of the Introduction.

**Table 1. Non-invasive and invasive herbaceous wetland plant species identified as suitable for stormwater constructed wetlands.**

*Modified from Taylor (1992).*

Species	Common Name	Comments
<i>Cephalanthus occidentalis</i>	Buttonbush	
<i>Sagittaria latifolia</i>	Common arrowhead	
<i>Phalaris arundinacea</i>	Reed canary grass	Highly invasive in many wetlands, streams, waterways
<i>Schoenoplectus spp.</i> (formerly <i>Scirpus</i> )	Bulrush	
<i>Carex spp.</i>	Sedges	
<i>Acorus calamus</i>	Sweet flag	
<i>Potamogeton crispus</i>	Curly-leaved pond weed	Submergent aquatic introduced invasive from Eurasia
<i>Leersia oryzoides</i>	Rice cutgrass	
<i>Nuphar variegata</i>	Bulhead lily	
<i>Eleocharis acicularis</i>	Spikerush	
<i>Pontederia cordata</i>	Pickerel weed	
<i>Sparganium eurycarpum</i>	Giant bur-reed	
<i>Typha latifolia</i>	Common cattail, broad leaf cattail	1.5 – 3m tall invasive marsh plant
<i>Typha angustifolia</i>	Narrow leaf cattail	1.5 – 3m tall invasive marsh plant
<i>Phragmites australis americanus</i>	Common reed – North American genotype	Considered much less vigorous than the Eurasian subspecies
<i>Phragmites australis australis</i>	Common reed – Eurasian genotype	Highly invasive in North America, 2 – 6m tall
<i>Elodea canadensis</i>	Elodea	Submerged aquatic

## 1.5 *Typha latifolia* Biology and Ecology

Kadlec and Wallace (2009) defined natural wetlands as land areas that are sufficiently wet during parts of the year which enable wetland plants to occur. Under this definition, wetland habitat includes marshes, swamps, fens, bogs, sloughs, wet riparian zones and shallow depression or pocket wetlands. *T. latifolia* is an herbaceous monocotyledon obligate wetland plant belonging to the Typhaceae family which comprises the genus *Typha* (cattails) and *Sparganium* (bur-reeds) (Figure 2). Obligate species are defined here as species that occur only in a specific type of habitat or condition. The earliest fossils of Typhaceae date to approximately 90 million years ago during the late Cretaceous period (Bremner, 2002), coinciding with the emergence of the first flowering plants. *T. latifolia* is the only extant native *Typha* species in the northern US and Canada (Smith, 1967; Grace and Harrison, 1986).



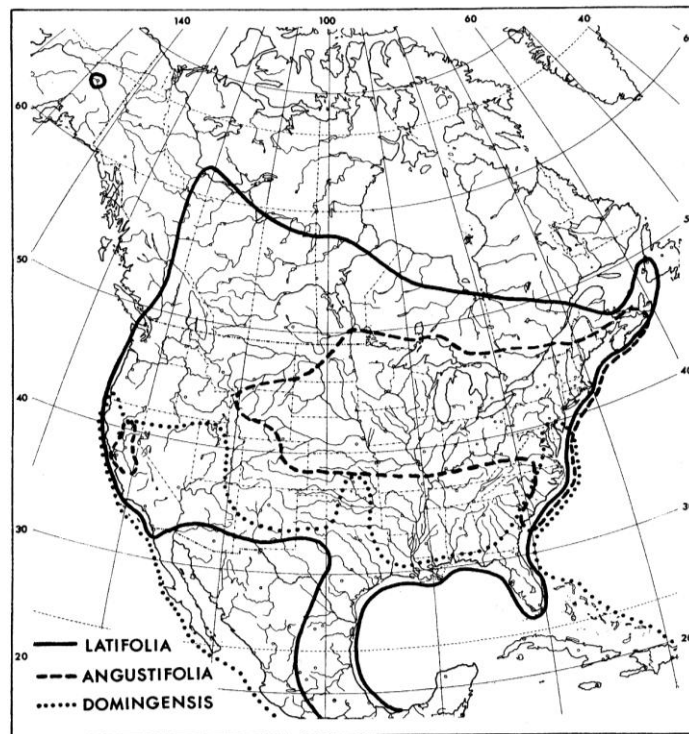
**Figure 2. Sexually mature *Typha latifolia* illustrating the staminate (male) spike and pistillate (female) spike of the inflorescence (flower).**

From [http://www.iowaplants.com/flora/family/Typhaceae/typha/t\\_latifolia/Typha\\_latifolia.html](http://www.iowaplants.com/flora/family/Typhaceae/typha/t_latifolia/Typha_latifolia.html)



### 1.5.1 Distribution and Habitat

*T. latifolia* is native to the Americas, Eurasia, and Africa and has been introduced to many other parts of the world including New Zealand, Hawaii and Australia where it is often invasive or a nuisance (Invasive Species Specialist Group, 2013). *T. latifolia* occurs in climates ranging from tropical to northern-temperate, dry to humid and at elevations from sea level to 2125 m (7000 ft) (Grace and Harrison, 1986). *T. latifolia* currently ranges throughout most of North America and its range historically overlapped with that of the native *Typha domingensis* in the southern U.S (Smith, 1967) (Figure 3). Since the mid-nineteenth century, the introduced *Typha angustifolia* has rapidly expanded its range westward from the northeast coast of the US into the historical range of *T. latifolia*. *T. latifolia* and *T. angustifolia* hybridize to form the natural hybrid *Typha x glauca* which has a similar range to that of *T. angustifolia* (Galatowitsch *et al.*, 1999).



**Figure 3. *Typha latifolia*, *Typha angustifolia* and *Typha domingensis* distributions in North America.**

*From Smith (1967), data from Hotchkiss and Dozier (1949).*

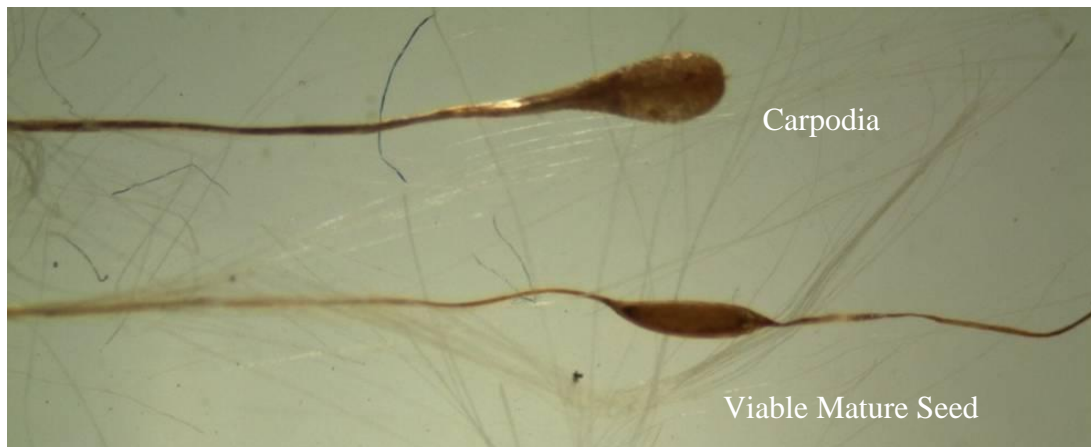
*T. latifolia* can establish in virtually any type of exposed substrate that remains sufficiently wet during most of the growing season (Grace and Harrison, 1986). Observations by Grace and Wetzel (1981b) and Grace (1985) that seeds germinate well in wet exposed sediment is consistent with the hypothesis that physical disturbance of vegetative cover (*e.g.* Shih and Finkelstein, 2008) and lower water levels that expose sediment are required for new *Typha spp.* stands to establish from seed. *T. latifolia* can tolerate moderate salinities (Hotchkiss and Dozier, 1949; Smith, 1967) but may be outcompeted by the more salt-tolerant *T. angustifolia* and *T. x glauca* in roadside ditches where salinities can be elevated where de-icing road salt is applied (Grace and Harrison, 1986; Olson *et al.*, 2009). It is reported in Smith (1967) to be the only acid-tolerant species of *Typha*. Jensen and Brix (1996) found *T. latifolia* seedlings grew rapidly at pH 5.0 and were able to tolerate a pH of 3.5 for several days before showing signs of stress. *T. latifolia* are shade-tolerant, capable of growing in wooded areas affected by tree canopy cover (Grace and Wetzel, 1981a). Arbuscular mycorrhizae colonize *T. latifolia* roots providing mineral nutrients in exchange for photosynthate (Ray and Inouye, 2006) which may enable *T. latifolia* to survive in acidic low-nutrient soils. Arbuscular mycorrhizae are mycorrhizae (fungus) which penetrate the cortex of plant roots.

The presence of *T. latifolia* in nutrient enriched systems is well document as previously indicated. *T. latifolia* also occurs in pristine wetlands within the circumpolar boreal regions (Bourgeois *et al.* 2012) and has been documented in Sifton Bog, Ontario, since the 1920's (City of London and the Upper Thames River Conservation Authority, 2009) indicating that *T. latifolia* can establish under conditions of low nutrient availability where they generally occur in isolated stands. Steward *et al.* (1997) noted that *T. latifolia* were restricted to areas of the Florida Everglades that had been impacted by anthropogenic nutrient loading possibly due to their exclusion by the competitive dominance of *Cladium jamaicense* in unimpacted oligotrophic areas. Grace (1987) observed *T. latifolia* dominance under moderate (mesotrophic) surface water nutrient conditions suggesting that factors other than high nutrient loading contribute to *T. latifolia* competitive dominance.

### 1.5.2 Germination

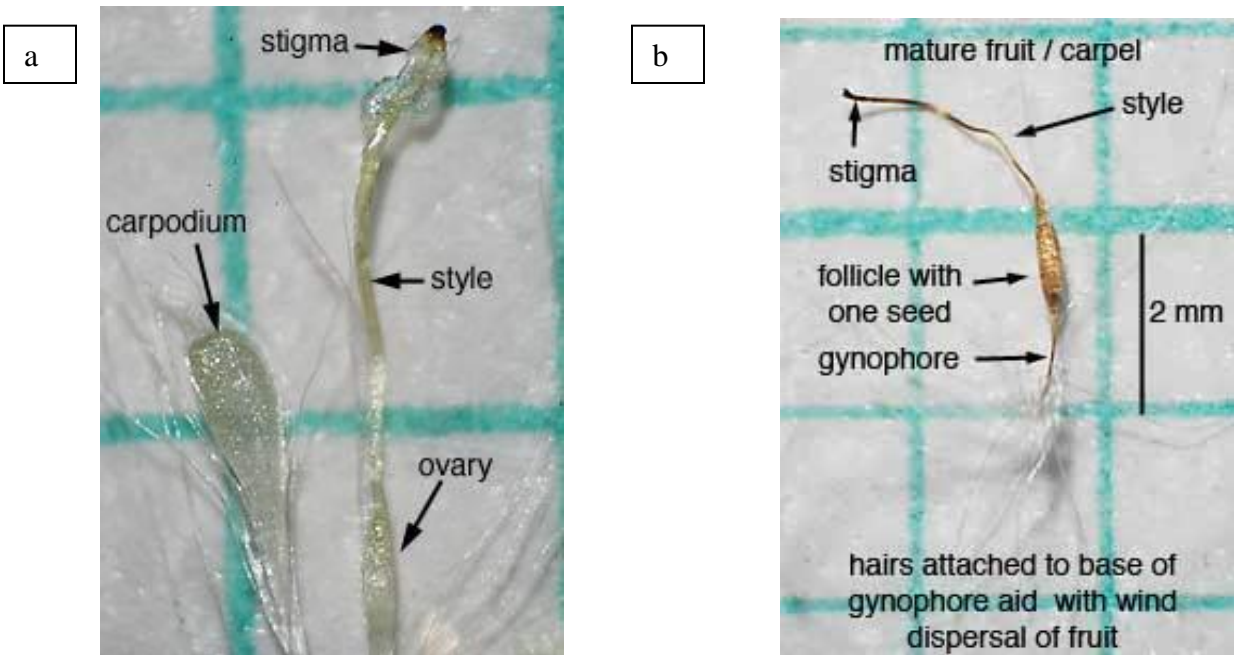
*T. latifolia* seeds are approximately 1.5 millimeters (mm) in length (Grace and Harrison, 1986) and weigh  $9 \pm 1$  micrograms ( $\mu\text{g}$ ) with all plumages and appendages removed (Shipley and Keddy, 1988). Fine hairs attached to the seed appendages facilitate wind dispersal (Yeo, 1964) (Figure 5b). *Typha spp.* are unique in that non-viable seeds termed carpodia (*pl.* = carpodium) or pistillodia by Krattinger (1975) (Figure 4, Figure 5) are produced (Hotchkiss and Dozier, 1949; Yeo, 1964), the evolutionary significance of which is unknown (Dr. S. G. Smith, emeritus, University of Wisconsin, personal communication).

Germination under natural conditions appears to occur under suitable light and moisture conditions that can occur at any time from spring through fall (Grace and Harrison, 1986). Moist soil conditions were stated by Grace and Wetzel (1981b) as a requirement for germination although submerged seeds also germinate (Sifton, 1959; Yeo, 1964; Bonnewell *et al.*, 1982; Grace, 1985). Grace (1985) observed a decreasing germination rate along a moisture gradient with distance from the water table. Sifton (1959) observed a higher germination rate in seeds submerged compared to seeds germinated on a moistened artificial substrate under laboratory conditions.



**Figure 4.** *Typha latifolia* carpodia and mature viable seed from the same inflorescence.

*Photo by Mark Tiley.*



**Figure 5. *Typha latifolia* seed structures for developing immature seed (a) and mature seed (b).**

From [http://www.iowaplants.com/flora/family/Typhaceae/typha/t\\_latifolia/Typha\\_latifolia.html](http://www.iowaplants.com/flora/family/Typhaceae/typha/t_latifolia/Typha_latifolia.html)

Yeo (1964) observed that seeds germinated at a depth of 76.2 centimeters (cm) (maximum depth of the vessel) were able to grow leaves that reached the water surface. Both Yeo (1964) and Grace (1985) noted that submerged seeds that germinated produced floating leaves before generating erect leaves that grew above the water surface. Yeo (1964) noted that seeds germinated in moist soil grew “leathery” leaves that differed morphologically from the leaves produced by submerged seeds.

### 1.5.2.1 Germination Under Laboratory and semi-controlled Conditions

Previous studies that investigated *T. latifolia* percent germination generally involved the quantification of only seeds that sank when placed in water with the floating material, assumed to be remnants of flower parts and nonviable seeds, being removed (*e.g.* Yeo, 1964; McNaughton, 1968; Bonnewell *et al.*, 1982). Percent germination near or at 100 percent (%) has often been observed under laboratory conditions (*e.g.* Morinaga, 1926; Sifton, 1959; McNaughton, 1968; Bonnewell *et al.*, 1982; Grace 1983; Bourgeois *et al.*, 2012). However, Yeo (1964) observed a very low percent germination (0 – 15.7 %) and Stewart *et al.* (1997) observed

a relatively low percent germination of 30 to 40 % for *T. latifolia* seeds collected from the Florida Everglades and germinated in environmental chambers (light intensities of 10, 764 or 32, 292 lux) and a shade house using paper towel and peat sediment as substrate. Yeo (1964) was later able to achieve 100 % germination using the same seeds by applying pressure to rupture the seed coat to expose a circular opening. Bourgeois *et al.*, (2012) observed a maximum percent germination of 36 % on seeds grown on peat moss under controlled conditions.

### **1.5.2.2 Temperature effect on Seed Germination under Laboratory Conditions**

McNaughton (1966) did not observe germination below 18°C in seeds collected from populations ranging from Texas to North Dakota. Sifton (1959) and Bonnewell *et al.*, (1982) observed a low germination rate at 15 degrees Celsius (°C) for seeds collected in Minnesota and a reduced time to maximum germination (48hrs) at 32°C. Bonnewell *et al.*, (1982) observed that the maximum percent germination of submerged seeds occurred at dissolved oxygen (DO) levels of between 2.3 milligrams per litre (mg/L) and 4.3 mg/L below or above which percent germination was significantly lower. Sifton (1959) stated that the required light intensity for germination is low and observed 0 – 10 % (average of 5 %) germination in total darkness (layers of paper wrapped around the seed-containing vessel. Seeds held in total darkness exposed to alternating temperature treatments (15 – 30°C, 15 - 35°C, 20 - 35°C) resulted a germination rate of up to 97.2 %. The observations obtained by Sifton (1959) indicated that *T. latifolia* seeds are capable of germination under low light conditions, including total darkness, which may provided a significant competitive advantage with the ability to germinate under shaded conditions.

### **1.5.2.3 Seed Tolerance to pH and Sedimentation**

Little information was found on the effects of pH on *T. latifolia* germination. Bourgeois *et al.* (2012) found evidence to suggest that the low pH found in moss cover grown on commercial peat (3.7) significantly reduced *T. latifolia* germination compared to seeds exposed to a pH of 4.3 grown on natural peat with a moss cover.

Percent germination for *T. latifolia* seeds subjected to burial by sediment was not located in the literature. (Wang *et al.*, 1994) observed > 70 % germination at the sediment surface and < 20

% when buried by 4mm of sediment for *T. x glauca* seeds. Similarly, Galiano and van der Valk (1986) observed 72 % germination for *T. x glauca* seeds at the sediment surface, 10 % germination with seeds buried by 1mm of sediment and 0 % germination with seeds buried by 2 mm of sediment.

### **1.5.3 Allelopathy**

There is conflicting evidence as to whether *T. latifolia* release allelotoxins hypothesized by McNaughton (1968) to be a competitive strategy for preventing the establishment of new *T. latifolia* genotypes. McNaughton (1968) observed a slight negative effect on the percent germination of seeds subjected to marsh water collected from a mature *T. latifolia* stand and a significant reduction in germination when seeds were subjected to water squeezed from sediments in which adult *T. latifolia* were growing. Seeds that germinated in aqueous leaf extract did not grow beyond a tiny radicle. Grace (1983) observed no effect on *T. latifolia* seed germination following exposure to adult *T. latifolia* soil substrate and *T. latifolia* leaf litter extract but did observe a lower germination rate in seeds exposed to moulds which became established in the cultures.

### **1.5.4 Growth and Vegetative Reproduction**

Newly established seedlings under natural conditions are capable of spreading rapidly and prolifically by rhizome (Grace and Harrison, 1986). A rhizome is an underground stem or shoot that later turns upward, penetrating through the sediments to access sunlight at which point is referred to as an aerial shoot or a ramet. A single plant germinated from seed is capable of producing a network of rhizomes reaching 58 m<sup>2</sup> over a two-year period (Grace and Wetzel, 1981b). Yeo (1964) observed astounding growth over a single 7 month growing period (April 01 – November 01) where rhizome expansion from a single seed uninhibited by competition or physical obstruction reached a diameter of 3.05m and vegetatively produced 34 aerial shoots (ramets or “suckers”) ranging in height from 46 cm to 122 cm.

Yeo (1964) observed that when the parent shoot produced from seed reaches approximately 30.0 cm, the base of the shoot forms a crown at which point growth slows. The

first rhizomes are produced shortly thereafter when parent shoots reach heights of approximately 36 – 46 cm. After 30 – 61cm of lateral growth, the rhizomes turned upward to form an aerial shoot. The genet consists of the parent ramet produced from seed and all vegetatively produced ramets (Grace and Wetzel, 1981b). *T. latifolia* aerial shoots can grow to depths of 1m but may be restricted to shallower waters through competition with *T. angustifolia* (Grace and Wetzel, 1981b; Weisner, 1993).

The leaf high amount of leaf litter and dead root mass produced by *T. latifolia* may significantly increase the organic matter of sediments and provide a further source of nutrients. Angeloni *et al.* (2006) observed greater than 10-fold higher N and P concentrations within the sediments of an expanding *T. x glauca* stand compared to the adjacent sediments where noninvasive native plant species occurred. The production of high amounts of leaf litter can completely blanket the sediments (personal observation) which may prevent the germination or vegetative regeneration of other plant species. The manipulation experiment conducted by Jordan *et al.* (1990) observed reduced *T. angustifolia* production after three years receiving three times the amount *T. angustifolia* litter compared to the unaltered reference condition. Van der Valk (1986) observed greater species richness and shoot density were greater in plots cleared of *T. x glauca* and *P. australis* litter compared to treatments that received a blanket of *T. x glauca* litter.

### **1.5.5 The *Typha latifolia* Life Cycle in Temperate Regions**

Northern US populations translocate photosynthate to the rhizomes beginning in mid to late summer for over-winter storage with rhizome maximum biomass being achieved by late fall (Grace and Harrison, 1986; Garver *et al.*, 1988). Garver *et al.* (1988) observed that in a Minnesota *T. latifolia* stand, 35 to 44 % of P in shoot tissue had been translocated to the rhizomes by late fall, 75 % of which was retained to the following spring.

Growth from rhizome in the spring in a Minnesota population was reported by Dubbe *et al.* (1988) to begin in early May. Initial growth from rhizome with growth continues until mid to late summer (Grace and Harrison, 1986; Garver *et al.*, 1988) at which point flowering may occur at the expense of vegetative production (Grace and Wetzel, 1981a). Rhizomes connecting ramet

clones die after 2 – 3 years (Kuhlen *et al.*, 1999) and typically die after flowering (Grace and Wetzel, 1981a).

### **1.5.6 Phenotypic Variability**

*T. latifolia* displays considerable phenotypic variability depending on habitat and environmental conditions (Grace and Harrison, 1986). Grace and Wetzel (1981a) did not observe differences in overall biomass between three *T. latifolia* populations occurring adjacent to the same lake based on meristem N and P tissue content, an indicator of N and P availability within the surrounding environment, but did observe significant differences in phenotypic expression associated with N and P allocation. The woods biotype growing under tree and shrub canopy had the highest meristem nutrient levels and was thus considered to be non-nutrient limited but had the largest leaf volume: dry mass ratio and was therefore considered to be light-limited. Despite having access to unlimited nutrient resources, the woods biotype did not allocate any photosynthate to sexual reproduction (no inflorescences were produced) (Grace and Wetzel, 1981a).

The *T. latifolia* population occurring within a marsh (cattail marsh biotype) and sheltered from wind and waves also did not produce inflorescences and was considered N-limited based on tissue N content (Grace and Wetzel, 1981a). The *T. latifolia* population growing in exposed open water habitat was considered to be both N and P co-limited yet was the only population to produce inflorescences (11 % of all ramets). Individuals within the open water population that did not produce inflorescences had twice the vegetative biomass compared to individuals that produced inflorescences. N-P-K nutrient additions applied to the same populations in a separate experiment under field conditions had no effect on the woods biotype but resulted in a fivefold increase in biomass in the open-water biotype (Grace and Wetzel, 1981a).

### **1.5.7 Nitrogen and Phosphorus Limitation in *Typha latifolia***

The relative observed importance of N and P to *T. latifolia* growth, biomass and reproductive potential (fecundity) differs between studies. Field experiments by Grace (1988) observed an increase in *T. latifolia* biomass in response to increased soil N but no response was



observed in response to increased sediment P which suggested N limitation. Conversely, Boyd and Hess (1970) observed a significant correlation between surface water PO<sub>4</sub>-P (200 – 320 µg PO<sub>4</sub>-P) and *T. latifolia* standing crop whereas no correlation was observed between standing crop and surface water NO<sub>3</sub>-N (10 – 450 µg NO<sub>3</sub>-N). Grace and Wetzel (1981a) observed possible differences in N and P limitation between three cattail stands occurring within the same wetland system depending on site-specific substrate N and P availability and other site-specific factors such as wind exposure and light availability.

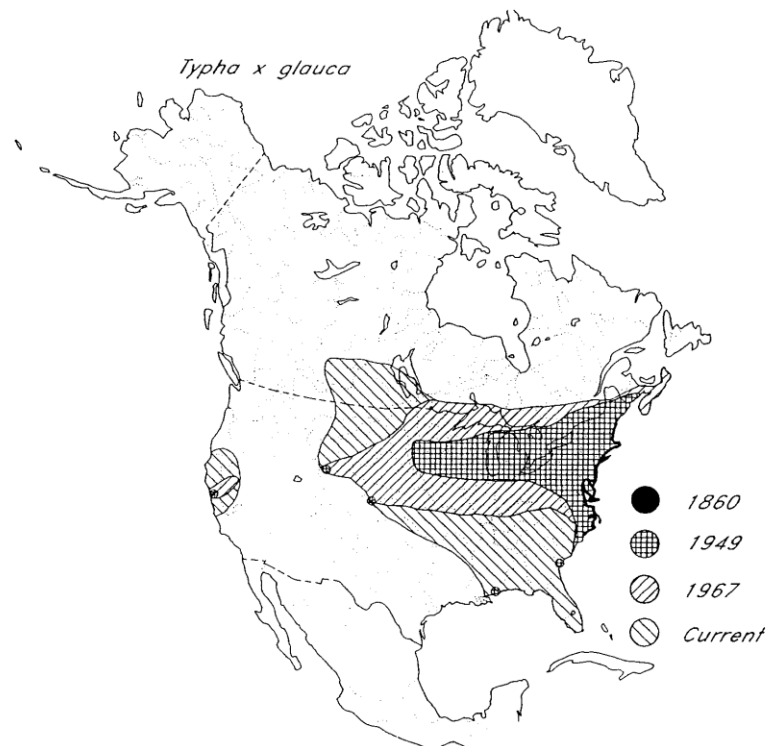
### **1.5.8 Sexual Reproduction and Genetic Diversity**

Stigmas are receptive one to two days before pollen is released from the staminate spike. Pollen is released over a 1 – 2 day period in which self-pollination can occur (Grace and Harrison, 1986; Zhang *et al.*, 2008). It has been hypothesized that stigmas remain receptive for up to 1 month to enable out-crossing (Grace and Harrison, 1986). Conversely, Krattinger (1975) observed that self-pollination was highly favoured with the majority of pollen falling directly from the male staminate spike onto the female flower below. All studies that have investigated the genetic traits of *T. latifolia* populations have found little to no genetic variation within populations and between populations within river drainage systems (*e.g.* Keane *et al.*, 1999; Lamote, 2005; Zhang *et al.*, 2008). The low genetic diversity is likely the result of self-pollination and the vegetative production of clones (Zhang *et al.*, 2008).

### **1.5.9 *Typha latifolia* Hybridization with *Typha angustifolia***

*Typha angustifolia*, an invasive either introduced from Europe in the nineteenth century or historically restricted to the Northeast coast of the United States, has also been used in North American constructed wetlands (Neil and Graham, 1989; Selbo and Snow, 2004). The range of *Typha angustifolia* has expanded rapidly across North America since approximately 1860 (Galatowitsch *et al.*, 1999; Shih and Finkelstein, 2008) (Figure 6). *T. angustifolia* often cross-pollinates with *T. latifolia* where the two species are sympatric producing the natural hybrid *Typha x glauca* (Smith, 1967; Waters and Shay, 1990; Galatowitsch *et al.*, 1999; Shih and Finkelstein, 2008; Olson *et al.*, 2009).

*T. x glauca* is considered to be more invasive than either of the parent species (Galatowitsch *et al.* 1999; Frieswyk and Zedler, 2007; Olson *et al.*, 2009) and has, over several decades, become the dominant species at many sites in the Lower Great Lakes region formerly occupied by *T. latifolia* (Kirk *et al.*, 2011) and is believed to have displaced *T. latifolia* and other species in many of the Great Lakes coastal wetlands (*e.g.* Frieswyk and Zedler, 2007; Wilcox *et al.*, 2008). *T. x glauca* was reported as becoming increasingly abundant in both North American and European prairie wetlands (Waters and Shay, 1992) and is now the most common large macrophyte in the North American prairie pothole region (Linz and Homan, 2011).



**Figure 6. The expansion of *Typha angustifolia* and *Typha x glauca* in North America.**

*From Galatowitsch et al., 1999.*

The expansion of *T. angustifolia* and *T. x glauca* across North America has often been attributed to the high salinities that are less favourable or intolerable to *T. latifolia*, altered hydroperiod, high nutrient loading and habitat disturbance (*e.g.* Grace and Harrison, 1986; Galatowitsch *et al.*, 1999; Woo and Zedler, 2002; Frieswyk and Zedler, 2007; Boers *et al.*, 2007; Boers and Zedler, 2008). *T. x glauca* is generally considered to be largely sterile (*e.g.* Smith,

1967; Galatowitsch *et al.*, 1999). However, Bedish (1967) observed a *T. x. glauca* percent germination of 48 %, Wang *et al.* (1994) up to 81 % and Galiano and van der Valk (1986) up to 75 % germination with all three studies being conducted under greenhouse conditions. Bedish (1967) observed no *T. x glauca* germination under field conditions using seeds from the same inflorescence as used in his greenhouse experiments and on the same natural wetland soil. The spread of *T. x glauca* through seed dispersal is therefore at least a possibility.

### **1.5.10 Herbivory and Wildlife Habitat**

Muskrat (*Ondatra zibethicus*) can be defined as a keystone wetland species as their grazing can profoundly affect wetland plant species distribution and diversity, habitat structure and heterogeneity, avian abundance and diversity, nutrient cycling and macroinvertebrate diversity and distribution (*e.g.* Errington *et al.*, 1963; de Szalay and Cassidy, 2001; Roberts and Crimmins, 2010) which may interact significantly with a variety of fish and wildlife species. Roberts and Crimmins (2010) documented a 75 % reduction in muskrat harvest rates in eastern North America since 1986. Roberts and Crimmins (2010) further recommended that a regional assessment of muskrat populations be conducted to determine whether a true decline in muskrat populations is occurring so that management efforts could be implemented to reverse declines which may be necessary to protect remaining functional wetland ecosystems.

*T. latifolia* is known to be an important if not primary diet of muskrat (Beule, 1979; Campbell and MacArthur, 1997; Mirka *et al.*, 1996). Muskrat herbivory was considered by van der Valk and Davis (1978) to have a major effect on *Typha* abundance and wetland plant community structure in the Prairie region. Curtis (1959) noted that *Typha spp.* densities were absent or rare in marshes where muskrat populations were high. Bedish (1967) experienced the decimation of some *T. x glauca* experimental plots, including the digging and consumption of buried rhizomes, to muskrat herbivory. A possible reduction in muskrat abundance which may have been caused by the loss of wetland habitat and land use development (*e.g.*, Whillans, 1982; Curtis, 1989; Snell, 1989; Dahl & Johnson, 1991) and may therefore be a contributing factor in *Typha* invasions.

Grazing pressure on *Typha* by other vertebrate herbivores is generally considered to be light although elk and moose (Stevens and Hoag, 2000) and deer (Beule, 1979) graze newly emerging shoots during the spring and geese feed on roots (Stevens and Hoag, 2000). Seeds are generally not considered palatable due to the plumage but do provide food for several duck species. A variety of insects feed on the various *T. latifolia* structures (Grace and Harrison, 1986). The near total loss of *T. latifolia* stands to the cattail armyworm (*Simyra henrici*) in constructed wetlands was reported in Taylor (1992) and Ye *et al.* (2001).

## **1.6 The History of *Typha* Invasions in North America**

### **1.6.1 Early Documentation of *Typha* spp. Invasions and the Development of Control Methods**

The shift from a dominance of *Zizania aquatica* (wild rice) to *T. latifolia* in the Cootes Paradise wetland approximately coincides with initial European settlement and probably in response to deforestation, increased sediment deposition and nutrient loading (Chow-Fraser, 2005). It is therefore likely that *T. latifolia* invasions have occurred throughout the species' range where disturbance and nutrient loading associated with human settlement has occurred.

The first document found describing a *Typha* invasion in North America was reported by Addy and MacNamara (1948) which documented the loss of shoreline bird nesting habitat, open water waterfowl habitat and wildlife food plants to *Typha* encroachment (Figure 7). Addy and MacNamara (1948) provided four methods for removing *Typha*: (1) raising water levels (drowning); (2) explosives (dynamite) to open channels and create new ponds; (3) removal by hand and (4) cutting. Other early documents related to *Typha* control cited in Beule (1979) included Heath and Lewis (1957); Martin *et al.*, (1957) and Steens *et al.* (1959).

The study by Beule and Janisch (1973) (unpublished) was possibly the first scientific study conducted that assessed a *Typha* invasion in a conservation context where the objective of experiments were to identify methods for recovering lost wetland habitat and wildlife diversity. Shortly after, Beules' manual titled *Control and management of cattails in southeastern Wisconsin wetlands*, published in 1979, provided various techniques for controlling *Typha*, some

of which were based on the information presented in Linde *et al* (1976). Beule (1979) continues to be commonly cited in the peer-reviewed literature and has undoubtedly contributed much to wetland restoration efforts.



Figure 20. Cattail control plot, Patuxent Research Refuge, Maryland. Plants cut July 19, 1946 just after flowering and followed by a slight cutting of volunteers one month later. (Photograph by F. M. Uhler.)



Figure 21. Stand of wildrice on August 6, 1947 growing in place of cattails in Figure 20. Wildrice seed sown Sept. 11, 1946 after the cattail was mowed. (Photograph by F. M. Uhler.)

**Figure 7. Wildlife habitat enhancement performed in 1946 through *Typha latifolia* removal (above) and re-seeding with wild rice (below).**

*From Addy and MacNamara, 1948.*

Wilcox *et al.* (1985) was the earliest peer-reviewed publication located in the primary literature that investigated the potential causes of a *Typha* invasion. The study used a combination of air photo interpretation and field sampling to quantify the expansion of *Typha* into grass-sedge meadow over a 44 year period within Cowles Bog National Natural Landmark, Indiana. The sedge-grass habitat had received Federal protection as a significant natural feature.

Results indicated that *T. latifolia* and *T. angustifolia* occurring in isolated stands increased in area slightly from 2 hectares (ha) in 1938 to 7 ha by 1970. By 1982 the *Typha* stands had merged to form large stands totaling 37.2 ha at the expense of the native sedge-grass habitat. Wilcox *et al.* (1985) attributed the cause of the *Typha* invasion to increased and stabilized water levels associated with seepage from diked fly-ash settling ponds constructed upslope of Cowles Bog in the mid-1960s and in 1971. It was hypothesized that the altered hydrology created habitat conditions that increased *Typha* sexual reproductive success and rhizomatous expansion and as new *Typha* stands became established, sedge-grass meadow species were shaded out.

Wilcox *et al.* (1985) predicted that the remaining sedge-grass meadow would eventually be lost as the control techniques such as herbicide application and flooding provided in Beule (1979) were deemed infeasible or likely to be ineffective. The study by Wilcox *et al.* (1985) is probably the first scientific study to document the loss of a unique ecosystem and possibly species extirpation as a result of a *Typha* invasion.

From 1947 – 1971 *T. x glauca* and *T. angustifolia* increased from 30 % sub-dominance to covering 80 % of the Horicon Marsh Wetland Area (HMWA) wetland in Wisconsin (Beule, 1979). This major loss in wildlife habitat resulted in the first long-term study of *Typha* invasion by Linde *et al.*, (1976). Started in 1969, the purpose of the study was to (1) identify the most effective *Typha* control measures and (2) determine strategies for maximizing wildlife habitat value and biodiversity. The study produced, along with extensive data on *Typha* phenotypic and life history characteristics, recommendations for future control such as cutting in late June when carbohydrate stores are at their lowest and immediately before new shoots are capable of significant photosynthesis. Though the study did not discuss causation of the *Typha* invasion, it

was the first major body of work to raise awareness regarding the impacts of *Typha* in an ecological context.

## 1.7 Potential Causes of *Typha latifolia* Invasions

Nutrient loading and altered hydroperiod (increased and/or stabilized water levels, unnatural water level fluctuations) or their combined effects are considered to be the primary stressors contributing to *Typha* invasions (e.g. Wilcox *et al.*, 1985; Woo and Zedler, 2002; Bradley and Wolf, 2005; Boers and Zedler, 2008; Samoah and Bork, 2010). Separating the effects of elevated and stabilized water levels and nutrient availability may be challenging in the case of wetlands subjected to elevated or stabilized water levels. Natural decreases in water levels during dry seasons and drought periods result in the oxidization of dewatered sediments which can increase P retention due to the increased availability of oxidized iron (Fe) which precipitates P (discussed further below). Artificially prolonged inundation or water level stabilization can result in the development of anaerobic conditions in the sediments. The release of bioavailable N, P and K may reach levels high enough to degrade surface water quality and alter wetland vegetation community and function, including wetlands not subjected to elevated external nutrient loading rates - a process termed internal eutrophication (Koerselman *et al.*, 1993; reviewed by Smolders *et al.*, 2006). Kercher and Zedler (2004) were the authors found in the literature on causation of *T. latifolia* invasion to consider the potential interaction between altered hydroperiod and nutrient availability to explain *Typha* invasions in North America.

*Typha spp.* are generally considered to be resistant to a variety of stressor such as pollution, sedimentation and moderate water level fluctuations which may exclude other less tolerant species or provide *Typha spp.* with a competitive advantage. Other confounding stressors often mentioned in study site descriptions in studies that have examined *Typha* invasions such as the presence of other invasive species, sedimentation, industrial chemical contaminants and physical habitat disturbance (e.g. Wei and Chow-Fraser, 2006) have generally not been quantified or included in analysis. Discussion in the following section is therefore primarily limited to the effects of nutrient loading and altered hydroperiod largely due to a lack of data relating other potential stressors to *Typha* invasions.

## 1.7.1 Nutrient Loading

### 1.7.1.1 Evidence from Field Research under Natural Conditions

Field studies that have examined the effects of increased nutrient availability on North American *Typha spp.* have generally observed an increase in biomass with increased nutrient availability. Dubbe *et al.* (1988) observed a parallel relationship between soil N, P and potassium (K) uptake and biomass production in *T. latifolia*, *T. angustifolia* and *T. x glauca*. Moore *et al.*, 1989 observed an inverse relationship between wetland plant species biodiversity and soil nutrient levels with rare species occurring only in wetlands with a low standing crop (biomass/m<sup>2</sup>) and comparatively low soil nutrient levels. Vegetation standing crop was also inversely related to species diversity with *Typha* stands having the highest standing crop and the lowest plant species biodiversity. Woo and Zedler (2002) documented increases in *T. x glauca* biomass planted in sedge-grass meadow in response to commercial lawn fertilizer additions whereas native sedges (*Carex spp.*) showed no response to the fertilizer treatments.

The substantial research conducted in the Florida Everglades provides strong support for the eutrophication hypothesis. The shift from *Cladium jamaicense* (sawgrass), a keystone species adapted to low nutrient conditions (Steward and Ornes, 1975), to *Typha domingensis* has been commonly associated with P inputs derived from agricultural and stormwater runoff although increased, stabilized water levels are also believed to be a contributing factor (*e.g.* Davis, 1991; Reddy *et al.*, 1993; Newman *et al.*, 1996; Noe *et al.*, 2001). *Cladium jamaicense* can store excess nutrients for later use during periods of nutrient scarcity (Newman *et al.*, 1996) and is thus well adapted to the oligotrophic conditions present in unimpacted areas of the Florida everglades (Craft and Richardson, 1993; 1998; Noe *et al.*, 2001). *T. domingensis* can rapidly assimilate nutrients into tissue biomass, a characteristic of all *Typha* species (*e.g.* Dubbe *et al.*, 1988). *T. domingensis* invasions observed in wetland cells impacted by nutrient loading suggests that *T. domingensis* out-competes other species adapted to nutrient-poor conditions or that are less competitive under high nutrient availability. Similarly, Drohan *et al.* (2006) attributed a *T. latifolia* invasion to a combination of nutrient loading as a result of cattle grazing and altered hydroperiod resulting from the damming effect of a road which increased and stabilized water levels.



The lowest P treatment found in the literature used to determine *Typha* response to P loading was by Newman *et al.* (1996) who applied “low” 50µg/L P (eutrophic levels according to Environment Canada (2004) and the EPA) as a reference to compare to 100µg/L P (hypereutrophic levels) with both treatments being achieved by diluting marsh water to determine the competitive response of *T. domingensis*. No data was found in the primary literature that documented the response of *Typha spp.* to oligotrophic nutrient levels in surface water or oligotrophic loading rates. Data on *T. latifolia* survival, growth and competitiveness under oligotrophic conditions could serve as a reference for historically natural oligotrophic systems that have become mesotrophic or eutrophic.

#### **1.7.1.2 *Typha spp.* response to high nutrients under Controlled and Semi-Controlled Conditions**

The TP and TN concentrations reported in Wetzel (1983) obtained from freshwater lakes and reservoirs and used to define the trophic status in freshwater ecosystems in general are presented in Table 2. Environment Canada definitions for freshwater trophic status (Environment Canada, 2004) are provided in Table 3. The United States Environmental Protection Agency (EPA) classification method established freshwater trophic status based on several parameters including total phosphorus (TP), chlorophyll a and water transparency but does not incorporate N concentrations. Oligotrophic, mesotrophic, eutrophic and hypereutrophic TP levels according to EPA classification are 0 -12, 12 – 24, 24 – 96 and 96 – 384 µg TP/L respectively. Controlled and semi-controlled experiments conducted under laboratory conditions with artificial light (Wetzel and van der Valk, 1998), under greenhouse conditions (Shipley and Keddy, 1988; Woo and Zedler, 2002) and in an outdoor setting (Keddy *et al.*, 2000) to determine *Typha spp.* response to elevated levels of N and P, measured as function of shoot height and/or biomass, and competitive hierarchy (relative competitiveness compared to other species) have involved loading rates that in general far exceed levels observed in even the most polluted natural wetlands (Table 4). All studies observed increased biomass or size with increased nutrient availability; however, as results have limited environmental relevance, there is still uncertainty as to how *T. latifolia* responds to N and P loading rates typically observed in eutrophicated natural wetlands.

**Table 2. Trophic status of freshwater ecosystems based on TN and TP concentrations.**

*From Wetzel (1983). a = EPA classification for hypereutrophic status.*

<b>Parametr</b>	<b>Oligotrophic</b>	<b>Mesotrophic</b>	<b>Eutrophic</b>	<b>Hypereutrophic <sup>a</sup></b>
Total P (µg/L)	Mean 8.0	Mean 26.7	Mean 84.4	Mean ≥ 96 <sup>a</sup>
	Range 3.0-17.7	Range 10.9-95.6	Range 16 - 386	Range 750 – 1200
Total N (µg/L)	Mean 661	Mean 753	Mean 1875	Mean -
	Range 307-1630	Range 361-1387	Range 393-6100	Range -

**Table 3. Total phosphorus concentrations in relation to freshwater trophic status as defined by Environment Canada (2004).**

<b>Trophic Status</b>	<b>Total Phosphorus (µg/L)</b>
Ultra oligotrophic	< 4.0
Oligotrophic	4 – 10
Mesotrophic	10 – 20
Meso-eutrophic	20 – 35
Eutrophic	35 – 100
Hypereutrophic	> 100

**Table 4. N and P treatment concentrations used in controlled and semi-controlled experiments conducted to determine response in *Typha latifoli* and *Typha x glauca* biomass and competitive hierarchy.**

Authors	<i>Typha</i> Species	N	P
Shiple and Keddy (1988)	<i>T. x glauca</i>	630µg NO <sub>3</sub> -N/L	<u>Low Treatment:</u> 155µg PO <sub>4</sub> -P/L <u>High Treatment:</u> 1550µg PO <sub>4</sub> -P/L
Keddy <i>et al.</i> , (2000)	<i>T. x glauca</i>	630µg NO <sub>3</sub> -N/L	<u>Low Treatment:</u> 155µg PO <sub>4</sub> -P/L <u>High Treatment:</u> 1550µg PO <sub>4</sub> -P/L
Wetzel and van der Valk (1998)	<i>T. latifolia</i>	2000µg NO <sub>3</sub> /g soil	5 x 10 <sup>3</sup> µg available P/g soil <u>High Treatment:</u> 20 x 10 <sup>3</sup> µg available P/g
Woo and Zedler (2002)	<i>T. x glauca</i>	<u>Treatment Range:</u> 70 - 2240 X 10 <sup>3</sup> µg NH <sub>4</sub> NO <sub>3</sub> -N/L	<u>Treatment Range:</u> 20 - 640 X 10 <sup>3</sup> µg PO <sub>4</sub> -P/L

Documented total phosphorus (TP) observed in the highly degraded Cootes Paradise, a coastal wetland that has received point source pollution from a sewage treatment plant since 1919, agricultural and stormwater runoff and combined sewer overflows (Warren, 1950; Painter *et al.*, 1988; Chow-Fraser *et al.*, 1998) has in recent years averaged 210 µg TP/L (Court and Bowman, 2011). Mayer *et al.* (2008) documented TP levels of 1,655 – 1,987 mg/kg in Cootes Paradise sediment. Surface water TP and sediment TP levels in sections of the Point Pelee National Park *Typha* wetland located in Southwestern Ontario, Canada, and impacted by agricultural land use were 233 µg TP/L and, depending on the sampling site, just over or below 1200 mg TP/kg of sediment at the sediment/surface water interface.

### 1.7.2 Altered Hydroperiod

A primary objective of this study was to specifically examine the effects of eutrophic N and P levels on *T. latifolia* and under continuous, stabilized water levels (a change in water level by no more than approximately 4 cm). Several authors have attributed *Typha* invasions to stabilized and increased water levels (*e.g.* Wilcox *et al.*, 1985; Bradley and Wolf, 2005; Boers *et al.*, 2007; Wilcox *et al.*, 2008); however, the effects of stabilized water level was not an objective of this study. Seasonal water level fluctuations that follow an historic cycle contribute to wetland plant diversity by preventing the drying of sediments and encroachment by terrestrial tree and shrub species. Seasonal water level fluctuations also maintains habitat heterogeneity, facilitates expansion of sedge-grass meadow and replenishes the seed bank (Frieswyk and Zedler, 2007). Additionally, periodic flooding creates habitat for subdominant species by killing competitively superior plant species such as *Typha* (Keddy and Reznicek, 1986). Many non-rhizomatous annual wetland species re-establish only from seed and therefore requires seasonally low water levels and exposed sediments for seed germination and seedling survival (*e.g.* Keddy and Reznicek, 1986; Court and Bowman, 2011). Stabilized water levels can therefore promote competitively dominant *Typha spp.* that can maintain populations through vegetative regeneration from rhizome. Artificially maintained water levels at higher elevations can also result in the expansion of invasive wetland species inland (Wilcox *et al.*, 2008).

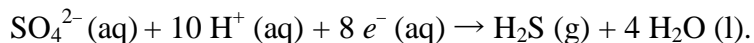
Research that has attempted to disentangle the individual effects of altered hydroperiod and nutrient loading where the two stressors co-occur to explain *Typha* invasion causation was not found in the literature. Wilcox *et al.*, (1985) and Wilcox *et al.*, (2008) who attributed *Typha* invasion to stabilized high water levels did not quantify nutrient levels. Boers *et al.* (2007) who attributed *T. x glauca* encroachment in a restored wetland to altered hydroperiod considered observed TP concentrations of 193 – 492 µg/L to be low and not a contributing factor in the expansion of *T. x glauca*; however, these TP levels far exceed the > 100µg TP/L minimum concentration considered hypereutrophic according to Environment Canada (2004).

### 1.7.3 Internal Eutrophication and Interaction between Hydroperiod Alternation and Internal Nutrient Loading.

Eutrophication can occur in aquatic systems not subjected to elevated external anthropogenic sources of P due a process termed internal eutrophication (Koerselman *et al.*, 1993; reviewed by Smolders *et al.*, 2006) involving elevated levels of sulfur (S), alkalinity and possibly increased  $\text{NO}_3^-$  which results in the transformation of existing unavailable P within sediments to bioavailable P and reduced sediment P retention capacity.

Increased deposition of sulfur (S) derived from the combustion of fossil fuels (*e.g.* coal) and gaseous ammonium ( $\text{NH}_3$ ) and ammonium sulfate aerosols have resulted in elevated levels of S and N in soils, surface water and groundwater in Europe and the mobilization of  $\text{PO}_4^{3-}$ , due to increased levels of  $\text{FeS}_x$  which reduces Fe availability for P retention and possibly due to competition between  $\text{PO}_4^{3-}$  and  $\text{SO}_4^{2-}$  for cation binding sites. Under anaerobic conditions the production of  $\text{H}_2\text{S}$  (Eq. 1) can result in the formation of FeS, further reducing Fe availability for P retention. To date, the significance of S deposition as a contributor to eutrophication in North America and *Typha* invasion has received very little attention.

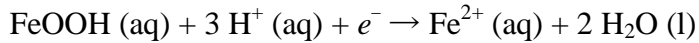
#### Eq. 1. Reduction of $\text{SO}_4^{2-}$ to $\text{H}_2\text{S}$



Oxidizing conditions in exposed sediments during naturally reduced water levels, including periodic draught conditions, can facilitate the removal of S through oxidization to soluble  $\text{SO}_4^{2-}$  and subsequent removal via wetland outflow during re-inundation (Smolders *et al.*, 2006). Artificially high or stabilized water levels may therefore also contribute to internal eutrophication. Boers and Zedler (2008) observed increased biomass and a higher P tissue content in *T. x. glauca* grown in microcosms subjected to continuous inundation which they attributed to internal eutrophication and greater P bioavailability. The elimination of natural water level fluctuations results in prolonged anaerobic conditions within sediments and the reduction of Fe hydroxides to soluble  $\text{Fe}^{2+}$  cation which has a lower P retention ability than  $\text{Fe}^{3+}$

due to a lower electrical charge difference between the  $\text{PO}_4^{3-}$  anion and Fe oxides and hydroxides (Eq. 2) (Smolders *et al.*, 2006):

**Eq. 2.** Reduction of iron oxide-hydroxide to  $\text{Fe}^{2+}$



Acidic conditions produced in anoxic sediments as a result of anaerobic microbial organic acid production reduce the decomposition rate of organic matter. An increase alkalinity as a result of oxygen release into the rhizosphere by macrophytes described further in Section 1.11 and the oxidization of  $\text{SO}_4^{2-}$  and denitrification, depending on the specific chemical and/or microbial process consumes  $\text{H}^+$  and/or produces  $\text{OH}^-$  (Smolders *et al.*, 2006). The release of oxygen into the rhizosphere can result in the neutralization of acids which inhibit decomposition and therefore can increase the mineralization rate of organic matter to bioavailable  $\text{NH}_4^+$ ,  $\text{PO}_4^{3-}$  and other nutrients.  $\text{HCO}_3^-$  may also compete for cation binding sites (Smolders *et al.*, 2006). Increases in alkalinity may occur due to changes in surface water and groundwater hydrology and possibly some industrial and land use practices may therefore contribute to *Typha* invasion through internal eutrophication.

Smolders *et al.* (2006) identified increased  $\text{NO}_3^-$  concentrations in surface and groundwater derived from agricultural fertilizers as a potential cause of internal eutrophication due to (1) increased availability of an electron acceptor for the microbial decomposition of organic matter; (2) the oxidization of pyrite ( $\text{FeS}_2$ ) by  $\text{NO}_3^-$  to  $\text{SO}_4^{2-}$  in ground water and subsequent discharge into surface water which may increase the bioavailable P in wetland sediments. The production of  $\text{OH}^-$  through ammonia ionization from  $\text{NH}_3$  to  $\text{NH}_4^+$  (Eq. 4) and nitrate dissimilation (Kadlec and Wallace, 2009) may also increase alkalinity and therefore the decomposition rate of organic matter. Conversely,  $\text{NO}_3^-$  may inhibit internal eutrophication as it is a preferred  $e^-$  acceptor over Fe and may therefore increase Fe availability for P retention (Smolders *et al.*, 2006).

Internal eutrophication may not occur in all wetlands subjected to altered hydroperiod. (Koerselman *et al.*, 1993) hypothesized, based on their lab results, that internal eutrophication

may be dependent on surface water chemistry, sediment characteristics (organic content, nutrient levels, buffering capacity) and elevated temperatures (global warming) which increases microbial metabolism and the mineralization rate of organic matter. To quantify the effects of internal eutrophication on *Typha* invasions, nutrient levels in pore water would need to be quantified in relation to timing, extent and duration of inundation. Pore water is defined as the water that occurs within the interstitial spaces between sediment particles. *T. latifolia* and other emergent macrophytes and under most conditions submergent species obtain nutrients almost exclusively from pore water (Bole and Allan, 1978; Carignan and Kalff, 1980; Reddy and D'Angelo, 1997).

#### **1.7.4 Other Potentially Contributing Factors to *Typha* Invasion**

Other potentially contributing factors to *Typha* invasions may also include increased levels of carbon dioxide (CO<sub>2</sub>) (Sullivan *et al.*, 2010) which would increase available carbon for biomass production, global warming, reduced (muskrat) grazing pressure and fire suppression. How ecosystem changes associated with eutrophication such as altered nutrient cycling, pH, sediment redox conditions, microbial communities and species composition and interaction may contribute to *Typha* invasion have not been assessed.

### **1.8 Potential Impacts of *Typha latifolia* Invasions.**

#### **1.8.1 Ecological Impact**

Despite the protection natural wetlands now receive (*e.g.* the *North American Wetlands Protection Act; Wetlands Reserve Program* (US)), habitat degradation continues in which invasive species are a major contributor (*e.g.*, Drexler and Bedford, 2002; Gilbert, 2012). Ecological impacts includes losses of native and ecologically significant vegetation (*e.g.*, listed species, species of conservation concern, ecotypes such as sedge-meadow habitat, plants adapted to nutrient-poor conditions (*e.g.*, sawgrass (*Cladium jamaicense*), isoetid submergent plants), shoreline bird breeding habitat, waterfowl breeding and open water rearing habitat (Addy and MacNamara, 1948; Curtis, 1959; Linde *et al.*, 1976; Beule, 1979; Newman *et al.*, 1996; Smolders *et al.*, 2002; Asamoah and Bork, 2010; Meyer *et al.*, 2010). The significance of effect

of *Typha* invasions upon macroinvertebrate, fish, amphibian and reptile populations was not located in the peer-reviewed literature but has likely been significant given that losses in plant species richness and diversity likely equates to losses in critical habitat for wildlife dependent on wetlands. *Typha spp.* have high transpiration rates (Otis, 1914) and *Typha* invasion has been shown to reduce open water habitat (e.g. Addy and MacNamara, 1948; Beule, 1979; Linde *et al.*, 1976; Mitch 2000) accelerating the transition from wetland to terrestrial habitat (e.g. Wilcox *et al.*, 1985). Gilbert (2012) has documented turtle mortality caused by the drying of wetland habitat resulting from transpiration through dense stands of invading *Pragmites australis australis* in southern Ontario.

Efficient piscivores (predatory fish) such as northern pike (*Esox lucius*) can be effective in controlling benthivorous and zooplanktivorous fishes such as common carp (*Cyprinus carpio*) and alewife (*Alosa pseudoharengus*). Reductions in zooplanktivorous fishes can enable large-bodied and efficient zooplankton grazers such as *Daphnia spp.* to increase in number and size to lower phytoplankton biomass and return turbid algae-dominated eutrophicated systems to a clear-water macrophyte-dominated state (e.g. Shapiro, 1990; Donk *et al.*, 1990; Meijer *et al.*, 1995; Lougheed *et al.*, 2004). Sedges such as *Schoenoplectus spp.* and *Carex spp.* provide important spawning habitat for northern pike (Cooper *et al.*, 2008) whereas Casselman and Lewis (1996) observed limited use of *Typha* by spawning pike.

## **1.9 Limitations of Control Methods for *Typha spp.***

A common method for controlling *Typha* is the application of chemical herbicides such as glyphosate (e.g. Linz and Homan, 2011). However, acute toxicity effects and physiological abnormality in chronic toxicity tests were observed in several life-history stages of four Ontario frog species to which Howe *et al.* (2004) attributed the combined effects of the surfactant carrier chemicals and the glyphosate compound found in common commercially-sold herbicides (e.g. Roundup Original<sup>®</sup>). Alternative methods to chemical herbicides such as direct physical removal, water level manipulation (drowning and/or desiccation) and controlled burns could have minor to severe impacts on other plant and animal species occurring within the same wetland depending on the method, extent and duration of removal, habitat requirements of individual species and specific life history stages affected at the time of removal. *Typha* removal



in its self would be a disturbance and would thus create an opportunity for the establishment of additional invasive species. A labour-intensive but low-impact method for reducing *Typha* abundance involves the bending or cutting of shoots below the water surface approximately 3 times per growing season which deprives roots of oxygen and eventually kills the plant (Sale and Wetzel, 1983). This method could however significantly reduce the amount of oxygen within the rhizosphere if cutting is done on a large scale. Additionally, the creation of reducing conditions within the sediment as a result of cutting could potentially result in the mobilization of sediment-bound P.

Potential biological control agents of *Typha spp.* have not been adequately evaluated. Beule (1979) noted that deer ate young *Typha* shoots developing in exposed mud flats and muskrats that gained access to cut areas of dense *Typha* stands were able to increase the area of open-water habitat through grazing. Curtis (1959) observed none to low numbers of *Typha* where muskrat population density was high. Muskrat introductions, if proven successful, may offer the most ecologically sound method of effective *Typha* control; however, there is also the potential for introducing disease and impact to other animal and plant species. The release of cattail armyworm is also a possible option; however, as with muskrat introductions, a substantial investment would be necessary for adequate evaluation and risk assessment.

*Typha* control programs can be economically driven. Unsustainable financial losses to sunflower growers in the prairie pothole region to consumption by pest bird species that nest and roost in *Typha* was estimated to be in the range of \$4 million to \$11 million dollars annually (Peer *et al.*, 2003). The widespread establishment and dominance of *Typha spp.* has provided habitat for millions of red-winged blackbirds, common grackle and yellow-winged blackbirds which feed on sunflower seeds (Linz and Homan, 2011).

Regardless of the removal method used, re-planting with carefully selected non-invasive species would be necessary to prevent or reduce re-colonization by *Typha* and other invasive species. The difficulty, expense and potential impacts associated with *Typha* control emphasizes the need to implement preventative measures by reducing disturbances that contributes to *Typha* invasions.

## 1.10 Wastewater Treatment Wetlands

Hammer *et al.* (1989) defined wastewater constructed wetlands (hence-forth referred to as constructed wetlands) as man-made or converted natural wetlands which are used to remove pollutants by the physical, chemical and biological processes that occur within wetland ecosystems. Widely accepted as a cost-effective means of removing surface water pollutants, the number of constructed wetlands (CW) has steadily increased globally in all climactic regions since the 1970's (Brix and Schierup, 1989; Kadlec and Wallace, 2009). Requiring only inexpensive and locally available materials, low technology and minimal operating skills, constructed wetlands are particularly well suited to smaller municipalities, rural communities and third world nations which have an availability of affordable lands and/or lack the monetary resources needed to build and operate technologically sophisticated and expensive wastewater treatment plants (WWTP) and the necessary sewage and stormwater conveyance infrastructure (Brix and Schierup, 1989; Kadlec and Wallace, 2009).

Constructed wetlands can also provide a practical and cost-effective means of treating non-point source runoff from urban, agricultural, rural and industrial lands (*e.g.* Taylor, 1992), areas where WWTP and conveyance systems may be impractical or cost-prohibitive. Constructed wetlands have also been developed for individual industrial operations such as olive mill, tannery, mining and slaughter house effluent (*e.g.* White *et al.*, 2000; Calheiros *et al.*, 2009; Marchand *et al.*, 2010; Yalcuk *et al.*, 2010) and individual households (Fraser *et al.*, 2004) where early interception and treatment of wastewater can potentially provide greater environmental protection at an affordable cost to both industry and society.

Constructed wetlands can be effective in removing pathogens (bacteria such as *E.coli* and fecal coliforms, helminthes, protozoans, fungi and viruses), organic matter and therefore BOD, COD, suspended sediments, nutrients, metals and toxic industrial and organic chemicals in all seasons and in a variety of climates. A tremendous volume of research now exists on constructed wetland design and performance from sources ranging from household grey water, sewage, industrial effluent, acid mine drainage and urban and agricultural runoff (EPA, 2000; Fraser *et al.*, 2004; Kadlec and Wallace, 2009).

Constructed wetlands have been built across Canada and by 1997 two wetlands had been constructed in subarctic regions and eight in the boreal forest region of the Yukon and the North West Territories. Pollutant removal has been effective during the winter months (Pries, 1997); however, pollutant removal processes can cease if the wetland freezes completely (Kadlec and Wallace, 2009).

Pollutant removal performance has varied from poor to excellent with key factors being pollutant loading rate, hydraulic retention time and wetland area (see reviews by Nichols, 1983; Brix and Schierup, 1989; Taylor, 1992; Brix, 1997; Reddy *et al.*, 1999; Sheoran and Sheoran, 2006; Vymazal, 2007; Lee *et al.*, 2009; Marchand *et al.*, 2010). The life expectancy of constructed wetlands varies depending on a number of factors including loading rate, the size of the wetland, the physical and geochemical characteristics of the substrate and the microbial communities which are essential for transforming pollutants into non-harmful forms. Constructed wetlands designed for industrial use may provide effective removal capacity for a decade whereas wetlands receiving Municipal wastewater may last for centuries (Pries, 1997).

Constructed wetlands could also potentially serve as mitigation for the loss of natural wetlands (Brix, 1997) as the loss of wetlands in North America has been extensive. In Southern Ontario 70 to 100 % of natural wetlands has been lost depending on the region (Whillans, 1982; Curtis, 1989; Snell, 1989). Approximately 90 % of wetlands have been lost primarily to agriculture in the prairie pothole region (Dahl and Johnson, 1991). The building of constructed wetlands in developed areas at a watershed scale could therefore potentially serve to restore natural flood control processes, improve stream and river hydrology and ground water recharge, promote riparian vegetation recovery which in turn could improve water quality, improve aquatic habitat and increase biological diversity. Constructed wetlands designed for wastewater treatment can also provide ancillary benefits including the creation of fish and wildlife habitat, recreational and educational opportunities and aesthetic value (*e.g.*, Taylor, 1992; Brix, 1997; EPA, 2000; Kadlec and Wallace, 2009).

Emergent vegetation produced in wetlands has been used to manufacture various types of building materials (Maddison *et al.*, 2009) and as fertilizer and biofuel (Agricultural Wetland

Research Network, 2011). With a predicted shortage of phosphate rock in the near future (Cordell *et al.*, 2009; Elser and Bennett, 2011) the harvesting and subsequent utilization of emergent vegetation and the P that that accumulates within constructed wetlands sediments could provide a means of P recovery.

### 1.10.1 The History of Constructed Wetlands for the Treatment of Wastewater

Natural wetlands have been used as convenient disposal sites for wastewater ever since effluent has been collected. Several North American wetlands have been receiving sewage for over 100 years (Kadlec and Wallace, 2009). The wetland formerly located in Ashbridges Bay, Toronto Ontario, Canada (Figure 8) historically received conveyed stormwater runoff and wastewater from animal and livestock holding areas (Bonnell and Fortin, 2009) and later from the Main Treatment Plant which became operational in 1910 (City of Toronto, 2013a). The Cootes Paradise marsh located within the tri-city area of Hamilton, Burlington and Dundas, Ontario, Canada, has been the repository for the Dundas Sewage Treatment Plant since 1919 (Painter *et al.*, 1988). Beining and Otte (1997) observed significant removal of arsenic (As) and Zinc (Zn) by a natural wetland in Ireland that had been receiving drainage from an abandoned mine since 1824.



**Figure 8. Ashbridges Bay wetland in 1909 before the completion of the Main Treatment Plant.**

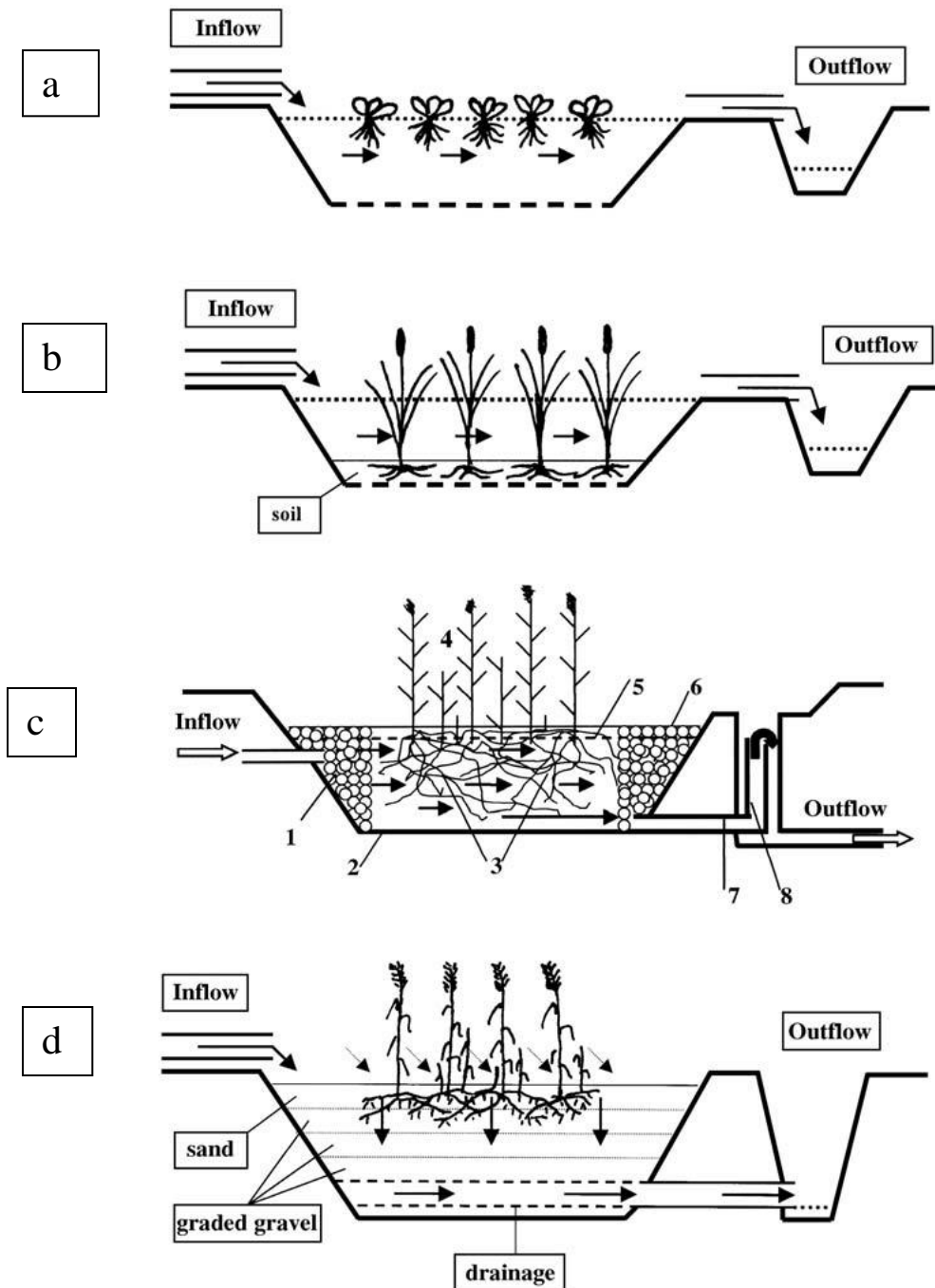
*Courtesy of City of Toronto Archives.*

Wetlands have been constructed as aquaculture ponds, rice paddies, ornamental wetlands, road-side ditches *etc*, all of which have been built for centuries (Taylor, 1992). Dr. Käthe Seidel (1907-1990) has been credited with conceiving the idea of constructing wetlands for wastewater treatment (Kadlec and Wallace, 2009). Dr. Reinhol Kickuth, a colleague of Dr. Seidel, has also been credited with contributing to Dr. Seidel's pioneering research (Lee *et al.*, 2009) which began in 1952 at the Max Planck Institute in West Germany. By the 1960's Dr. Seidel's work had confirmed the effectiveness of constructed wetlands as an effective pollutant removal technique. Research on wastewater treatment constructed wetlands began in the western hemisphere in 1971 by Dr. Robert Kadlec at the University of Michigan (Kadlec and Wallace, 2009).

### **1.10.2 Types of Constructed Wetlands**

Constructed wetlands can be classified as free surface flow constructed wetlands (FWS) which resemble natural marshes (EPA, 2000; Kadlec and Knight, 2009) and subsurface constructed wetlands (SSFCW) which can be further classified as vertical flow (VF) and horizontal flow (HF) designs (*e.g.* Vymazal and Kropfelova, 2011) (Figure 9). Floating vegetation wetlands (FVW) are used in tropical climates and typically use water hyacinth (*Eichhornia crassipes*) which has been shown to have a higher capacity for removing nutrients compared to rooted macrophytes (Vymazal, 2007). As floating vegetation wetlands are not relevant to *T. latifolia* invasion and temperate climates, literature referenced in this thesis pertain primarily to FWS and secondarily to VF and VH wetlands.

In SSFCW effluent enters directly into the substrate and vegetative root systems (Van de Moortel *et al.*, 2009; Yalcuk *et al.*, 2010). VF and HF wetlands are designed for specific hydraulic loading rates, thus are not capable of treating stormwater runoff or dynamic flow changes. Though young SSFW with sufficient P adsorption sites can be effective in removing P (Vrhovsek *et al.*, 1996), the primary limitation of SSFCW designs for P removal is the inability to accrete peat biomass (Vymazal, 2007). Peat is partially decomposed plant matter which can retain nutrients and other pollutants. Peat accretion and subsequent burial is now generally accepted as the only long- term P removal mechanism in FWS (discussed in a later section).



**Figure 9. Floating vegetation (a), surface flow (b) subsurface horizontal flow (c) and subsurface vertical flow (d) constructed wetlands.**

*From Vymazal (2007).*

In SSFCW wetlands vegetation is planted into topsoil which is not inundated, therefore significant peat accretion, which develops from the decomposition of wetland vegetation, does

not occur (Vymazal, 2007). VF constructed wetlands are effective at oxidizing  $\text{NH}_4^+$  to  $\text{NO}_3^-$  but have low ability to remove  $\text{NO}_3^-$  (Vymazal, 2007) possibly due to higher aerobic conditions within the sediments. Thus, hybrid constructed wetland systems which combine both VF and HF wetlands have been utilized to achieve adequate N removal (Kadlec and Wallace, 2009).

### **1.10.3 Constructed Wetlands for the Restoration of Eutrophic Systems**

The use of constructed wetlands for restoring eutrophicated freshwater ecosystems has long been considered (*e.g.* Boyd, 1970; Yount and Crossman, 1970; Brix and Schierup, 1989; Theÿsmeÿer *et al.*, 1999). Feasibility or pilot studies on the potential suitability of constructed wetlands for the restoration of eutrophicated lakes, either as a stand-alone strategy or in combination with other restoration efforts, have recently been conducted by Li *et al.* (2008), Ham *et al.* (2010) and Özkundack *et al.* (2010). There has also been some interest in developing constructed wetlands at the watershed scale to provide flood control and nutrient removal services to prevent or reverse the impacts of eutrophication on coastal marine ecosystems (*e.g.* Boesch *et al.*, 2001; Arheimer and Wittgren, 2002; Zedler, 2003; Woo, 2009; Gren, 2010 and Kim, 2010). If constructed wetlands are developed at the watershed scale to reduce non-point source pollution, the risk of spreading invasive species to ecologically sensitive areas may therefore increase. Ecologically sensitive areas are defined here as areas or ecosystems that support endemic, rare, threatened, endangered species or species of special concern, unique or important ecosystems including wetlands classified as significant or high biological diversity warrant protection (Environment Canada, 2013).

### **1.10.4 Nitrogen and Phosphorus Removal Limitations of Constructed Wetlands**

It is generally known that P removal capacity often declines as constructed wetlands age and P binding sites become saturated (*e.g.* Richardson, 1985; EPA, 2000, Vymazal, 2007). The TP objective for surface waters set by the Ontario Ministry of Environment in Ontario (MOE) is as low as 20 µg TP/L for lakes during the growing season, depending on the historic lake trophic status, and 30 µg TP/L for streams (MOE *et al.*, 2010). Vymazal (2007) provided the TP removal rates for 85 mature surface flow wetlands. The mean TP concentration in influent was 4.4 mg P/L and the mean TP in effluent was 2.15 mg P/L for a mean removal rate of 48.8 %.

The data presented in Vymazal (2007) indicates that at typical P loading and removal rates, constructed wetlands, using currently technology, are unable to reduce P to levels that would ensure no ecological impact unless to receiving water bodies. Even if effluent becomes sufficiently diluted, the tendency for particulate P to settle at inflow outfall areas suggests that even if effluent P eventually becomes diluted to low levels, localized impacts and the establishment of invasive species may still occur within the inflow area of receiving aquatic systems. The mean P loading rate in the influent for 251 constructed wetlands reviewed by Vymazal (2007) was 151.3 grams (g) P/ square meter (m<sup>2</sup>)/year (yr) with a mean removal of 47.9 % giving a mean of 86.25 g P/m<sup>2</sup>/yr in effluent (ranging from 54 – 127 g P/m<sup>2</sup>/yr which is nearly seven times greater than the mean TP loading rate of 13.1 g P/m<sup>2</sup>/yr observed by Jeppensen *et al.* (1991) for shallow culturally eutrophicated lakes in Denmark indicating that effluent from constructed wetlands used for nutrient removal would, in most cases, still contain ecologically impacting levels of P.

Similarly, N removal performance by constructed wetlands is variable and sometimes does not meet water quality objectives or standards (reviewed by Lee *et al.*, 2009). James *et al.* (2005) found that diverse submergent plant communities in the UK occurred only in lakes with NO<sub>3</sub>-N levels of 1-2 mg/L, a concentration that is likely unachievable for most constructed wetlands, particularly those subjected to moderate to high loading rates. Based on data presented in Vymazal (2007) the typical N removal rate for mature wetlands was 40 – 55 % at a mean influent loading rate of 792.5 g N/m<sup>2</sup>/yr (ranging from 466 to 1222 g N/m<sup>2</sup>/yr ) giving a mean of 409 g N/m<sup>2</sup>/yr constructed water effluent (ranging from 219 – 592 g N/m<sup>2</sup>/yr). Jeppensen (1991) observed a loading rate of 142.0 g N/m<sup>2</sup>/yr for shallow lakes impacted by cultural eutrophication in Denmark which is substantially less than the N loading in constructed wetland effluent documented by Vymazal (2007). As the majority of natural wetland ecosystems are N limited (Bedford *et al.*, 1999) and constructed wetlands can only remove a percentage of the N loading, constructed wetland effluent would likely contain N concentrations sufficiently elevated to impact receiving aquatic ecosystems and potentially contribute to the establishment and expansion of invasive species unless the effluent is substantially diluted immediately by a receiving water body.



### **1.10.5 *Typha* Productivity in Wastewater and Constructed Wetlands Effluent**

Several studies have documented increased *T. latifolia* biomass in wetlands subjected to wastewater effluent or levels characteristic of wastewater entering constructed wetlands and drainages (*e.g.* Martín and Fernández, 1992; Weng *et al.*, 2006; Maddison *et al.*, 2009). Martín and Fernández (1992) documented a significant increase in *T. latifolia* biomass within a river channel receiving secondarily treated wastewater with mean concentrations of 6.4mg PO<sub>4</sub>-P/L, 11.6 mg NH<sub>4</sub>-N /L and 6.4mg NO<sub>3</sub>-N/L, which far exceeds hypereutrophic levels, as well as elevated levels of magnesium (Mg), potassium (K) and calcium (Ca). The research documenting high *Typha* production within constructed wetlands (Maddison *et al.*, 2009) simulated constructed wetlands influent (Weng *et al.*, 2006) or in downstream wetlands receiving wastewater effluent (Martín and Fernández, 1992;) suggests that *Typha* could establish highly productive monotypic stands in constructed wetland effluent given suitable hydrology and substrates.

### **1.10.6 Hybridization and Formation of *T. x glauca* populations.**

There is the potential risk that constructed wetlands planted with *T. latifolia* will hybridize with the expanding *T. angustifolia* and form new highly invasive *T. x glauca* populations. The only document found that had tested for hybridization in a constructed wetland planted with both *T. latifolia* and *T. angustifolia* was by Selbo and Snow (2004). Although no evidence of hybridization was found, there appears to be a reasonably high probability that future hybridizations will occur in or adjacent to constructed wetlands planted with *T. latifolia*.

## **1.11 The Role of Wetland Vegetation in Nitrogen and Phosphorus Removal Processes – An Overview**

The first objective of this section is to summarize the most significant contribution of aquatic macrophytes to N and P removal processes in constructed wetlands. Only the most important removal pathways are discussed in detail. Diagrams depicting the numerous N and P pollutant are provided for completeness in order to illustrate to the reader the overall complexity of pollutant pathways within constructed wetlands. The second objective of this section is to

present evidence to support of the hypothesis that, given that the direct macrophyte uptake of nutrients represents a comparatively minor removal pathway, constructed wetlands planted with native non-invasive plant species should provide comparable nutrient removal performance to constructed wetlands planted with invasive species. Constructed wetlands planted with non-invasive native species should therefore offer effective water treatment services without the ecological risks associated with planting constructed wetlands with invasive species. A detailed discussion on the pollutant removal processes in wetlands for other contaminant types including suspended solids, organic matter, sulfur, metals, halogens and organic chemicals is beyond the scope of this thesis.

Wetlands are effective as pollutant-removing systems owing to their high productivity and the combination of aerobic and anaerobic sediments in which emergent wetland vegetation play a critical role (reviewed by Nichols, 1983; Brix, 1994; Brix and Schierup, 1989; Brix, 1997). Wetland emergent plants are uniquely adapted to survive in flooded anoxic sediments owing to their internal aerenchyma system. Aerenchyma tissue in *Typha spp* consists of air-filled spaces (lucunar) within leaves (Figure 9) and rhizomes that enables gas exchange to occur between the roots and the atmosphere (e.g. Sale and Wetzel, 1983; Tornbjerg *et al.*, 1994). In *T. latifolia* (and



**Figure 10. *Typha latifolia* leaf aerenchyma tissue from a leaf cross sectional (right) and horizontal view (Left). Magnification X 2.**

*Photos by Mark Tiley.*

*T. angustifolia*), air enters through the stomata of middle-aged leaves against a small pressure gradient and then diffuses through the leaf blade and rhizome aerenchyma into root tissue through which excess oxygen escapes into the sediments (Tornbjerg *et al.*, 1994).

The amount of oxygen escape into the rhizosphere depends on the oxygen demand of the sediments, oxygen concentration within the plant and roots and root age. Older root tissue becomes impermeable to gas exchange in order to maximize oxygen transport to the root apical meristem; thus, oxygen escape occurs primarily at the root tip (Brix, 1994). Gases produced by respiration and metabolism and gases which diffuse into the root system from the sediments (CO<sub>2</sub>, NH<sub>3</sub>, H<sub>2</sub>S) pass through old or damaged leaves (Tornbjerg *et al.*, 1994). More oxygen enters the roots than is needed and diffuses through root tissue into the rhizosphere where it becomes available to aerobic microbial communities within the rhizosphere (Sale and Wetzel, 1983; Kadlec and Wallace, 2009). Dead culms (flowering stems) and leaves enables gas exchange to continue through the winter months, which is essential for rhizome survival (Linde *et al.*, 1976) and enables aerobic microbial transformation of pollutants to continue throughout the year.

The dense macrophyte root systems provide increased surface area for the colonization of microbial communities involved in the transformation of organic matter and nutrients (*e.g.* Brix, 1994; Vymazal, 2007), toxic industrial organic contaminants (Reddy and D'Angelo, 1997) and metals. Metals undergo similar chemical and biological transformations through the formation of compound complexes, hydrolysis microbial oxidation and microbial and plant uptake and assimilation as to other contaminants (Sheoran and Sheoran. 2006; Marchand *et al.*, 2010). Plant photosynthates leached from roots and decaying vegetation are the primary carbon sources for pollutant-transforming microbes which are critical to the pollutant-removal performance of constructed wetlands (*e.g.* Brix, 1994; Kadlec and Wallace, 2009). Root penetration loosens sediments and after death and decay, leaves channels (macropores) which increases hydraulic conductivity and the ability for water to be channel through the root bed (Brix, 1994). Macropores may increase microbial surface area, oxygenation of the sediments and dispersal of nutrients and contaminants and therefore increase pollutant removal efficiency.

Dead plant matter provides adsorptive surface area for microbes, nutrients, metals and some organic industrial compounds. Peat can contain a significant amount of absorbed and adsorbed nutrients and contaminants. The burial or retention of peat within the sediments (peat accretion) is the only long-term removal pathway for P (*e.g.* Nichols, 1983; Vymazal, 2007).

Under anaerobic conditions, anaerobes reduce the pH of the sediments through the production of organic acids which slows the decomposition rate of organic matter and therefore increases P retention (Smolders *et al.*, 2006). Dead plant matter has a cooling effect during the summer and can prevent the sediment from freezing during the winter months (Brix, 1997). Tall species such as *Phragmites*, *Typha spp.* and *Schoenoplectus spp.* can also reduce algal production through shading which can improve water quality (Brix, 1997).

In addition to providing greater ancillary benefits including wildlife habitat, education, recreation and aesthetics, a greater plant species diversity in constructed wetlands may also increase pollutant removal capability owing to species-specific differences in pollutant tolerances, disease and insect infestations (Brix and Scheirup, 1989; Taylor, 1992; Kadlec and Wallace, 2009). Engelhardt and Ritchie (2001) found higher TP removal rates in microcosms containing higher submergent plant species diversity compared to single-species controls. A higher level of increased biodiversity would also provide greater resilience to variations in water levels, episodic weather events and herbivore pest out-breaks and thus provide some insurance against poor pollutant – removal performance or failure (Taylor, 1992; Kadlec and Wallace, 2009).

## **1.12 Nitrogen**

Sources of anthropogenic N include sewage effluent, urban and agricultural runoff, crop and garden fertilizers. Atmospheric nitrogen oxide pollutants (NO and NO<sub>2</sub>) produced by the combustion of fossil fuels can enter aquatic ecosystems via atmospheric dry deposition or washout (Kelly *et al.*, 1990; Camargo *et al.*, 2005). Increased levels of nitric acid (HNO<sub>3</sub>) and NO<sub>3</sub><sup>-</sup> derived from the oxidation of increased nitric acid deposition can result in elevated nitrate levels in lakes and lower pH (Kelly *et al.*, 1990), potentially increasing the bioavailability of toxic metals and P. Unionized ammonia (NH<sub>3</sub>), formed by the hydrolysis of urea and decomposition of amino acids, is highly toxic to fish and other aquatic organisms (Kadlec and Wallace, 2009). Ammonium or ionized ammonia (NH<sub>4</sub><sup>+</sup>), NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup> ions can also be toxic to aquatic animals and humans (reviews by Ip *et al.*, 2001; Camargo *et al.*, 2005).

The eutrophication of estuarine and marine coastal ecosystems in response to elevated N loading largely derived from agricultural fertilizer is well documented (*e.g.* Boesch, 2001; Zedler, 2003; Smith and Schindler, 2009). Wetland ecosystems are often N limited (Bedford *et al.*, 1999) and high N loading can also contribute to the eutrophication of other freshwater ecosystems (Elser *et al.*, 1990; González *et al.*, 2005) thus effective removal of N from various anthropogenic sources is often a primary objective of constructed wetlands (Sauders and Kalff, 2001; Kadlec and Wallace, 2009).  $\text{NH}_4^+$  and  $\text{NO}_3^-$  are generally the only bioavailable forms of N which can be assimilated into aquatic macrophyte and algal biomass. The increased growth of N fixing cyanobacteria (heterocyst-forming blue-green algae) under high P availability may potentially also increase the amount of atmospheric N transfer into surface waters where it can be oxidized into bioavailable nitrate ( $\text{NO}_3^-$ ), further increasing algal production and the negative effects of eutrophication (Smith and Schindler, 2009).

N removal processes in wetlands includes ammonification, ammonia volatilization, nitrification, aerobic and anaerobic denitrification, anaerobic ammonia oxidization and reduction of  $\text{NO}_2^-$  to  $\text{N}_2$  (anammox), adsorption to peat, permanent burial of organic N, and uptake of  $\text{NO}_3^-$  and  $\text{NH}_4^+$  by macrophytes, algae and microbes (*e.g.*, van Kessel, 1978; Wetzel, 1983; Vymazal, 2007; Kadlec and Wallace, 2009) (Figure 11). The mean  $\text{NO}_3^-$  removal rate of surface flow and horizontal subsurface flow wetlands was 47.4 % with a mean inflow influent concentration of 5 mg  $\text{NO}_3^-$ -N /L.  $\text{NO}_3^-$  actually increased in vertical flow subsurface wetlands which are not considered effective for  $\text{NO}_3^-$  removal (Vymazal, 2007). The mean  $\text{NH}_4^+$  removal rates reported in Vymazal (2007) for 295 constructed wetlands of all types except floating vegetation wetlands was 62.52 % with a mean influent concentration of 35.6 mg  $\text{NH}_4^+$ -N/L. The removal rates for both  $\text{NO}_3^-$  and  $\text{NH}_4^+$  are thus insufficient to prevent ecological impact to receiving systems unless further treatment or significant dilution occurs.

The primary contribution to N removal by wetland vegetation is physical whereby macrophytes reduce water velocity which increases the sedimentation rate (deposition to the sediments) of particulate inorganic and organic N allowing ammonification, nitrification, denitrification and burial processes to permanently remove N. Macrophytes also increase the surface area for N-removing biofilms which achieve greater biomass on solid surfaces such as

live and dead vegetation (Craft and Richardson, 1993; Saunders and Kalff, 2001). As is the case for P, the removal rate of N is determined by hydraulic loading rate (Brix and Schierup, 1989).

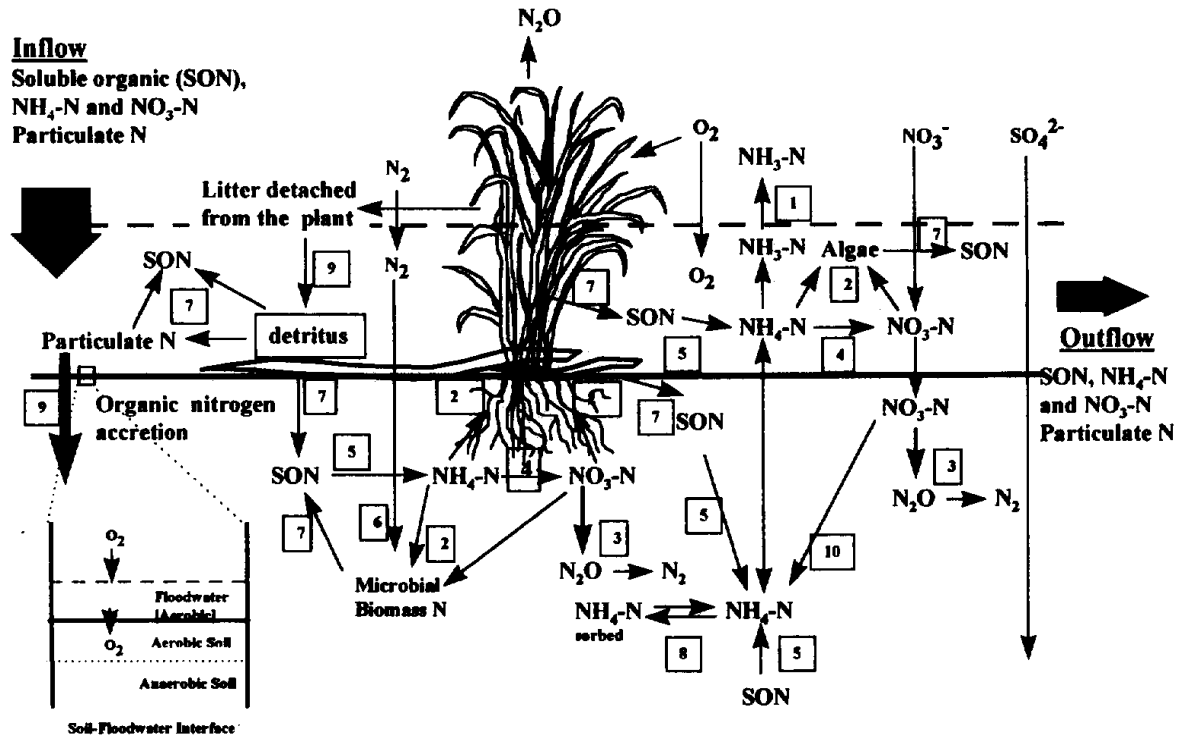


Figure 11. Nitrogen transformations in constructed wetlands.

- (1) volatilization, (2) plant and microbial uptake, (3) denitrification, (4) nitrification, (5) mineralization, (6) nitrogen fixation, (7) fragmentation and leaching, (8) sorption and decomposition, (9) burial and (10) nitrate reduction to ammonium.

From Reddy and D' Angelo. 1997.

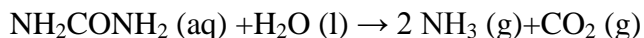
### 1.12.1 Ammonification

$\text{NH}_3$  is rapidly converted to  $\text{NH}_4^+$  under conditions of moderate pH and temperatures  $< 30^\circ\text{C}$  which predominate in wetlands (Kadlec and Wallace, 2009).  $\text{NH}_3$  concentrations can however rapidly accumulate under anoxic conditions (Wetzel, 1983); or at very high pH ( $>9.0$ ) and temperatures of approximately  $30^\circ\text{C}$  (Kadlec and Wallace, 2009). In constructed wetlands where high  $\text{NH}_3$  does not pose a threat to fish and other aquatic organisms, submergent

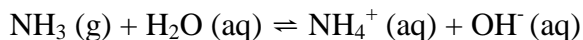
macrophytes can be used to elevate pH by consuming CO<sub>2</sub> during photosynthesis and increase the amount of N removed by NH<sub>3</sub> volatilization (Brix and Schierup, 1989).

The ammonification process (Eq. 3 and Eq. 4) begins with the release of unionized and volatile ammonia (NH<sub>3</sub>) as a result of organic N decomposition by chemotropic and heterotrophic microorganisms under both aerobic and anaerobic conditions (Kadlec and Wallace, 2009).

**Eq. 3.** Ammonification (*e.g.* urea breakdown).



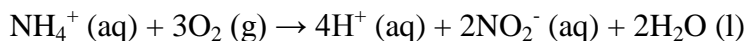
**Eq. 4.** Ammonia ionization/deionization.



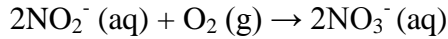
### 1.12.2 Nitrification

Nitrification is the oxidation of reduced forms of N which largely occurs under aerobic conditions and is the primary mechanism for reducing NH<sub>4</sub><sup>+</sup> in wetlands. Nitrification of NH<sub>4</sub><sup>+</sup> (Eq. 5 and Eq. 6) involves the successive oxidation of NH<sub>4</sub><sup>+</sup> to intermediate NO<sub>2</sub><sup>-</sup> and finally NO<sub>3</sub><sup>-</sup> by aerobic chemoautotrophic and heterotrophic bacteria for adenosine triphosphate (ATP) production and assimilation of CO<sub>2</sub> for growth. As the nitrification of ammonia consumes CaCO<sub>3</sub>, alkalinity decreases along with a reduction in pH (Kadlec and Wallace, 2009).

**Eq. 5:** Nitritation by *Nitrosomonas*



**Eq. 6.** Nitrification by *Nitrobacter*

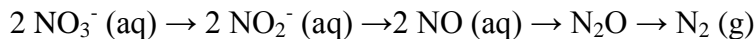


The oxidation of  $\text{NH}_4^+$  to intermediate nitrite ( $\text{NO}_2^-$ ) is primarily by *Nitrosomonas* bacteria followed by further oxidation to  $\text{NO}_3^-$  primarily by *Nitrobacter* bacteria with oxygen as the terminal electron ( $e^-$ ) acceptor (Wetzel, 1983). Oxygenation of the rhizosphere by macrophytes increases nitrification by stimulating the growth of nitrifying bacteria within the sediments.  $\text{NO}_3^-$  within the aerobic zone of the sediments then diffuses into the oxygen-scarce facultative anaerobe and anoxic obligate microbial zones where denitrification and permanent removal of N occurs (e.g. Reddy and D'Angelo, 1997; Kadlec and Wallace, 2009).

### 1.12.3 Denitrification

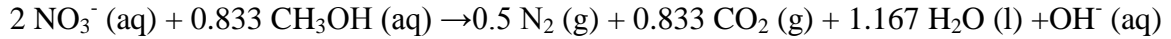
Denitrification (nitrate dissimilation) is the conversion of  $\text{NO}_3^-$  to  $\text{N}_2$  gas by denitrifying facultative heterotrophic organisms, particularly those that are anaerobic (e.g. *Pseudomonas*, *Vibrio*, *Aeromonas*, *Bacillus*, *Thiobacillus*, *Nitrosomonas*) (Eq. 7) which utilize nitrate as a terminal  $e^-$  acceptor. Denitrification by heterotrophic bacteria requires a carbon energy source generally in the form of organic matter (Vymazal, 2007) (Eq. 8) which, in constructed wetlands, can be provided by decomposing plant tissue and other organic molecules produced by macrophytes. Autotrophic ( $\text{CO}_2$  fixers) *Thiobacillus denitrificans* can reduce  $\text{NO}_3^-$  to  $\text{N}_2$  by utilizing elemental sulfur ( $\text{S}^0$ ), sulfide ( $\text{S}^{2-}$ ), thiosulfate ( $\text{S}_2\text{O}_3^{2-}$ ) and sulfite ( $\text{SO}_3^{3-}$ ) as terminal  $e^-$  acceptors (Kadlec and Wallace, 2009). Denitrification under anaerobic conditions can also occur in the absence of a carbon source through the reduction of  $\text{NO}_2^-$  to  $\text{N}_2$  by the oxidation of ammonia by the bacteria *Planctomycetes* and *Nitrosomas eutropha* (Eq. 9) (Kadlec and Wallace, 2009).

**Eq. 7:** Denitrification:

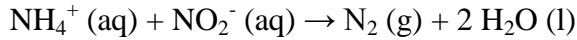




**Eq. 8.** Denitrification with methanol (Kadlec and Wallace, 2009).



**Eq. 9.** Anaerobic ammonia oxidation (anammox) (Kadlec and Wallace, 2009)



Denitrification is the primary N removal mechanism in lakes, streams and wetlands (Saunders and Kalff, 2001). Denitrification predominantly occurs under anoxic conditions but can also occur under aerobic conditions (Wetzel, 1983; Kadlec and Wallace, 2009). Denitrification interacts with other pollutant removal processes by generating alkalinity through the production of  $\text{OH}^-$  which raises pH and thereby affecting the decomposition rate of organic matter and the cycling and motility of metals (Marchand *et al.*, 2010) and P (Smolders *et al.*, 2006).

#### 1.12.4 Macrophyte Nitrogen Assimilation and Peat Accretion

Aquatic macrophyte species obtain N largely from sediment pore water although submergent species of vegetation can obtain a small amount of N from surface water (Reddy and D'Angelo, 1997). Most plant species can store and rapidly assimilate either  $\text{NO}_3^-$  or  $\text{NH}_4^+$  into plant tissue (Hirel and Lea, 2001). Direct ammonia uptake is more efficient for plant growth as energy is required for the successive reduction of  $\text{NO}_3^-$  to  $\text{NO}_2^-$  and  $\text{NH}_4^+$  (*e.g.* Wetzel, 1983; Hirel and Lea, 2001; Vymazal, 2007). The selection of either  $\text{NO}_3^-$  or  $\text{NH}_4^+$  by macrophytes is often determined by availability with nitrate reductase and nitrite reductase enzymes being produced when  $\text{NO}_3^-$  is more abundant (Vymazal, 2007; Kadlec and Wallace, 2009). Although *T. latifolia* is capable of assimilating either  $\text{NO}_3^-$  or  $\text{NH}_4^+$  Brix *et al.* (2002) observed greater *T. latifolia* growth in plants fed  $\text{NH}_4^+$  compared to plants provided with  $\text{NO}_3^-$ .

The amount of N uptake by macrophytes and algae is considered to be minor relative to N loading rates typically received by both natural (Nichols, 1983) and constructed wetlands (*e.g.* Mustafa *et al.*, 2011). Mustafa *et al.* (2011) reported that N uptake by *T. latifolia* accounted for

<1 % of N removed by a constructed wetland over a 12 month period. Emergent rhizomatous vegetation translocate a significant amount of carbohydrate and nutrient from the shoots to the rhizomes for overwinter storage (*e.g.* Dubbe *et al.*, 1988); however, similar to submergent vegetation, the amount of N retained within the rhizome structure is relatively minor (Brix, 1997; Sauders and Kalff, 2001; Mustafa *et al.*, 2011). While macrophyte N retention can be significant during the growing season (Brix, 1997; Kadlec and Wallace, 2009) most of the assimilated N is re-released after the growing season through decomposition following death (Vymazal, 2007; Kadlec and Wallace, 2009).

Retention by peat accretion in constructed wetlands is restricted to surface flow wetlands and is generally considered to be low (Vymazal, 2007). Accreted peat generally contains < 5 % N (Kadlec and Wallace, 2009). Kadlec and Wallace (2009) reported that the removal of N through peat accretion was in the order of 10 g N/m<sup>2</sup>/yr which may be of significance for constructed wetlands subjected to very low loading rates but insignificant in wetlands subjected loading rates typically received by constructed wetlands. Recall that the mean TN influent loading rate for 254 constructed wetlands of all types presented in Vymazal (2007) was 792.5 g N/m<sup>2</sup>/yr.

### **1.13 Phosphorus**

P is generally the nutrient most limiting to algal production in freshwater ecosystems and the primary cause of freshwater eutrophication (*e.g.*, Schindler, 1974; 1977; Reddy *et al.*, 1999; Smith and Schindler, 2009). P limitation has also been observed in estuarine ecosystems during certain times of the year (Boesch *et al.*, 2001). Anthropogenic P enters aquatic ecosystems in sewage effluent from wastewater treatment plants, combined sewer overflows, agricultural runoff containing livestock waste, fertilizer and farm field soils, stormwater runoff containing garden fertilizers and particulate P and atmospheric deposition of particulate P (wind-blown soil, dust, pollen and plant parts). Evidence of P loading in sediment cores collected from the Bay of Quinte dates to the late 17<sup>th</sup> century coinciding with early European settlement (Ahl, 1988). Ahl (1988) attributed increased P levels in centuries old Northern European sediments in remote areas of low human population density to the aerial transport of eroded agricultural soils.

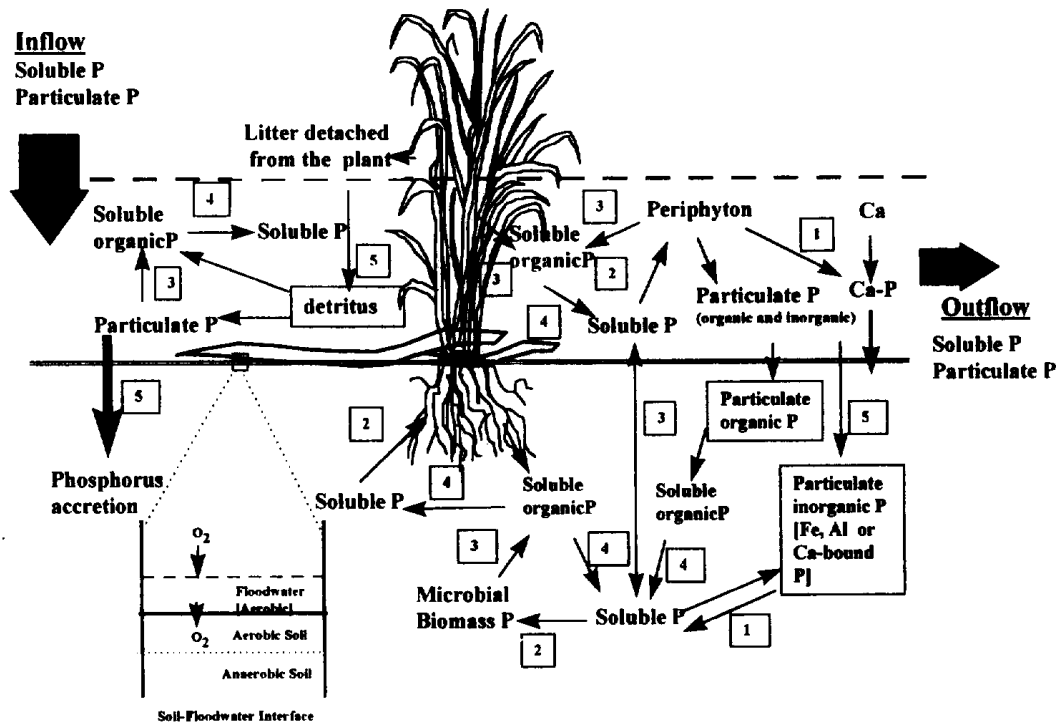
P fractions differ greatly in ecological significance. Total phosphorus (TP) represents all fractions of P including dissolved, particulate, organic and inorganic forms. Approximately 90 % of P in unpolluted freshwater systems occurs in unavailable organic forms (Wetzel, 1983; EPA, 2000). Orthophosphate ( $\text{PO}_4^{3-}$ ) is the only readily bioavailable fraction (*e.g.* Wetzel, 1983; Bostrom *et al.*, 1988; Correll, 1998) and the most ecologically significant owing to its' rapid utilization by algae and aquatic plants (*e.g.* Rigler, 1964; Schachtman *et al.*, 1998; Jarvie *et al.* 2006). The commonly sampled soluble reactive phosphate (SRP) fraction represents  $\text{PO}_4^{3-}$  and the fraction of organic P that can be rapidly converted to  $\text{PO}_4^{3-}$  by bacteria (Wetzel, 1983).

The major P removal pathways in constructed wetlands are sedimentation of particulate inorganic and organic P, co-precipitation and retention of  $\text{PO}_4^{3-}$  with Fe, Al and Ca cations to the sediments, macrophyte uptake, burial and peat accretion (*e.g.* Brix, 1994; Richardson, 1985; Vymazla, 2007) (Figure 12).

The removal of P to protect receiving aquatic ecosystems and meet water quality standards and guidelines is often a primary consideration in constructed wetland design (*e.g.* Cameron *et al.*, 2003; Maddison *et al.*, 2009; Vohla *et al.*, 2011); thus, the P removal efficiencies of constructed wetland designs under various hydraulic and nutrient loading rates has been extensively researched (*e.g.*; White *et al.*, 2000; Cameron *et al.*, 2003; Song *et al.*, 2006; Maddison, *et al.*, 2009; Slayton, 2009; reviewed by Vymazal, 2007; Vohla *et al.*, 2011). The amount of P in wastewater or stormwater influent entering constructed wetlands is often orders of magnitude higher than natural wetlands (Vymazal, 2007). Thus, to enable P in surface water to diffuse into constructed wetland sediments where P has already accumulated, surface water P must be high as observed by Patrick and Khalid (1974).

P removal efficiencies in constructed wetlands are variable. Vymazal (2007) found that the typical rate of total phosphorus (TP) removal by mature constructed wetlands ranged between 41 – 60 %. Newly constructed wetlands typically achieve higher P retention with removal rates of over 90 % being documented (*e.g.* Cameron *et al.*, 2003) owing to the greater availability of sediment P adsorption sites (Kadlec and Wallace, 2009). Slayton (2009) reported that the average TP removal rate by the 1190 ha Orlando Easterly Wetlands (OEW), a natural

wetland subjected to seasonal changes in hydraulic and nutrient loading rates, averaged 67.71 % over a 17 year period. The TP removal efficiency for 326 constructed wetlands of all types was reviewed by Vymazal (2007) where influent TP ranged from 3.8 – 10.5 mg P/L and averaged 6.8mg P/L. Mean effluent TP was 3.4mg P/L for a mean removal rate of 48 %. Recalling that the mean TP concentration for eutrophic lakes and reservoirs presented in Wetzel (1983) was 0.084 mg P/L and EPA guidelines define 0.096 mg P/L as an indication of hyereutrophic levels, the TP in both the influent and effluent of constructed wetlands is far above levels typical of even the most polluted of natural aquatic ecosystems. The EPA (2000) considered constructed wetlands to have low P retention capacity and unsuitable for P management to protect surface waters. Similarly, Richardson (1985) also observed low P removal capacity in various types of natural wetland systems relative to terrestrial soil and recommended that peat wetlands not be considered as effective P sinks for water quality management purposes.



**Figure 12. Phosphorus transformations in constructed wetlands.**

(1) adsorption and desorption, (2) plant and microbial uptake, (3) fragmentation and leaching, (4) mineralization and (5) sedimentation and burial.

*From Reddy and D'Angelo. 1997.*

Microbial populations can provide short-term P storage but are generally regarded as insignificant in P retention within constructed wetlands owing to their low biomass and rapid turnover rate (*e.g.*, Nichols, 1983; Vymazal, 2007). Similarly, average annual P retention by periphyton and phytoplankton, although potentially significant during the growing season, is also generally considered to be insignificant due to low biomass and high turnover rate. Microbial P absorption and assimilation may however be more significant in natural wetlands and constructed wetlands that receive low nutrient loading. Reddy *et al.* (1999) cited two studies where 60 % of P in the treatment wetlands was retained by microbial uptake and assimilation where P loading was low.

The only gaseous phases of P that are known to form in wetlands are phosphine (PH<sub>3</sub>) and diphosphine (P<sub>2</sub>H<sub>4</sub>) which are believed to form under highly reducing conditions (Kadlec and Wallace, 2009). The mechanisms responsible for phosphine formation in wetlands are unknown. The research reviewed in Kadlec and Wallace (2009) found that most phosphine remained trapped within wetland sediments and then at very low concentrations (<1 µg/L). However, at a sediment concentration of 2.2 g P/kg (very high for natural wetlands) a loss rate of 5.7 % of P (38 mg P/m<sup>2</sup>/yr) was explained by volatilization. The significance of P loss to volatilization remains unknown and to date has been ignored with respect to P removal by wetlands (Kadlec and Wallace, 2009).

### **1.13.1 Sedimentation of Particulate Phosphorus**

Suspended solids (SS) which includes all forms of inorganic and organic settleable, supracolloidal, colloidal and soluble solids ranging from 1µm to > 100 µm (EPA, 2000) is commonly the dominant source of total phosphorus (TP) in aquatic ecosystems (Ahl, 1988) and the primary form of P in agricultural runoff (*e.g.* Cooke and Williams, 1973, Jarvie *et al.*, 2006). Sediment P levels are typically highest in the influent outfall area for both natural and constructed wetlands where the majority of SS settle to the substrate (*e.g.* White *et al.*, 2000; Mayer *et al.*, 2006; Kadlec and Wallace, 2009). Constructed wetlands are consistently effective in reducing SS (Kadlec and Wallace, 2009) with nearly 100 % being removed if sufficient retention time is provided (Verhoeven and Meuleman, 1999; EPA, 2000; Cameron *et al.*, 2003). The physical removal of SS and particulate P by wetlands is enhanced by the attenuation of

inflow velocities by ponded water, coarse substrate material and wetland vegetation, each of which contributes to filtering, interception, aggregation and sedimentation by gravitational forces (EPA, 2000; Kadlec and Wallace, 2009). However, the 41 – 60 % P removal rate suggests that some of the particulate P is transformed into dissolved P which is transported via effluent out of the wetland.

### **1.13.2 Adsorption and Precipitation**

The voluminous body of literature on phosphate ( $\text{PO}_4^{3-}$ ) removal, cycling and storage in wetlands has identified coprecipitation with the cations, oxides and hydroxides of Al and Fe at neutral to acidic conditions as the most important factors in  $\text{PO}_4^{3-}$  removal (*e.g.*, Nichols, 1983; Reddy and D'Angelo, 1997). As most wetlands have a slightly acidic to neutral pH (Kadlec and Wallace, 2009),  $\text{PO}_4$  adsorption to Al and Fe are generally considered to be the primary factors in  $\text{PO}_4^{3-}$  retention in both natural and constructed wetlands. Richardson (1985) found that sediment Al concentration explained 87 % of the variation in P retention for fen, bog, swamp and marsh wetlands while multiple regression analysis showed little effect of Fe and other variables. Sakadevan and Bavor (1998) also found Al to be the most important determinant of P removal in constructed wetlands but that Fe also played a significant role. Battay *et al.* (2002) documented significant P retention on the surfaces of Fe, Mn and Al hydroxide plaques that formed on the surface of macrophyte roots. The significance of root plaque formation and P retention is unknown. Phosphate adsorption to Ca and Mn cations and positively charged silica and clay cation exchange sites are the primary P removal and retention mechanisms under alkaline conditions (*e.g.* Nichols, 1983; Danen-Louwerse *et al.*, 1993; Brix, 1997; Reddy and D'Angelo, 1997; Reddy *et al.*, 1999; Gray *et al.*, 2000; Vymazal, 2007; Vohla *et al.*, 2011).

### **1.13.3 The Influence of Reduction-Oxidation Potential in Phosphorus Removal and Retention**

Sediment reduction and oxidation (redox) potentials are considered to be the primary determining factors in P retention in constructed and natural wetlands. In aerobic surface waters the movement of P is generally unidirectional into the sediments (Wetzel, 1983). Under anaerobic conditions the reduction of  $\text{Fe}^{3+}$  and  $\text{Mn}^{4+}$  is believed to result in the release of Fe and

Mn bound P (Patrick and Khalid, 1974). A thin oxic layer at the sediment/water interface of usually no more than 1 cm in thickness can prevent or inhibit P release into the water column (Wetzel, 1983). Aerobic conditions within the rhizosphere can also increase P retention within the sediments through P co-precipitation with Fe, Al, Mn, Ca, Mg, clay and colloids in pore water (e.g. Wetzel, 1983; Danen-Louwerse, 1993; Van de Moortel *et al.*, 2009, Yates and Prasher, 2009).

Differences in root oxygen escape between macrophyte species may be a significant factor in P retention whereby greater oxidizing conditions within the sediments would increase Fe availability therefore P retention capacity. Szögi *et al.* (2004) observed considerably higher oxidizing conditions over a 12 month period in a wetland planted with *Schoenoplectus* compared to a *Typha* wetland receiving the same effluent. Neil and Graham (1989) found *Schoenoplectus tabernaemontani* to be more effective at P removal than *Typha angustifolia* which may have been due to higher oxidizing conditions within the *S. tabernaemontani* rhizosphere.

#### **1.13.4 Phosphorus Assimilation by Macrophytes**

Davis (1982) cited in Reddy *et al.*, (1999) observed that only 2 – 4 % of P<sup>32</sup> added to surface water was eventually absorbed by *Typha domingensis*. Submergent macrophytes have been shown to obtain some P from surface water under eutrophic conditions; however, the majority of P is still obtained from the sediments (Bole and Allan, 1978; Carignan and Kalff, 1980). During the growing season, direct plant uptake in constructed wetlands can remove a significant amount of P, but the proportion of P removed under typically high P loading rates is generally considered small to minor compared to sedimentation, precipitation and burial within the sediments (Brix and Schierup, 1989; Brix, 1997; Vymazal, 2007; Mustafa & Miklas Scholz, 2011). The proportion of surface and sediment water P retained within wetland vegetation following senescence and death is generally minor (Brix, 1997; Vymazal, 2007). Mustafa *et al.* (2011) reported that < 1 % of the P loading was retained by *T. latifolia* over a 12 month period.

After death, approximately 35 to 75 % of the P assimilated within plant tissue is re-released during decomposition in natural wetlands (Nichols, 1983). The seasonal effect of P retention by macrophytes is well illustrated in the review by Nichols (1983) where a *Typha* stand

subjected to storm sewer water retained 83 % of influent P during the summer, only 1 % in the fall and 8 % in the spring giving a 10 % removal rate overall.

### 1.13.5 Phosphorus Retention by Peat Accretion

Peat accretion is generally accepted as the only long-term P retention mechanism in constructed wetlands (*e.g.* Nichols, 1983; Reddy and D'Angelo, 1997; Vymazal, 2007; Mustafa *et al.*, 2011). Acidic conditions in the deeper anoxic sediments as a result of organic acid production by anaerobic bacteria reduce the decomposition rate of organic matter potentially resulting in the permanent burial of organically-bound P (Smolders *et al.*, 2006). Conversely, bicarbonate alkalinity produced within the rhizosphere through macrophyte metabolism can buffer against organic acid production associated with microbial decomposition and reduce acidity and thus increase microbial biomass and the mineralization of organic P into bioavailable P (Smolders *et al.*, 2006).

Natural wetland peat generally contains very low amounts of P levels with P commonly representing < 0.1 % of peat biomass (Nichols, 1983). Peat accretion data presented in Richardson (1985) indicated a range of 0.005 to 0.24g P/m<sup>2</sup>·yr in natural wetlands. Peat in constructed wetlands contains much higher levels of P due to much higher P loading rates (Vymazal, 2007). 1.6g P/ m<sup>2</sup>·yr has been reported under nutrient-rich conditions for a *Phragmites* stand and 1.95g P m<sup>2</sup>·yr has been reported for a *Schoenoplectus fluviatilis* stand (Kadlec and Wallace, 2009). The amount of P removed in peat however is still very low relative to the total P loading rate over a 12 month period. Macrophyte Harvesting for Phosphorus Removal

The harvesting of above-ground macrophyte biomass has been suggested as a means of reducing nutrients in eutrophicated systems by several authors (*e.g.* Boyd, 1970, Yount and Crossman; 1970, Brix and Schrierup, 1989; Vyamzal, 2007; Maddison *et al.*, 2009). However, P yields by macrophyte harvesting is generally low as is the proportion of P removed relative to loading rate (Vyamzal, 2007), is costly and requires specialized equipment (Kadlec and Wallace, 2009: 363). Vymazal (2004, 2007) suggested that under relatively low loading rates for constructed wetlands of < 20 g P/m<sup>2</sup>/y, macrophyte harvesting could potentially be effective in



reducing P in eutrophicated systems. Selective harvesting for P recovery could also provide a means of maintaining optimal macrophyte density and diversity as well as a means of removing unwanted invasive species.

*Typha spp.* is currently being harvested where the Red River empties into Lake Winnipeg after which it is made into biofuel pellets (Austin, 2011). The high nutrient-containing ash produced from the pellet burning is being used as a crop fertilizer. In Estonia, *Typha* chips and inflorescences are used in cost-efficient building blocks and clay plaster respectively, with the former also increasing the insulation value of the blocks (Maddison *et al.*, 2005). Organic matter and vegetation from the constructed wetland described in Maine *et al.* (2009) was used as fertilizer and compost for ornamental plants. It is recommended that harvested vegetation be tested for toxic pollutants if the harvested material is to be used for further uses such as green fertilizer, biofuels or building materials where further release of toxic metals and chemicals is possible.

#### **1.14 Phosphorus Removal Limitation in Constructed Wetlands**

There is a finite P removal capacity in constructed wetlands and natural wetlands used for wastewater treatment as on-going P loading reduces or saturates the availability of Fe, Al, Ca and other P sediment adsorption sites over time (EPA, 2000; White *et al.*, 2000; Vymazal, 2007; Kadlec and Wallace, 2009). Constructed wetlands may also lose P removal capacity during winter ice-over when the diffusion of atmospheric oxygen into the surface water ceases. The biological oxygen demand (BOD) associated with the decomposition of vegetation and other forms of organic matter may consume all available oxygen during ice-over periods resulting in anoxic conditions throughout the sediments and water column resulting in the mobilization of P into surface water (Wetzel, 1983). Retention within the sediments is also partially determined by pH where high pH and low pH can result in the solubilization of Al and Fe bound P (Wetzel, 1983). There may therefore be a trade-off between maximizing peat accretion by increasing acidity and the loss of Al and Fe binding sites to Al and Fe solubilization. Artificial Substrates for Phosphorus Removal

There has been a considerable body of work in assessing the P retention capabilities of various types of substrates ranging from natural material (*e.g.* apatite, sand, gravel, peat), industrial by-products (blast furnace slag, fly ash, iron ore) and manmade material (*e.g.* Alunite, Filtralite™, oyster shell) for use in constructed wetlands to improve P removal capability (*e.g.* Sakadevan and Bavor, 1998; Drizzo *et al.*, 1999; reviewed by Vohla *et al.*, 2011). A very high P retention was often achieved (> 90 %) for several types of natural and artificial substrates (*e.g.* Cameron *et al.*, 2003; Vohla *et al.*, 2011). Although most of the data has been collected from small-scale experiments with data available from only a few fully operating constructed wetlands, results appear promising and an overall improvement in P removal may be realized such that the use of constructed wetlands for ecosystem protection and restoration may become more feasible. There is also interest in assessing the feasibility of recycling the adsorbed P for use as fertilizer (Vohla, *et al.*, 2011).

### **1.15 Predominant use of Invasive Vegetation in Constructed Wetlands**

*Typha spp* and *Phragmites australis* are by far the most common species of emergent vegetation used in constructed wetlands both in North America and Europe (Vymazal, 2007; Kadlec and Wallace, 2009; Maddison *et al.*, 2009). Brisson and Chazarenc, (2009) and Marchand *et al.*, (2010) questioned the general dependency on these species in the context of pollutant removal efficiency and recommended that further research be conducted to identify plants that may be more effective for removing specific types of heavy metals and nutrients.

*T. latifolia* is probably the most commonly used species used in the constructed wetland research (*e.g.* Martín, and Fernández, 1992; Cameron *et al.*, 2003; Weng *et al.*, 2006; Maddison *et al.*, 2009; Calheiros *et al.*, 2009; Yalcuk *et al.*, 2010; Mustafa and Scholz, 2011). *Phragmites australis australis* (common reed) is the most commonly used macrophyte species used in European constructed wetlands (Brix and Schierup, 1989). While there has been a steadily increasing volume of literature documenting the invasive nature of *Typha spp* and *Phragmites australis australis* in North America spanning several decades (*e.g.* Linde *et al.*, 1976; Beule, 1979; Grace and Harrison, 1986; Galatowitsch *et al.*, 1999; Mal and Narine. 2004; Shih and Finkelstein, 2008), the common use of invasive species such as *T. latifolia* in constructed

wetlands and constructed wetlands research has gone almost unquestioned in respect to potential environmental impact with the exception of Livingston (1989) and Taylor (1992).

Based on the lack of evidence to justify the wide-spread use of *T. latifolia* and other invasive species in constructed wetlands where organic matter and nutrient removal are the primary objectives, vegetation selected for constructed wetlands should be restricted to locally adapted native non-invasive species, particularly in areas where invasive species have not yet had significant impact. Numerous native species used in constructed wetlands have proven to be effective at pollutant removal. White *et al.* (2000) determined that a 1246 ha restored natural prairie wetland dominated by native vegetation (*Schoenoplectus acutus*, *Stuckenia pectinata* L. (formerly *Potamogeton pectinatus*), *Myriophyllum albescens*, *Potamogeton richardsonii*) removed 60 % of P received from a beef slaughter house and municipal sewage effluent over a five-year period. *S. pectinata* is native to most of North America and can be invasive in eutrophic conditions but is also an important food plant for wildlife (Casey, 2010).

## 1.16 Summary

The shift from diverse wetland ecosystem communities to monotypic stands of *Typha latifolia* and other *Typha* species have often been attributed, in whole or in part, to cultural eutrophication (*e.g.* Moore *et al.*, 1989; McJannet *et al.*, 1995; Galatowitsch *et al.* 1999; Shih and Finkelstein, 2008). However, data to support this hypothesis is lacking for *T. latifolia* and other *Typha* species occurring in the Northern US and Canada. Although several previous field studies on *Typha* invasions have found evidence to support the eutrophication hypothesis, the effects of nutrient loading have been difficult to elucidate due to other confounding variables, particularly altered hydroperiod. Additionally, controlled and semi-controlled experiments conducted under artificial light (Wetzel and van der Valk, 1998), in greenhouse (Woo and Zedler, 2002) and in outdoor settings (Shiple and Keddy, 1988; Keddy *et al.*, 2000) to determine *Typha spp.* response to N and P treatments have involved loading rates that in general far exceed levels observed in even the most polluted of natural wetlands.

The need to determine the effects of N and P loading at environmentally relevant levels is timely given that there is significant interest in building constructed wetlands for the purpose of

removing nutrients from non-point sources at the watershed scale. Constructed wetlands technology has continued to emphasize the use of *Typha*, *Phragmites* and *Scirpus spp.* (bulrushes) including *Schoenoplectus spp.* (Kadlec and Wallace, 2009) in North America, Europe and Asia with the former two species which are commonly known to be invasive. Furthermore, constructed wetlands have a limited capacity to remove both N and P. The widespread use of constructed wetlands could therefore introduce invasive species to unimpacted areas and, by releasing high levels of N and P in effluent, create suitable conditions for invasion by *T. latifolia* and possibly other invasive species. The effectiveness of constructed wetlands for removing N and P to prevent impact and cultural eutrophication appears highly limited and improved nutrient management should be emphasized.

There is currently effort underway to genetically engineer *T. latifolia* for use in constructed wetlands (Kadlec and Wallace, 2009). These genetically engineered varieties may introduce foreign genetic material into wild populations and potentially increase their invasiveness or have a negative impact on the survival of wild *T. latifolia* populations under certain conditions. There may also be risks to the wildlife that consume genetically modified varieties.

The lack of evidence demonstrating that *T. latifolia* and other invasive species commonly used in constructed wetlands are superior to native non-invasive species suggests that non-invasive species can be equally as effective in removing N and P without posing a potentially significant environmental risk. The use of native non-invasive species in constructed wetlands, particularly within or adjacent to ecosystems not significantly impacted by *Typha latifolia* and other invasive species is a preventative measure that must be considered.

The overall objective of this thesis was to test the eutrophication hypothesis by determining the response of *T. latifolia* to N and P concentrations representative of unimpacted (oligotrophic) wetlands and to N and P concentrations representative of eutrophicated natural ecosystems under semi-controlled laboratory conditions. Sub-objective 1 was to develop a protocol for rearing *T. latifolia* from seed to adult in order to provide specimens of known history uniform age and equal fitness for effective hypothesis testing. Sub-objective 2 was to determine the feasibility of conducting future experiments with the non-invasive species

*Schoenoplectus acutus* and *Schoenoplectus tabernaemontani* under semi-controlled laboratory conditions for future experiments that could compare nutrient removal capability and competitive ability against *T. latifolia* under various nutrient regimes. Sub-objective 3 was to raise the issue of invasive species use in constructed wetlands and encourage research into the feasibility of utilizing native non-invasive species.

## 2 MATERIALS AND METHODS

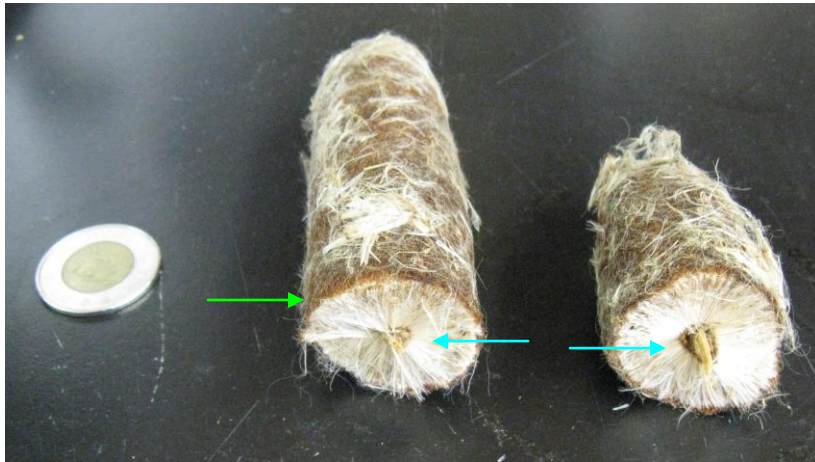
### 2.1 Modified MOE Cleaning Protocol

The procedure for cleaning all glassware and experimental unit vessels followed a modified Ontario Ministry of Environment (MOE) protocol, hence forth known as the Ryerson Protocol offered by K. Puddephatt, as preparation for all trials and experiments to prevent introducing contaminants as a potential source of experimental error, and was as follows:

1. All vessels, seed trays and laboratory equipment including glass and plastic pipettes, volumetric flasks, Petri dishes, graduated cylinders and glass beakers were carefully inspected for cracks, visible remnants of soil, algae or reagent residue and if necessary rinsed with municipal drinking water (MDW).
2. Prior to initial use at the beginning of each experiment all equipment was soaked for at least 15 minutes in Extran<sup>®</sup> (phosphate-free laboratory glassware washing detergent with nitrilotriacetic acid trisodium salt, NaOH and Brij 35<sup>®</sup> as the active ingredients) to remove all traces of organic contaminants.
3. All seed trays, Petri dishes and vessels used in *T. latifolia* experiments involving N and P treatments were then finger scrubbed with MDW to removal all traces of Extran. Volumetric flasks and graduated cylinders were thoroughly flushed with running MDW.
4. All equipment was then thoroughly rinsed in 10% HCl to remove all traces of residue, Extran, monochloramine (NH<sub>2</sub>Cl) and/or dichloramine (NHCl<sub>2</sub>) and their organic compound complexes RNH<sub>2</sub>Cl and RNHCl<sub>2</sub>, metals and bases.
5. All items were rinsed three times with distilled water, inverted and allowed to air dry. Any remaining distilled water in volumetric flasks was rinsed with the solvent to be used for reagents used in colourimetric water quality analysis.
6. Following the initial cleaning procedure described above, glassware used for inorganic nutrient analysis or for making *T. latifolia* grow media was subjected to each of the above steps except for soaking in Extran (step 2).

## 2.2 Germination Protocol

A major component of this study was the development of a whole lifecycle protocol for *Typha latifolia* as the existing literature was very depauperate in this aspect. Thus, germination of seeds is the first crucial step. Seeds are removed from the inflorescence by pinching off a small grab of plumage (Figure 13) using a forceps which typically yielded over 300 seeds.



**Figure 13.** An internal view of the *Typha latifolia* inflorescence used for all germination trials and experiments illustrating the plumage (blue arrows) and seeds (green arrow).

*Photo by Mark Tiley*

Seeds were then placed into the designated vessel or Petri dish containing MDW with sufficient depth to effectively mix and separate individual seeds. Randomly selected seeds intended for one experimental unit vessel were then transferred to a second Petri dish filled to approximately 1.0cm using a dissecting needle to which the seeds would stick when wet. The purpose of the transfer of seeds to a second Petri dish was to ensure that the correct number of seeds was transferred to an awaiting experimental unit given that seeds can be difficult to accurately count due to clumping with carpodium and flower parts. Seeds were then individually transferred using the dissecting needle to an awaiting vessel or experimental unit. The process of randomly selecting and transferring seeds was repeated until all experimental units contained the desired number of seeds.

### 2.2.1 Source of Seeds

A single mature *Typha latifolia* inflorescence (flower) Figure 14, collected during the fall, 2010, in the Walsingham, Ontario region (Catherine Riley-Arenburg, Acorus Restoration office manager, personal communication) was received by Acorus Restoration and used in all propagation trials and experiments. Acorus Restoration also provided *Schoenoplectus acutus* and *Schoenoplectus tabernaemontani* seeds to test the hypothesis that P removal by a native would be as effective as *T. latifolia*. Trials to germinate *Schoenoplectus* seed under artificial light, natural light, on top soil and paper towel substrate and outdoors using seeds exposed to outdoor overwintering conditions were all unsuccessful and will not be detailed here.



**Figure 14. The ripe *Typha latifolia* inflorescence used for all germination and rearing trials and experiments**

### 2.2.2 Germination and Experimental Vessels

Food-grade Snaptite<sup>®</sup> 3.1 liter (L) polypropylene containers (Figure 15) were used as experimental unit vessels for the *T. latifolia* age 0 to 85 day (d) old seedling response to as 40 units could fit under a bench light bank 242 cm in length, 67.4 cm in width and 51.2 cm to bulb height, henceforth referred to as light bank 1 (LB1), with ten replicates/treatment. The 3.1 L vessels were also considered suitable for the subsequent *T. latifolia* response to N and P experiment as preliminary trials indicated that the interior dimensions of the 3.1 L vessel (16.4 cm in height, 14.0 cm wide and 20.0 cm in length) were easily handled when containing wet sediment and approximately 1L of surface water without spilling. The interior dimensions of the



bottom of the container were 10.0 cm in width and 16.4 cm in length. The clear plastic had a negligible effect on photosynthetically active radiation (PAR) with a decrease in 3 microeinsteins per meter squared per second ( $\mu\text{E m}^{-2} \text{s}^{-1}$ ) observed by holding the Field Scout light meter under a light bank and moving the vessel over the light meter sensor.



**Figure 15. Snaplute 3.1 L polypropylene vessel.**

Polypropylene is chemically resistant to organic solvents, acids and bases (Ipex, 2001). Information on the chemical resistance of polypropylene to acids, bases and gases that may be naturally produced in microcosms (*e.g.* nitric acid ( $\text{HNO}_3$ ); hydrogen sulfide ( $\text{H}_2\text{S}$ ); ammonia ( $\text{NH}_3$ ) phosphoric acid ( $\text{H}_3\text{PO}_4$ ), in addition to other compounds, was compiled by Ipex Inc., (2001), Borealis (2001) and Engineering Toolbox (2013). A summary of the compounds assessed by Engineering Toolbox (2013) for chemical reactivity between polypropylene and compounds that may occur within microcosm wetlands is provided in Appendix F. Microcosms were defined by Fraser and Keddy (1997) as small ecosystems. In this experiment, each microcosm consisted of a simplified ecosystem consisting of sediment, water, microflora and microfauna introduced via the sediment, air and water, and the *T. latifolia* plants.

All labels were removed from newly purchased containers and both containers and lids were washed as per the glassware cleaning protocol before use. The entire bottom 6 cm of each vessel was covered in black duct tape (Figure 16) to simulate natural conditions by preventing direct light from contacting subsurface sediments on the sides of the container. The addition of 1.205 ml of water measured using a graduated plastic water jug indicated the 6 cm mark below which

black duct tape was applied. All vessels were numbered and randomly assigned to one of the four treatments.



**Figure 16.** Microcosm vessels with black duct tape covering the lower 6 cm of each vessel and identifying labels after the removal of *Typha latifolia* shoots.

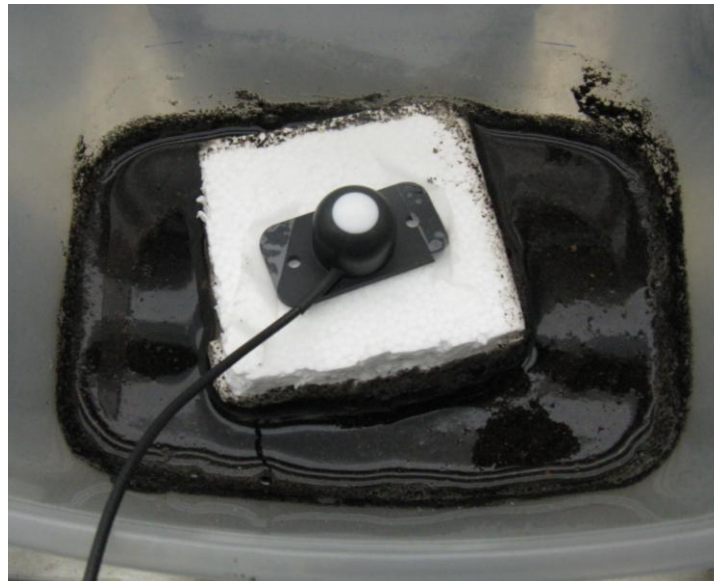
### 2.2.3 Germination Lighting

Light bank 2 (LB2) which shared the same bench as LB1 was used to confirm that viable seeds were correctly being distinguished from carpodium based on percent germination where carpodium should have 0 percent germination. As for LB1, LB2 was fitted with Dura-Test T8 Vita-Lite bulbs, two of which had been removed for the purpose of providing the correct lighting for developing algal cultures. After extensive research on commercially available light bulbs used for plant propagation, Puddephatt (2013) identified the Dura-Test T8 Vita-Lite bulb as most closely resembling natural solar irradiation and it had also previously been used by the USEPA (Lewis *et al.*, 1994) and Environment Canada (2007). Details of the Dura-Test T8 Vita-Lite bulb are provided in Puddephatt (2013). PAR was measured directly over the middle of each Petri dish using a Spectrum Technologies Field Scout Type 2 (cable attached sensor), light meter taped to 30 cm ruler.

### 2.2.4 Seedling Response to Phosphorus

All sixteen bulbs installed into LB1 at the start of the seedling response to P were operating. Light intensity was measured at the center of each vessel with the Field Scout Type 2 light meter

taped to one of four 2.54 cm (1 in) thick pieces of styrofoam, cleaned according to Ryerson cleaning protocol, after the addition of the 650 ml of phosphorus treatment to document light conditions before the addition of seeds and the onset of germination. Each piece of Styrofoam was allocated to a specific treatment to avoid cross contamination between treatments. Light measurements were collected approximately three hours before seeds were added on May 25, 2012 (day 0).



**Figure 17. Light sensor attached to styrofoam to determine light intensity immediately prior to seeds being added.**

### **2.2.5 Percent Germination of Viable Seeds.**

To ensure that all seeds added to all seedling P treatment replicates were potentially viable and percent germination data did not incorporate error caused by the addition of carpodium (nonviable seed), an investigation was run in parallel with the percent germination experiment (section 2.2.4) to confirm that all seeds added to reference and P treatments were potentially viable. If identified correctly, carpodium would have 0 percent germination. Approximately ten randomly selected seeds were then transferred to one of twelve awaiting Petri dishes containing standing MDW following the seed germination protocol (section 2.2). Twelve Petri dishes were placed under a second bench light bank (hence forth referred to as light bank 2 (LB2)). As LB2 is located within 2m of LB1, air temperature and relative humidity conditions monitored under

LB1 were considered representative of LB2 temperature and RH conditions. Petri dish locations were randomized on the first day only as the experiment was conducted over four days.

### **2.2.6 Percent Germination of Seeds added to Reference and P treatment Microcosms**

Three randomly selected *T. latifolia* seeds were added to each reference and phosphorus treatment microcosm (3 x 40 for a total of 120 seeds) on day 0 using the procedure following the germination protocol (section 2.2). Percent germination was monitored from day 2 to day 6.

## **2.3 Seedling Response to Phosphorus.**

To test the hypothesis that increased P concentrations will result in increased *T. latifolia* post germination seedling survival and biomass at concentrations representative of P concentrations observed in surface water and agricultural runoff, four treatments of 0, 50, 100 and 300  $\mu\text{g PO}_4\text{-P/L}$  were selected based on surface water P levels spanning the range of trophic levels defined by Wetzel (1983) with the 0 (no P added) treatment serving as a reference and to simulate oligotrophic (nutrient poor) sediments. A treatment of 300  $\mu\text{g PO}_4\text{-P/L}$  was selected as P in agricultural runoff can exceed 300  $\mu\text{g PO}_4\text{-P/L}$  (Yates and Prasher, 2009). The 3.1 L polypropylene vessels described in section 2.2.2 were used to construct microcosms.

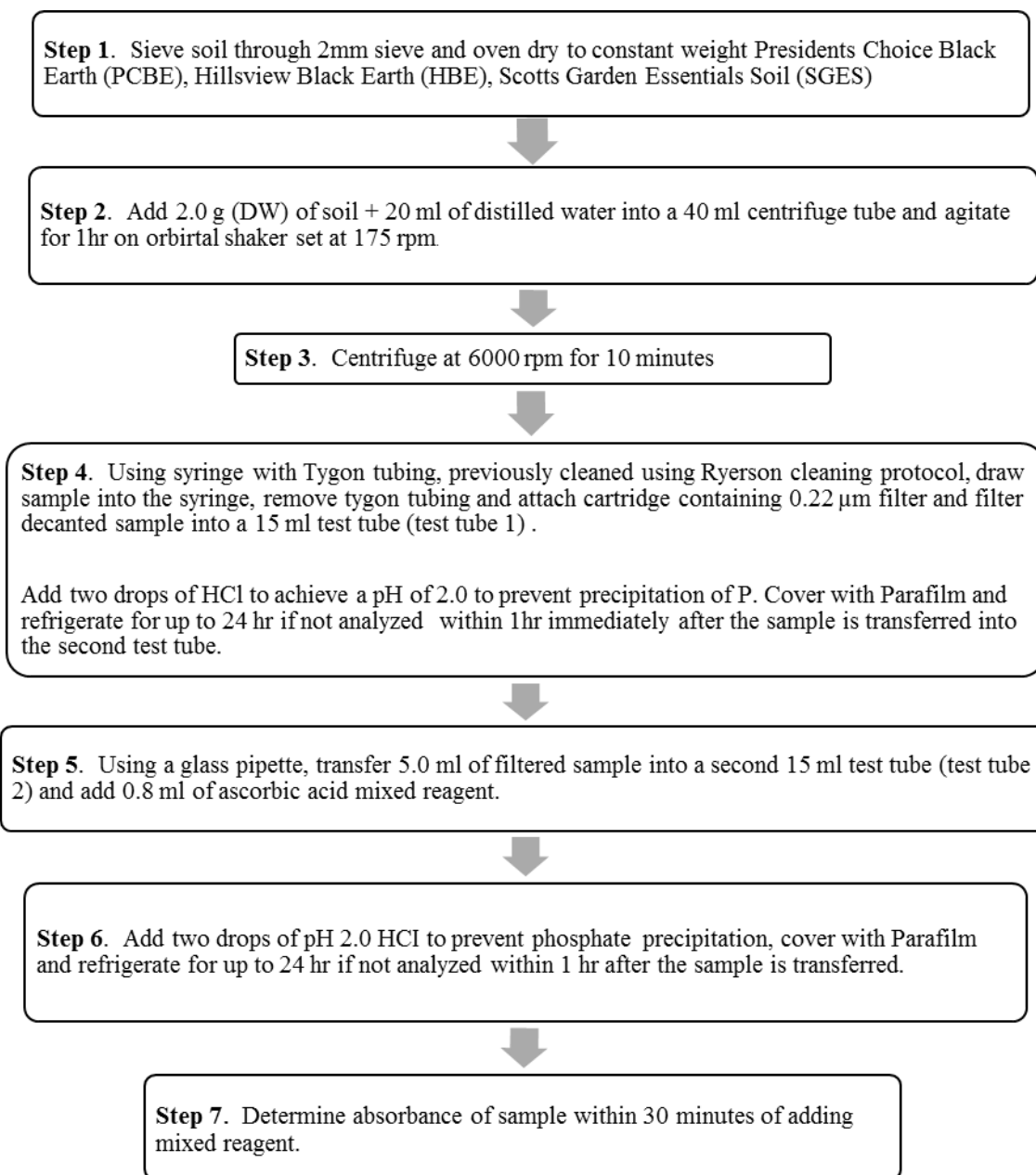
### **2.3.1 Formulated Sediment**

It was extremely crucial that the substrate used to conduct experiments testing the effects of P loading on *T. latifolia* growth, survival and biomass be low in P so as not to skew experimental results. Thus, an organic substrate with oligotrophic levels of P was crucial. Two of three commercially-available soils, including Presidents Choice Black Earth topsoil (PCTS) used in MOE experiments (Puddephatt, 2013) were considered but did not provide details on P availability. Thus, PCTS and Hillview Black Earth (HBE) topsoils were tested for P using the Deionized Water Extractible P Method (DWEPM) (Self-Davis et al., 2009) and the Murphy and Riley (1962) ascorbic acid method. The DWEPM procedure is summarized in Figure 18 and described below.

### 2.3.2 Deionized Water Extractible Phosphate Method

The DWEPM described in Self-Davis *et al.* (2009) was modified by using a VWR orbital shaker instead of the specified reciprocating or end-to-end shaker. The orbital shaker was set at 175 rpm which was sufficient to agitate the prescribed 2.0 g of sediment for one hour, oven-dried to constant weight, to 20 ml of deionized water placed into a 40 ml centrifuge tube. Instead of the indicated 0.45 $\mu$ m filter, a 0.22  $\mu$ m polypropylene filter was used to improve data accuracy of the phosphate analysis by removing greater amounts of microbes and colloids (Haygarth *et al.*, 2009). Scotts Garden Essentials Soil (SGES), known to be high in P, was used as a reference to validate the modified methodology.

Three randomly selected 10 g samples of PCTS and SGES were dried to constant weight at 60°C and sieved through a 2 mm mesh screen. The HBE was dry weighted but not sieved. To decant the water extracted P from the sediment, samples were centrifuged for 10 minutes at 6000 rpm. 15 ml of the decanted portion of the sample was drawn by syringe with a piece of Tygon tubing attached. The Tygon tubing was then removed and a filter cartridge containing a 0.22  $\mu$ m filter paper previously flushed with deionized water was attached to the syringe. The sample was then filtered into a second test tube and acidified to pH 2.0 with two drops of concentrated HCl to preserve the sample and prevent P precipitation. A small piece of Parafilm was placed over the test tube mouth to prevent evaporation or entry of airborne contaminants until all decanted samples had been filtered. 5.0 ml of filtered samples were then transferred to a second test tube to which 0.8 ml of ascorbic acid mixed reagent was added and absorbance determined within 30 minutes. The modified ascorbic acid procedure as described below was then followed to obtain phosphate concentrations for each brand of topsoil.



**Figure 18. Flow chart summarizing the Deionized Water Extractable Phosphate Method.**

### **Ascorbic Acid Method**

A modified Murphy and Riley (1962) method, described as the ascorbic acid method in APHA (1998), was used to determine deionized water extractable phosphate and pore water phosphate. The ascorbic acid reagent is comprised of sulfuric acid, ammonium molybdate, potassium antimonyl tartate and ascorbic acid. A mixed reagent containing sulfuric acid,

ammonium molybdate and antimonyl tartate was made according to APHA (1998) and stored in a glass amber jug (reagent A). When all water sampling had been completed a mixed reagent was made by adding 0.132 g of ascorbic acid into a 25.0 ml volumetric flask to which reagent A was added to the 25.0 ml mark to make mixed reagent B. The ascorbic acid reduces phosphomolybdic acid and produces a blue colour proportional to the amount of phosphate in the sample. Mixed reagent B remains stable for 24 hour (hr) (APHA, 1998). The method was modified by reducing sample volume and, proportionally, reagent weights and solvent volumes by a factor of 10 to reduce reagent consumption and waste such that 0.8 ml of mixed reagent was added to 5.0 ml of filtered sample water instead of 8.0ml of reagent being added to 50.0 ml of sample water. To improve sample accuracy a 25 mm diameter 0.22 micrometers ( $\mu\text{m}$ ) polypropylene filter was used instead of the standard 0.45  $\mu\text{m}$  filter paper to increase the removal of colloids, cells and microbes  $< 0.45 \mu\text{m}$  which are a source of P. All other procedures were followed according to APHA (1998).

### **Cadmium Reduction Method**

For the seedling response to P experiment, it was desirable to have levels of N that were non-limiting. To determine whether bioavailable N levels in Presidents Choice Black Earth topsoil (PCTS) may be a limiting factor to *T. latifolia* seedling growth and survival, nitrate ( $\text{NO}_3\text{-N} + \text{NO}_2\text{-N}$ ) levels in the surface water of a 3.1 L microcosm containing flooded PCTS were determined by analyzing one sample collected by a 60 ml syringe using the standard cadmium reduction method described in APHA (1998). The method was modified by reducing the sample volume and, proportionally, reagent weights and volumes by a factor of 10 to reduce Cd containing wastewater and reagent consumption. The City of Toronto drinking water summary reports indicated nitrate levels averaging 0.41 – 0.46 mg  $\text{NO}_3\text{-N} + \text{NO}_2\text{-N}$  /L from 2006 to 2012 (City of Toronto 2013b) and considered biologically significant (Dr. A. Laursen, Ryerson University, personal communication). The modified cadmium reduction method was therefore used to determine whether laboratory MDW nitrate levels were similar to the concentrations reported in City of Toronto (2013b) in order to estimate the total N added to grow media and the N: P ratios for each treatment. The details of the Cd reduction method are as follows:

## **Background**

The cadmium (Cd) reduction method is recommended for samples with concentrations  $\leq$  1.0 mg/L (PPM) (APHA, 1998).  $\text{NO}_3^-$  is reduced to  $\text{NO}_2^-$  in the presence of Cd. The latter reacts with colour reagent and is therefore quantifiable using colourimetric techniques.

## **Buffer Solution**

- Dissolve 13.0 g  $\text{NH}_4\text{Cl}$  and 1.7 g Ammonium chloride disodium ethylenediamine tetraacetate (EDTA) in 900 ml Millipore water using a clean 1000 ml beaker.  $\text{NO}_3^-$  is reduced to  $\text{NO}_2^-$  in the presence of Cd, the latter reacting with colour reagent and therefore quantifiable with colourimetric technique.
- Adjust to pH 8.5 with  $\text{NH}_4\text{OH}$ .
- Dilute to 1.0 L in a clean 1000 ml volumetric flask

## **Colour Reagent**

- To 800 ml millipore water add 100 ml 85 % phosphoric acid and 10.0 g sulfanilamide.
- After dissolving sulfanilamide completely, add 1.0 g N-(1-naphyl)-ethylenediamine dihydrochloride. Mix to dissolve.
- Dilute to 1.0 L.
- According to the APHA (1998) colour reagent remains chemically stable for approx. 1 month under refrigeration in a dark bottle.

## **$\text{NO}_3\text{-N}$ Standard Curve**

- Use potassium nitrate ( $\text{KNO}_3$ ) or sodium nitrite ( $\text{NaNO}_2$ ) for standard curves as per APHA (1998) protocol.
- Dissolve 0.7218 g  $\text{KNO}_3$  ( $= 0.007139\text{mol KNO}_3 = 0.1\text{g N}$ ) dried at 80-84°C to constant weight (APHA recommends 60°C) for 24 hr. Dilute to 1000 ml and invert to mix to make 100 mg  $\text{NO}_3\text{-N/L}$ .



- Stock can be stored in fridge for at least 6 months with 2 ml  $\text{CHCl}_3/\text{L}$ .
- Make 10 mg  $\text{NO}_3\text{-N}$  intermediate  $\text{NO}_3\text{-N}$  solution by diluting 100 ml of 100 mg  $\text{NO}_3\text{-N}/\text{L}$  stock solution with 900 ml Millipore water (1:10) using a 1.0 L volumetric flask to obtain 10 mg  $\text{NO}_3\text{-N}/\text{L}$ .
- Create the following standards: 0.05, 0.1, 0.2, 0.5, 0.8, and 1.0 mg  $\text{NO}_3\text{-N}/\text{L}$  by diluting the following intermediate solution to 100ml using 100 ml volumetric flasks: 0.5, 1.0, 2.0, 5.0, 8.0, 10.0 ml.

### **$\text{NO}_2\text{-N}$ Standards**

- $\text{NO}_2^-$  standards are necessary to ensure that the column is reducing  $\text{NO}_3^-$  efficiently.
- To make a 10.0 mg  $\text{NO}_2^- \text{N}/\text{L}$  stock, 0.0493 g  $\text{NaNO}_2$  to 1.0 L millipore water. Preserve with 1 ml  $\text{CHCl}_3$  (chloroform).
- Or, following APHA, add 1.232 g  $\text{NaNO}_2$  to 1.0 L Millipore water for 250 mg  $\text{NaNO}_3\text{-N}/\text{L}$  or 250  $\mu\text{g NaNO}_2\text{-N}/\text{ml}$ . Preserve with 1 ml  $\text{CHCl}_3$  (chloroform).
- Pass same concentration of  $\text{NO}_2\text{-N}$  and  $\text{NO}_3\text{-N}$  standards through the column to determine the reducing efficiency of the column and to confirm colour reagent and buffer is effective.

2.5 ml of water sample is added to 7.5 ml of buffer in a 15 ml test tube and passed through a Cd reducing column using a pump. 0.4 ml of colour reagent is then added to the collected sample, reduced to  $\text{NO}_2^-$ , to obtain a colour reaction, the intensity of which is proportional to the amount of  $\text{NO}_2^-$  in the sample, and analyzed to determine absorbance.

### **Colourimetric Sample Analysis**

A Perkin Elmer UV/VIS Lamda 20 spectrometer was used to analyze all phosphate and nitrate + nitrite samples. The spectrometer wavelength was set at 880 nm and 543 nm as specified in APHA (1998) for phosphate and nitrate analysis respectively. Six standards of known concentration and two Millipore water blanks were used to generate standard curves. In all cases, sample concentrations were within the concentration range of the standard curve except

for deionized water extracted P from SGES. The R-squared values of the standard curves always exceeded 0.99. At least one P and N standard was analyzed to ensure data accuracy for each sample run. The same cuvette was used for all sample water analysis. To prevent sample cross contamination, the cuvette was rinsed with distilled water then rinsed with 10 % HCl followed by three rinses with distilled water. The cuvette was inverted and shaken to remove nearly all of the distilled water. The cuvette exterior was thoroughly wiped with Kimwipe paper before each sample analysis. The order of sample analysis was from lowest to highest colour intensity which was equivalent to analyzing from low concentration to high concentration.

### **2.3.3 Formulated Sediment Top Soil Selection**

The phosphate levels in PCTS were very low at 0.03  $\mu\text{g PO}_4\text{-P/L}$  which was less than the 10.0  $\mu\text{g PO}_4\text{-P}$  below which the accuracy of the ascorbic acid method is unreliable (APHA, 1998). The phosphate extracted from HVTS and SGES was 192.4  $\mu\text{g PO}_4\text{-P/L}$  and > 2159.1  $\mu\text{g PO}_4\text{-P/L}$  respectively. A concentration of 16.2 mg  $\text{NO}_3\text{-N}+\text{NO}_2\text{-N/L}$  observed in the single sample collected from surface water with PCTS as the substrate suggested that PCTS had available N in sufficient quantities that would not likely be limiting to *T. latifolia* survival and growth in the event that soil  $\text{NH}_4\text{-N}$  were low and added available N in grow media was insufficient during the early stages of the experiment.

The PCTS was therefore selected as the topsoil component of a formulated sediment recipe of 80 % sand and 20 % topsoil given the observed very low phosphate concentrations. The PCTS consisted of 80 % organic matter (Presidents choice staff representative, personal communication) which provides a source of carbon (C) and, through microbial mineralization processes, produces electrons which are transported through microbial electron transport chains to produce adenosine triphosphate.

A high sand content was added to achieve clear water conditions, improve the detection of newly emerging ramets, and to provide a substrate commonly used in formulated sediments described in Environment Canada (1995) (*e.g.* 69 % sand in Naylor and Rodriques (1994) and 25-75 % sand in Hamr *et al.* (1994)). As *Typha spp.* can be grown hydroponically (Ye *et al.*, 1997; Woo and Zedler, 2002), a high sand content was considered feasible despite the low

organic matter and nutrient content and would minimize error associated with turbidity in future experiments involving colourimetric water quality analysis. As high seedling mortality can occur in flooded anoxic sediments (Stevens and Hoag, 2000), as was observed in the first preliminary propagation trial, the high porosity of sand substrate may facilitate the diffusion of oxygen into the sediments and improve seedling survival and growth. Several studies involving *T. latifolia* have utilized sand and/or topsoil as a rooting substrate (*e.g.* Wetzel and van der Valk, 1998; Kercher and Zedler, 2004; Wang *et al.*, 2008).

### 2.3.3.1 Sediment Preparation and Percent Moisture Content

To ensure that all microcosms received the same weight of sediment, differences in moisture content had to be corrected for such that each microcosm received the same amount of sediment in dry weight. All sand and PCTC was sieved through an M&L Testing Equipment Co. 2mm metal sieve. The dry weight of sand and top soil was determined separately. The moisture content of sand and PCTS was determined by first placing the empty, 18.9 L mixing bucket (Figure 19), cleaned according to the Ryerson cleaning protocol, on a Starfrit digital scale (11 kg capacity) and then adding sand or sediment until 5.0 kg had been added which was less than half the volume of the mix bucket. The custom-made mixing bucket was made of high density



**Figure 19. The 18.9 L sediment mixing bucket.**

polyethylene and fitted with ten 7.8 cm x 5.0 cm baffles attached with stainless steel L brackets and bolts set at various angles. Duct tape was used to seal the holes through which a ½ inch (in) (1.27 cm) I.D. copper pipe was fitted. The lid was tightly secured to the bucket and with one end

of the copper pipe resting on an elevated solid surface and the bucket was rotated end over end 25 times. Immediately after mixing, five samples of approximately 5 to 10 g and were collected from different locations at the sediment surface and were placed into a pre-weighed 80 ml glass beaker and oven dried at 60°C to constant weight.

To calculate percent moisture, sediment dry weight was divided by sediment wet weight x 100. The mean percent moisture content of the five samples was then used to convert each 5 kilogram (kg) batch of topsoil sediment to dry weight with the total dry weight value divided by 41. An equivalent amount of sediment was added for one extra microcosm (41) to ensure that a sufficient amount of soil from each batch was available for all 40 microcosms to account for minor spills. The oven-drying of sand stored in the laboratory to constant weight indicated a moisture content of 0 %. Each microcosm received 1.0 kg as dry weight (DW) of mixed sand and topsoil (800 g sand (DW) + 200 g topsoil (DW)). After the sediment and sand had been added separately, each microcosm vessel, with the corresponding labeled lid firmly secured, was turned end over end 25 times to thoroughly mix the formulated sediment.

## **2.3.4 Environmental Conditions**

### **2.3.4.1 Lighting and Temperature**

The vessels were arranged in ten rows of four under light bank 1 (LB1) with individual vessel location being randomly selected every three or four days using randomizing software to eliminate photosynthetically active radiation (PAR), measured as  $\mu\text{E m}^{-2} \text{ s}^{-1}$ , as a factor in *T. latifolia* response to the P treatments. Observed minimum and maximum temperatures and instantaneous relative humidity (RH) measurements were recorded approximately daily using a Noma digital temperature thermometer/RH sensor. As daily temperatures were not collected over the course of 24hrs, observed minimum and maximum values are reported and are an approximation of the true minimum and maximum temperatures.

### 2.3.4.2 Soil Moisture Monitoring

Soil moisture content was monitored weekly to ensure that the watering regimen provided moist - wet soil conditions throughout the experiment. Soil moisture was monitored with a Spectrum Technologies Inc. Economy Soil Moisture Tester which provides a quantitative rank of soil moisture levels as being either dry (rank = 0), average dry (2-4), average (4-6), average wet (6-8) and wet (10). Six standardized sample locations considered representative of soil moisture conditions throughout a given microcosm were established (Figure 20).



**Figure 20. Soil moisture monitoring locations**

Initially moisture conditions were determined immediately before and after watering. However, for moisture conditions from week 5 onwards, only sediment moisture conditions immediately prior to watering were measured. Moisture levels were measured approximately 2 cm below the soil surface and, with the tip of the probe in contact with the vessel bottom, at a soil depth of approximately 5.5 cm. As soil moisture testing introduced oxygen into the sediments which may affect sediment root development, oxidation reduction potential (ORP), microbial communities and potentially damage roots, soil moisture testing was restricted to microcosms which had experienced 100 % mortality after six weeks. Microcosms that had experienced 100 % mortality were treated the same way as microcosms with seedlings and continued receiving the same P treatment throughout the experiment to serve as a physical presence and to enable soil monitoring without the risk of injury to seedling roots and the alteration of sediment redox conditions.

### 2.3.5 Seedling Addition to Experimental Vessels

After each vessel had received 1.0 kg dry weight (DW) of mixed sediment, vessels were randomly assigned to one of the four treatments. A 50 mg/L PO<sub>4</sub>-P stock solution was made according to APHA (1998) by adding 2.193 g KH<sub>2</sub>PO<sub>4</sub> with the modification that MDW was used as the solvent instead of Millipore water to provide the same amount of MDW constituents to each microcosm. Seven 1.0L batches for each of the 50, 100 and 300 µg PO<sub>4</sub>-P /L treatments were made by adding 1.0, 2.0 and 6.0 ml of stock solution calculated using the dilution formula  $C_1V_1 = C_2V_2$  to the 1.0 L mark with MDW where  $C_1V_1$  = initial concentration and volume and  $C_2V_2$  = the final concentration and volume. 650 ml of MDW (reference water), or the appropriate phosphorus treatment, was added to each vessel to achieve approximately 1.0cm of standing water. A 60ml syringe was used to wash all sediment off the vessel walls using some of the 650ml of P treatment water added. Calculation details for determining reagent weights and solvent volumes of grow media stocks and treatment concentrations for all grow media are provided in Appendix B.

650 ml of MDW/P treatment were added to each microcosm to ensure that approximately 1cm of standing water was available when seeds were initially added as Sifton (1959) had observed higher germination rates with seeds submerged compared to seeds placed on a moist artificial substrate. Furthermore, the locations where germinated seeds became rooted would be randomized as there was no control over where the seeds would take root as seeds were agitated off of the dissecting needle.

#### 2.3.5.1 Grow Media Preparation

A modified grow media used by Weng *et al.* (2006) (Table 5) to simulate primary and secondary waste waters was selected as it had been effective in growing *T. latifolia* clones in gravel and *T. latifolia* in preliminary trials. As done in preliminary trials, the grow media was modified by reducing the concentration of ammonium sulfate (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> used by Weng *et al.* (2006) by 50% to reduce the risk of ammonia toxicity to newly germinated seedlings.

**Table 5. The modified grow media used in the *Typha latifolia* post germination response to phosphorus loading adopted from Weng *et al.* (2006). Micronutrients were made and stored in 250 ml volumes.**

Stock	Reagent	g/L	g/250ml	Volume of stock added to each 1.0L of grow media solution
1	MnSO <sub>4</sub> · H <sub>2</sub> O	0.098	0.392	4
1	CaCl <sub>2</sub> · H <sub>2</sub> O	0.256	1.024	2
2	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	5.2395	–	1 <sup>b</sup>
3	MgSO <sub>4</sub> · 7H <sub>2</sub> O	2.56	10.24	2
4	FeCl <sub>3</sub> · 6H <sub>2</sub> O	0.0512	0.2048	2

<sup>b</sup> 50% of (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> used by Weng *et al.* (2006).

Micronutrient stock solutions were made with Millipore water as the solvent. MnSO<sub>4</sub>·H<sub>2</sub>O and CaCl<sub>2</sub>·2H<sub>2</sub>O were maintained as a single stock solution. All grow media contained 1.11 mg NH<sub>4</sub>-N/L. A 50 mg/L PO<sub>4</sub>-P stock solution was made with Millipore water as this stock solution would be used over a long-term period. All grow media were made using acid washed volumetric flasks and stored under refrigeration (1 – 5°C).

The N : P ratios for each treatment after mixed grow media addition were estimated after the completion of the experiment using the 1.11 mg NH<sub>4</sub>-N/L added to grow media as (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> in addition to the 0.43 mg NO<sub>3</sub>-N/L indicated in the City of Toronto 2012 drinking water summary report which became available in June or July 2013 (City of Toronto, 2013b). The estimated N : P ratios for each treatment are provided in Table 6. The mean concentration of nine MDW samples determined using the CD reduction method (APHA, 1998) was 0.36 mg ± 0.05 NO<sub>3</sub>-N/L and therefore similar to the 0.43mg NO<sub>3</sub>-N/L stated in City of Toronto (2013b). It is therefore likely that the estimated TN and N: P ratios are an accurate representation of the total available N added to each of the four grow media treatments.

**Table 6. Total N, total P and N : P ratios for each of the four 0, 50, 100 and 300 µg PO<sub>4</sub>-P/L treatment grow medias**

<b>Treatment (µg PO<sub>4</sub>-P)</b>	<b>NH<sub>4</sub>-N (mg/L)</b>	<b>NO<sub>3</sub>-N (mg/L)</b>	<b>TN (mg/L)</b>	<b>PO<sub>4</sub>-P (mg/L)</b>	<b>N:P</b>
0	1.11	0.43	1.54		
50	1.11	0.43	1.54	0.05	30.80
100	1.11	0.43	1.54	0.1	15.40
300	1.11	0.43	1.54	0.3	5.13

### 2.3.6 Watering Regimen and Grow Media Addition

Grow media was added after sufficient post-germination seedling survival was confirmed and when restarting the experiment due to low seedling survival would be unnecessary. It was not feasible to add another 650 ml of grow media in one single addition as seedlings would potentially be drowned and the addition may possibly cause ammonia toxicity and excessive algal and bacterial growth. Thus, treatment grow media solutions were administered every two or three days to maintain the desired moist sediment conditions by adding P treatment in 60 ml volume increments using a 900 ml polypropylene spray bottle. At the point at which the 650 ml target was nearly reached, incremental volumes were adjusted accordingly to each microcosm to ensure all microcosms received a total of 650 ml of grow media.

One spray bottle was allocated specifically for adding MDW to all microcosms daily. A clearly labeled separate spray bottle was allocated specifically to one of the four P treatments. The volume of water added to each microcosm was determined by counting the number of sprays. The volume of water produced with each spray was determined before and after each watering and grow media addition by measuring the total volume of 100 sprays aimed directly into a 250 ml graduated cylinder tested for accuracy against a 250 ml volumetric flask. The total volume of grow media or MDW added to each microcosm for each watering was determined as follows:

$$(\text{Observed volume after 100 sprays (ml)/100}) \times \text{total number of sprays per microcosm.}$$



After grow media was added, MDW was added if necessary until it was estimated visually that approximately 10 % of the surface sediment was saturated as evidenced by standing water. Generally, standing water remained on the sediment surface for only a few minutes due to the high porosity associated with a high sand content. The amount of water added to each microcosm was increased to 50% saturation, also visually estimated, on week 8 by which time the majority of seedlings had grown to heights where burial by sediment was not a risk and by which time the water requirements of larger seedlings had likely increased. The 50% saturation procedure was maintained until the termination of the experiment.

### **2.3.7 Pore Water Sampling**

A lack of difference between mean P treatment seedling height to tallest leaf (HTTL) by week 10 raised the question as to whether a difference in pore water  $\text{PO}_4^{3-}$  concentrations differed between treatments given  $\text{PO}_4^{3-}$  is readily adsorbed to sediment P binding sites or coprecipitated with soluble cations. Pore water was sampled on August 09, 10 and 13, seven to ten days before shoots were removed to determine microcosm biomass. Other more commonly used methods for sampling pore water were less effective in collecting representative samples from sandy sediments (Berg and McGlathery, 2001). Dialysis cells which must remain in sediments for at least one week are representative of average conditions over time and so may not be representative of conditions at a specific point in time. The squeezing of sediment cores is not possible with sandy sediments as pore water drains out of the sample as soon as the core is sectioned (Berg and McGlathery, 2001).

Berg and McGlathery (2001) developed a 45 cm long stainless steel pore water sampler 2.4 mm in outer diameter, 1.8 mm inner diameter and capped at the end with silver soldering. Four 0.38 mm diameter holes were drilled 2mm above the pore water sampler tip. To approximate the pore water sampler designed by Berg and McGlathery (2001), plastic disposable 2 ml pipettes, 4.06 mm I.D., 6.58 mm O.D., 28.5 cm in length, were modified for sampling sandy sediments by drilling four roughly equidistant 1.18 mm diameter holes perpendicular to the pipette along the 0 ml mark using a 1.18 mm (0.0465 in) drill bit.

The main dispensing hole of the pipette was sealed by tightly wrapping Parafilm around the pipette tip. A pipette pump was used to draw pore water. Each modified pipette was tested by filling a beaker with distilled water to a depth just below the 1.18 mm holes and then engaging the pipette pump. The pipette was re-wrapped with Parafilm if distilled water entered the pipette. All pipettes were thoroughly flushed with distilled water and acid rinsed after each use.

Up to ten 2 ml subsamples were collected depending upon pore water availability. The pipette was inserted into the sediment until the tip came into contact with the vessel bottom ensuring that pore water was collected from a consistent depth. In general, five subsamples were collected across one length at approximately 1/3 of the width of the microcosm, spaced approximately 2 – 3 cm apart. A second set of five subsamples was collected at approximately 2/3 of the width of the microcosm. Subsampling immediately adjacent to shoots was avoided to prevent root damage. Microcosms with and without plants were also sampled to determine seedling effect on pore water phosphate concentrations. Each subsample was dispensed into an 80 ml beaker to make one composite 15 to 20 ml sample. After pore water sampling was completed for each microcosm, approximately 20 ml of MDW was added to replace the water removed.

After the final subsample had been collected, pore water pH and temperature were measured immediately with a Eutech Instruments/Oakton Multiparameter PCS tester™ 35. Temperature was measured with a Sper Scientific dissolved oxygen meter. Dissolved oxygen (DO) and oxidation reduction potential (ORP) in pore water were also determined; however, several minutes could pass before the composite sample was collected which resulted in elevated ORP and DO readings. Also, oxygen would have been introduced into the sediment during the subsampling process thus the data were not used in analysis. On August 13, 2012 DO and ORP were measured immediately after subsampling to reduce error. Similarly, although DO and ORP readings were lower, the accuracy of the DO and ORP data may be erroneous thus the data collected on August 13, 2012 were also not analyzed. All meter probes were rinsed thoroughly with distilled water after use, placed into a beaker containing distilled water and then rinsed again before data were collected from the subsequent sample.

### 2.3.8 Seedling Measurements and Selection

Seedling growth was monitored bimonthly by determining mean HTTL measured to the nearest mm and the mean number of leaves per microcosm. Maximum leaf width, the largest leaf width for a given shoot, was determined to the nearest mm with a flexible plastic ruler. Preliminary trials indicated that the use of a caliper was not practical for young *T. latifolia* due to the very soft leaf tissue and the difficulty of manipulating the caliper within the confines of the microcosm vessel which risked causing injury to seedlings. The post germination mortality rate for each microcosm was monitored throughout the study. A seedling was not considered a mortality until the entire shoot had turned completely brown.

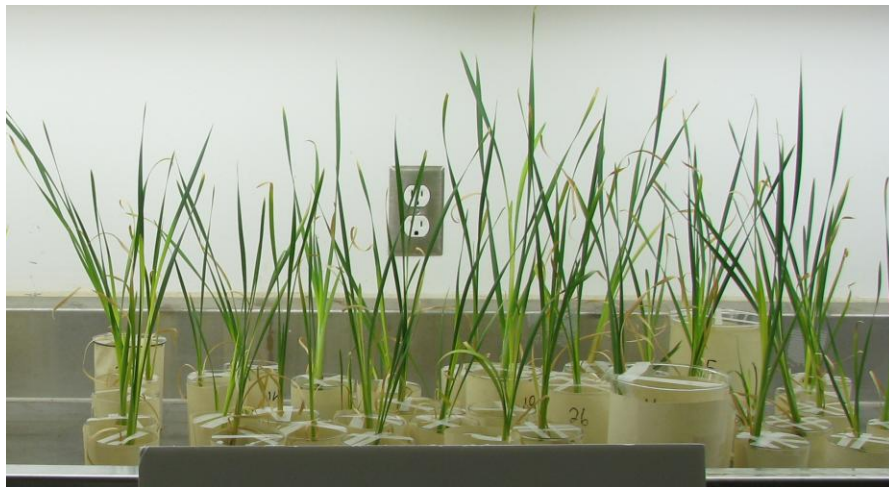
Due to the uniformity in pore water  $\text{PO}_4^{3-}$  concentration and seedling HTTL (Results, Section 3.1.5), and the decrease from 40 microcosm replicates to 29 due to mortality, it was decided to terminate the experiment on day 82 to redistribute surviving seedlings to reestablish 40 microcosm replicates in preparation for the subsequent experiment involving additions of both N and P under flooded sediment conditions. After measuring HTTL, the number of leaves and maximum leaf width for each shoot, wet weight (fresh weight, live weight) biomass was determined for shoots used in the P experiment at the end of the study on day 82, 84 and 85 using a Sartorius digital balance accurate to 0.01 g. Prior to seedling removal, MDW was added to each microcosm to loosen soil in order to reduce damage to roots and rhizomes. Some damage however was unavoidable and major losses of root hairs occurred.

Removed seedlings were immediately rinsed with MDW to remove all traces of sediment, pat dried with paper and weighed to the nearest 0.01 g once the weight indicated on the balance had stabilized. Each seedling was then transferred to a previously prepared beaker of appropriate size such that the root mass could easily fit within the beaker without causing injury. The sediment was sifted through by hand to locate torn roots which were placed on a 2 mm sieve, rinsed over a bucket to receive sediment-laden water, pat dried, weighed to the nearest 0.01 g and then discarded.

## 2.4 *Typha latifolia* response to Phosphorus and Nitrogen Experiment

### 2.4.1 Seedling Recovery

Immediately after weighing, the high P treatment grow media ( $300 \mu\text{g PO}_4\text{-P/L}$ ) was added to each recovery beaker and repeated approximately weekly until shoots were fully recovered and replanted in 3.1 L microcosms. 1.0 g of Plant Prod Smartcote 14N-14P-14K slow release fertilizer pellets was also added to each beaker. Watering was conducted as required in order to keep the seedling root masses submerged. For the majority of seedlings one to four Parafilm strips were placed across the mouth of the beaker to enable the shoot to maintain vertical position. Beakers were then wrapped in paper to provide shade to reduce algal growth (Figure 21).

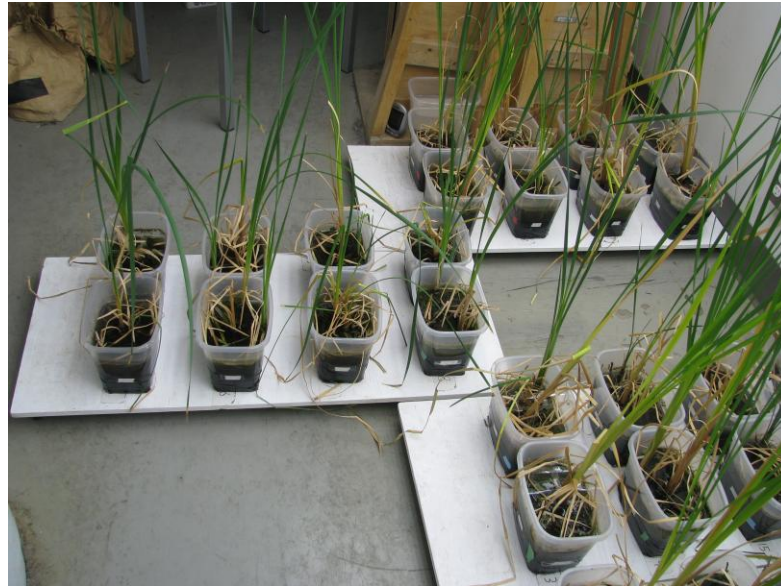


**Figure 21. *Typha latifolia* recovering shoots being grown hydroponically in grow media on August 31, 2012 after removed from sediment, August 16 to 19, 2012.**

Shoot locations were randomized every three or four days to provide approximately equal levels of PAR to each shoot during the recovery period. After approximately one to two weeks of showing signs of stress (yellowing leaves), shoots started to recover as evidenced by the production of new leaves, root hairs and ramets (vegetatively produced shoots) from rhizome. All shoots grown in SGES were unearthed on August 29 and recovered in the same manner as for the shoots used in the *T. latifolia* post germination response to P loading experiment. Lighting conditions were the same as described in Section 2.4.4.

## 2.4.2 Seedling Access

Light bank 3 (LB3), the light bank suitable for the propagation of plants to heights of > 50.0cm, was necessarily situated against a wall which limited access. To fully utilize the area beneath the light bank and access microcosms with minimal risk of injury to shoots, five custom-built mobile carts were constructed (Figure 22).



**Figure 22. Cart being pulled out from underneath light bank 3 to access microcosms.**

The carts were made from 5/8 in (1.6 cm) plywood cut to 48.2 cm in width and 107.0 cm in length and painted off-site with latex anti-fungicide white paint on all sides to prevent the establishment and spread of fungus and warping of the plywood. Five caster wheels were attached to the bottom of each cart with one wheel placed at each corner of the plywood sheet and a fifth wheel being placed in the middle of the cart to prevent sagging from the weight of eight microcosms. When access was required for watering, the adding of grow media or sampling, the cart was pulled out from underneath the light bank with the individual microcosm removed and the cart promptly returned to its original position while watering, feeding or sampling was conducted

## 2.4.3 Seedling Acclimation

### 2.4.3.1 Formulated Sediment Preparation

Following the termination of the *T. latifolia* post-germination response to P experiment, the remaining sediment was air-dried to facilitate mixing for reuse in the later experiment conducted to determine *T. latifolia* responses to oligotrophic and eutrophic N and P loading rates. The reuse of sediment was considered justifiable given that no differences in *T. latifolia* biomass were observed between the four treatments and that a very low amount of P had been added to all treatments relative to sediment P levels reported for eutrophicated wetlands (*e.g.* Mayer *et al.*, 2006; Mayer *et al.*, 2008).

To reduce variation in sediment composition between individual microcosms, sediment was air-dried and thoroughly mixed. Reference sediment was air-dried first and kept separate from P treatment sediments and reused as reference sediment for the subsequent experiment. All recovered reference sediment was stored in an 18 L food-grade bucket, cleaned following the Ryerson cleaning protocol, with the lid on until final mixing and sediment percent moisture was determined.

The 50, 100 and 300 ug PO<sub>4</sub>-P/L treatment sediments were combined into three separate batches of under 10 kg by adding 1/3 of the sediment recovered from each microcosm vessel into one of three clean plastic buckets. Before the sediment from individual vessels was combined, the sediment in each vessel was individually mixed by rotating the vessel end over end 25 times with the lid secured. The weight of sediment was divided by 3 with 1/3 of the sediment being added to one of three 18 L food-grade plastic buckets that were cleaned according to the glassware cleaning protocol and stored with the lid in place until percent moisture was determined.

Sediment percent moisture content for the reference sediment and the combined treatment sediment was determined three to seven days prior to the transplanting of shoots using the same procedure as described in section 2.1.4.2 with the only modification being that just over 9 kg wet weight (WW) of sediment was added to the mix bucket instead of the previously added 5 kg to

speed the sediment preparation process and initiate the shoot acclimation process as soon as possible. The mixing bucket was cleaned following the glassware washing protocol before the reference sediment was mixed.

Preliminary trials indicated that large shoots tended to lean or topple due to either an insufficient rooting surface area and/or depth and/or loosely bound sediment. It was therefore decided to add granite stone with relatively inert chemical reactivity to the bottom of each microcosm to provide better anchoring support for shoots. Commercially available Premier beach/river granite stones were sieved through a 12.7mm (1/2in) Hoskin Scientific metal sieve to remove small stones. Stones were then rinsed with MDW to remove fine particulates and washed according to the glassware cleaning protocol to remove N and P and other contaminants. As the size of individual stones differed considerably, 7 to 17 stones were added to each 3.1 L vessel to achieve an approximately equal weight of 412 to 427g per microcosm and stored with the lids on. All microcosm vessels were washed according to the glassware washing protocol before the granite stones were added.

#### **2.4.4 Seedling Transplanting**

By September 12 all seedlings used in the *T. latifolia* response to P experiment, except the smallest seedlings, had recovered and many had produced new ramets. A total of 36 genets (original shoot plus newly produced ramets) that were used in the post germination *T. latifolia* experiment plus four seedlings of similar size that had been grown in SGES were transplanted into 3.1 L microcosm vessels containing river stone from September 12 to September 14.

Genets were gently rinsed with MDW to remove high nutrient grow media and slow release fertilizer pellets then rinsed with distilled water. Shoots were briefly placed in a beaker to allow bulk water to drain off before weighing. Roots were not pat-dried to avoid root damage. All genets were weighed to the nearest 0.01 g using a Sartorius digital balance. Genet roots were then gently laid on top of the granite stones (Figure 23) and, with the shoot(s) held in a vertical position, 725 g of formulated sediment (DW) were added on top of the root mass and around the shoot. 500 ml of grow media were promptly added to the microcosm and placed under LB3, a

custom-built light bank 2.4 m in length and 1.2 m in width fitted with the same 16 X 121.9 cm T8 VitaLux light bulbs used in previous trials and experiments.

Genets were weighed to the nearest g (WW) using a Starfrit digital balance to determine if biomass would be a factor in genet survival during the acclimation period in the event that significant shoot mortality was observed. As shoots were not pat-dried, weights are not comparable to shoot weights obtained on August 16, 18 and 19. As substantial growth had occurred since the start of the acclimation period on September 12 – 14, pre-acclimation genet biomass was not considered representative of genet biomass by the time the experiment was started on November 08. Pre-acclimation biomass was therefore not included in future analysis.



**Figure 23. Genet roots being gently laid on granite stones. Note the digital balance used to measure the weight of the genet and, immediately afterward, used to add 725 g of formulated sediment as DW.**

Three genets that were showing signs of stress (yellowing and decreasing number of leaves) acclimating in microcosms 33, 17 and 11 were replaced with three genets growing under LB1 in hydroponic grow media solution on September 24, 2012. Three further genets replanted into microcosms 39, 16 and 7 died by October 25, 2012 and were replaced with recovered genets which served as a physical presence only as they were not acclimated under oligotrophic conditions sufficiently long enough to be considered representative of the other shoots. The addition of three genets on October 25 served to eliminate gaps in the canopy and therefore



prevent unnecessary edge effect which could create a significant source of error by affecting PAR availability and competition for light between individual genets.

Each microcosm was randomly assigned to one of 40 locations under LB3 at the start of the acclimation period. Eight locations were provided on each of the five carts. As cart locations were randomized, microcosms remained on the same cart for the duration of the acclimation period and the duration of the subsequent experiment.

#### **2.4.5 Grow Media and Watering Regimen**

All forty microcosms were acclimated with the 10  $\mu\text{g PO}_4\text{-P/L}$  added to the reference grow media described in the previous experiment from September 12 until the allocation of treatment P and N levels were started on November 08. The 1.54 mg TN /L (1.11 mg  $\text{NH}_4\text{-N}$  + the estimated 0.43 mg  $\text{NO}_3\text{-N}$  in MDW) was within the range observed in oligotrophic lakes (Wetzel, 1983).

A large Rubbermaid plastic garbage can cleaned according to the glassware washing protocol was used to make 21.0 L of grow media as described in Table 5, Section 2.1.8 in order to provide 500 ml of oligotrophic reference grow media for each microcosm. Using a 1.0 L volumetric flask, 21.0 L of grow media was made by initially adding 5.0 L of MDW in 5 X 1.0 L increments to the Rubbermaid can to facilitate reagent mixing followed by the addition of 1.0 L containing each stock solution at 21 X the volume added to 1.0 L (*e.g.* 1 ml of  $(\text{NH}_4)_2\text{SO}_4\text{/L}$  X 21.0 L = 21 ml of  $(\text{NH}_4)_2\text{SO}_4$ ) followed by the addition of 15.0 L of MDW added in 15 X 1.0 L increments. 500 ml of grow media was added to each microcosm immediately after each seedling was re-planted into formulated sediment to ensure moist to wet sediment conditions. The grow media was stirred periodically between transplants. 200 to 300 ml of grow media were added every 2 to 5 days thereafter as required to maintain moist soil conditions.

Subsequent batches of 2.1 to 4.1 L of grow media, depending on the volume needed, were made fresh from stock using 1.0 L acid rinsed volumetric flasks. The volume of stock solution added to the 1.0 volumetric flask was sufficient to provide equal volumes of grow media to 41 microcosms, with extra grow media equivalent to the volume needed for one additional

microcosm in the event of minor spills. A 250 ml graduated cylinder determined to be accurate by filling with a 250 ml volumetric flask, was used to measure and dispense grow media volume. The graduated cylinder was labeled and used throughout the acclimation period and the subsequent experiment. Grow media and MDW were dispensed through a 5.0 L watering can plastic nozzle to prevent sediment scour and the exposure of roots until November 08 when microcosms were filled to the 900 ml mark. If necessary, MDW was added in addition to grow media and, on days between grow media additions, MDW was added to maintain moist soil conditions until October 25 when microcosms were filled to the height of the black duct tape (approximately 1.0 cm) until November 08 in order to acclimate seedlings to saturated soil and low sediment redox conditions. Moist soil conditions were preferred over flooded or saturated conditions initially in order to allow the sediments to compact to improve root anchoring, reduce turbidity and enable oxygen to penetrate the sediments which may be a contributing factor to shoot growth and survival in young *T. latifolia* shoots.

#### **2.4.6 Environmental Conditions**

Minimum and maximum temperature and relative humidity (RH) were recorded every one to three days for the duration of the acclimation period using a Noma digital thermometer/RH humidity sensor and an Accutemp digital thermometer/RH sensor with the latter serving as a backup in the event of equipment failure. As daily temperatures were not collected over the course of 24hrs, minimum and maximum values were an approximation of the true minimum and maximum value. To eliminate bias associated with edge effect and variations in light intensity, temperature and RH, microcosm locations were randomized throughout the acclimation period every three or four days. Randomization was achieved by temporarily removing the southerly most cart from underneath the light bank, shifting the adjacent cart to the southerly most position, then shifting each microcosm position on the cart moving in a clockwise direction. The process was repeated for all five carts. The cart that was removed was moved to the northern most position. This method of randomization was repeated throughout the entire acclimation process and continued throughout the following experiment.

### 2.4.7 Biological Data

Pretreatment data were collected to determine whether differences in growth rate as a function of change in total leaf length (TLL), defined as the sum length of all green leaves, differed significantly between genets at the time they were randomly assigned to one of three N and P treatments on November 08, the day N and P treatments were first applied. A green leaf was defined as a leaf with any amount of green pigment capable of photosynthesis (Figure 24).



**Figure 24. A representative senescent leaf showing green pigment and counted as a green leaf.**

TLL was used by Woo and Zedler (2002) as a non-destructive method for monitoring *T. x glauca* response to N and P treatments and was reported to be a highly effective indicator of genet biomass (J.B. Zedler, University of Wisconsin Botany Professor and wetland restoration ecologist, personal communication). A standard tape measure was used to measure TLL to the nearest mm. To avoid cross-contamination between microcosms, the tape measure was cleaned before each microcosm TLL was obtained by firstly pre-rinsing in Millipore water followed by rinsing in 10% HCl and lastly rinsing again with Millipore water.

Also documented was the number of shoots per microcosm, the total number of leaves and the height to green tissue in senescent leaves. However, no studies were found in the literature that measured *Typha* leaves to green height as an indicator of fitness or to correlate with biomass. The number of leaves and height to green tissue were considered redundant with TLL and were therefore not analyzed but are available for comparison with future research.

## **2.5 *Typha latifolia* Response to Oligotrophic and Eutrophic N and P Loading Rates under Laboratory Conditions.**

A full-factorial experimental design was used to test the hypothesis that environmentally relevant (eutrophic to hypereutrophic) N and P loading rates loading would increase biomass and that high N availability combined with high P availability would result in significantly more biomass production. A reference treatment of 10  $\mu\text{g PO}_4\text{-P/L}$  was selected as this concentration was similar to the 8.0  $\mu\text{g TP/L}$  mean concentration indicated in Wetzel (1983) for oligotrophic systems. A treatment of 300  $\mu\text{g PO}_4\text{-P/L}$  was selected as  $> 300 \mu\text{g PO}_4\text{-P/L}$  has been observed in agricultural runoff (Yates and Prasher, 2009). The 1.0 mg TN/L (0.5 mg  $\text{NO}_3\text{-N /L}$  + 0.5 mg  $\text{NH}_4\text{-N/L}$ ) and 3.0 mg TN/L treatments (1.5 mg  $\text{NO}_3\text{-N /L}$  + 1.5 mg  $\text{NH}_4\text{-N/L}$ ) as a TN loading rate were selected as 1.0 mg TN/L is within the oligotrophic and 3.0 mg TN/L is within the eutrophic range indicated in Wetzel (1983) respectively. The 3.0 mg TN/L treatment is also a concentration in which species richness in submergent vegetation is comparatively low indicating environmental impact or possibly displacement of species adapted to N – poor systems by competitive species (James *et al.*, 2005).

## **2.6 Sediment**

The preparation of the sediment used in this experiment is described in Section 2.3.1.

## **2.7 Environmental Conditions**

Lighting and photoperiod conditions were the same as those described for the acclimation period. A 16.0 L: 8D photoperiod was maintained throughout the experiment except for an unknown period between December 29, 2012 and January 03, 2013 during which the main

power to the light bank was shut off for a building-wide electrical upgrade. The mean height of the light bank bulbs to the sediment surface with microcosms placed on carts was  $1.73 \text{ m} \pm 0.008 \text{ m}$ ,  $n = 5$ . The Field Scout Type 2 light meter was taped to an extension sufficiently long enough to determine PAR for all microcosms without having to move carts (Figure 25).

To characterize light intensity, PAR was measured at (1) sediment height adjacent to each microcosm; (2) as close to the center of each microcosm as possible with the sensor extension resting ontop of the lip and perpendicular to the vessel, and (3) at HTTL on November 09, 2012, one day after treatments were first added, and on May 08, 2013, two days before the start of plant removal. Temperature and RH monitoring was continued as described in section 2.6.4.



**Figure 25. Measuring light intensity under LB3 with an extension on May 08, 2013.**

### **2.7.1 Grow Media and Watering Regimen**

Micronutrients  $\text{MnSO}_4 \cdot \text{H}_2\text{O}$ ,  $\text{CaCl}_2 \cdot 2 \text{H}_2\text{O}$ ,  $\text{MgSO}_4 \cdot 7 \text{H}_2\text{O}$  and  $\text{FeCl}_3 \cdot 6 \text{H}_2\text{O}$  were added to each microcosm one final time on November 08 with only N and P being added thereafter. Grow media was made fresh from stock solutions on the day of addition.  $\text{NO}_3\text{-N}$  stock solution was made from  $\text{NaNO}_3$ . Details and calculations on stock solution reagent concentrations and grow media preparation are provided in Appendix B.

From November 08 to December 08 either 200 ml or 300 ml of grow media at the desired concentration was added every six to nine days from November 08 to November 22, 2012 at treatment concentrations (10 µg TP/L or 300 µg TP/L and 1.0 mg TN/L or 3.0 mg TN/L depending on treatment) with all microcosms receiving the same volume of grow media for every watering or feeding event. MDW containing a mean concentration of 0.43 mg NO<sub>3</sub>-N (City of Toronto, 2013b) was added to each microcosm in equal volumes between November 08 and December 08 but was discontinued thereafter due to high salinity levels in some microcosms were approaching brackish levels of 0.5 mg/L.

The volume of grow media and MDW added was determined by measuring the distance between the observed water level and the full water mark of the microcosm with the greatest amount of surface water remaining since the last watering and/or grow media addition. It had previously been determined that a 1.0 cm decrease in water level was the close equivalent of a 200ml water loss to evapotranspiration. After the same volume of grow media and MDW was added, Millipore water was added until the 900 ml mark was reached.

As the amounts of N and P being added up to December 08 may have been insufficient to obtain a detectable a response in *T. latifolia* biomass, N and P concentrations in grow media was increased such that, after addition, treatment concentrations were achieved at the 900ml volume (full water mark). The addition of grow media continued on a weekly basis. Watering with Millipore water to the 900ml was conducted every third or fourth day. Both feeding and watering were conducted on the same day that microcosm locations were randomized to reduce handling effect.

## **2.7.2 Water Quality Monitoring**

Water temperature, dissolved oxygen, pH, conductivity and salinity were monitored monthly to (1) determine if microcosm surface water conditions were representative of natural wetlands; (2) identify changes which may affect P and N fate and speciation and therefore explain observed *T. latifolia* response; (3) differences in water quality between treatments that may account for observed differences in growth and biomass and (4) to identify potential issues

such as elevated salinity levels that may affect *T. latifolia* survival which could be rectified. Three randomly selected microcosms from each of the four treatments were used to monitor surface water dissolved oxygen, temperature, pH, conductivity and salinity monthly. Dissolved oxygen and temperature was measured with a Sper Scientific model 850041 DO meter, pH with a Metler Toledo Five Easy pH meter and conductivity and salinity with a PCS tester™ 35. All meters were calibrated before use according to manufacturer specifications. When it became apparent that salinities were high, the salinity levels in the surface water in all microcosms were sampled monthly. The water quality in all microcosms were measured in the winter, 2013 to determine if the three randomly selected microcosms were representative of water quality conditions for all treatment replicates and at the termination of the experiment.

### **2.7.3 Biological Endpoints**

The determination of TLL as described under section 2.6.5 was conducted bi-monthly to monitor growth and to estimate when genet biomass for all treatments either reached a plateau or declined at which point the study would be terminated. At the end of the experiment the TLL for each microcosm was measured immediately before the microcosm was dismantled to determine the amount of variation in biomass explained by TLL. Mean growth expressed as TLL for each treatment from November 08 onwards was determined by subtracting the TLL from the preceding sample collected two weeks prior (e.g., growth between November 07 or November 08 to November 22 = week 4 TLL – week 2 TLL). The average growth for all two week intervals for each treatment were then averaged to determine mean growth for the duration of the experiment. The removal of shoots from all microcosms was conducted over a seven day period from May 10 to May 16. The order of microcosm removal was randomly determined. Microcosm maximum leaf width was determined at the end of the experiment using a Mastercraft digital caliper accurate to 1/100<sup>th</sup> of a mm to potentially explain differences in above ground biomass between shoots of similar heights.

Shoot density, the number of shoots/m<sup>2</sup>, was also determined to enable comparisons with shoot density reported in the literature. The surface area used to calculate shoot density was calculated from the mean of 18 length measurements measured every 1.0 cm and 11 width measurements measured every 1.0 cm obtained from one microcosm. The surface area value

was also used to determine the loading rate for each treatment by summing the total weight of N and P added to each microcosm. The number of days from November 08, 2012 to May 10, 2013, the forming being the start of experiments and May 10 being the start of microcosm dismantling, was divided by 365 to convert the duration of the experiment to years to provide a loading rate to  $\text{g N/m}^2/\text{yr}$  and  $\text{g P/m}^2/\text{yr}$  in order to compare N and P loading rates provided in the literature.

Above ground and below ground biomass, measured as dry weight, were determined to identify differences in N and P allocation between treatment groups. Above ground biomass as was determined by cutting shoots at the soil surface. Perched shoots (shoots in which a portion of the root mass was above ground) were cut immediately above the first root or rhizome. All leaf litter was removed, rinsed with MDW, gently squeezed to remove bulk water and placed with shoots into open plastic bags. The soil mass containing roots was placed into a bucket to soak briefly to ease root removal (Figure 26). Loose roots mainly floated and could be easily



**Figure 26. Root mass soaking to facilitate sediment removal.**

removed. The sediment at the bottom of the receiving bucket was sifted by hand for roots. Roots gently squeezed to remove bulk water, air-dried and allowed to air dry until all shoots and roots had been removed and washed. Root masses were separated and placed in to separate paper bags from the shoots and leaf litter obtain below ground biomass (DW) and above ground biomass (DW). Paper bags containing root masses and combined shoot and leaf litter were oven-dried at  $60^{\circ}\text{C}$  to constant weight. Dried root masses and combined shoots and leaf litter were measured to the nearest g using a Sartorius digital balance. Total biomass for each



microcosm, measured as the sum of the below ground and above ground biomass, was used to determine the overall effect of N and P on biomass production and therefore potential invasiveness.

#### 2.7.4 Statistical Analysis

All statistical analysis was performed in Systat<sup>®</sup> version 12 statistical software. Statistical tests used were selected based on procedures described in Zar (1984). The null hypothesis rejection region was set at  $\alpha = 0.05$ . The Levenes test was used to test each data set for equal variance. The Shapiro-Wilks test was used to determine if data residuals were normally distributed. Where parametric assumptions were met, the ONEWAY ANOVA test (referred to as ANOVA hence forward) was used to compare means between treatment groups. Where a significant difference was indicated, the post-hoc Tukey multiple range test was used to identify significant differences between individual groups. Where the assumptions of equal variance and normal distribution were violated and transformed data did not meet the parametric ANOVA test assumptions, the non-parametric Kruskal-Wallis rank sum test was used. Where significant differences were observed, the Mann -Whitney U test was used to conduct pair-wise comparisons to identify significant differences between individual groups (Dr. A. Laursen, Ryerson University, personal communication).

Tests to determine differences in pore water mean PO<sub>4</sub>-P concentrations, live shoot biomass and mean HTTL between 0, 50, 100 and 300 µg PO<sub>4</sub>-P/L treatments were conducted for the first protocol examining *T. latifolia* seedling response to P treatments. For microcosms containing more than one shoot germinated from seed, the mean HTTL between two or three shoots was used in statistical analysis and figures. Biomass for a given microcosm was the sum of all live shoots, live and dead roots and leaf litter.

At the termination of the acclimation period and the start of the *T. latifolia* response to oligotrophic and eutrophic N and P loading rates experiment, differences between TLL, HTTL, shoot density and growth rate for each microcosm observed on November 08 was tested for to determine if statistically significant differences existed prior to the addition of N and P

treatments. At the termination of the experiment, differences between TLL, HTTL, shoot density and shoot, root, root plus shoot biomass and root to shoot ratio were tested for. Regression-correlation was used to identify relationships between TLL and shoot, root, and root plus shoot dryweight biomass.

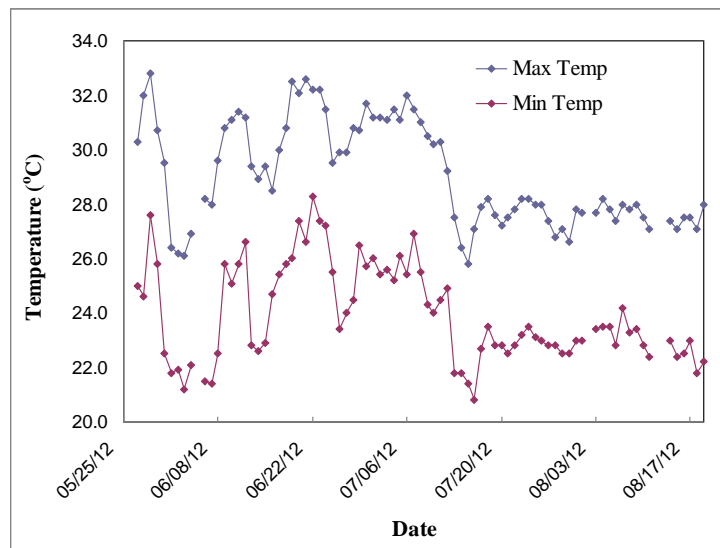
### 3 RESULTS

The objectives of the experiments were to (1) establish whether *T. latifolia* could successfully be propagated from seed to adult under laboratory conditions and using a formulated sediment that could be effective for hypothesis testing and (2) test the hypothesis that *T. latifolia* is invasive under environmentally relevant high N and high P (eutrophic to hypereutrophic) concentrations.

#### 3.1 *Typha latifolia* Post Germination Response to 0, 50, 100 and 300 $\mu$ g PO<sub>4</sub>-P/L Treatments.

##### 3.1.1 Environmental Conditions

Laboratory air temperature fluctuated widely from May 25 to July 16, 2012 ranging from 20.8°C to 32.8°C during which period air conditioning was not operational (Figure 27). Temperatures stabilized after air conditioning was restored with air temperatures generally remaining between 22 to 28°C until the termination of the experiment on August 16.



**Figure 27. Observed minimum and maximum laboratory air temperatures collected beneath the light bank used to determine *Typha latifolia* post termination response to P treatments.**

Relative humidity (RH) was monitored in order to fully describe the environmental conditions that occurred for the duration of all experiments. As the Noma digital temperature/relative humidity sensor did not record minimum and maximum RH as it does for temperature, an Accutemp thermometer/RH sensor was used. Observed RH ranged from 22% to 60% RH. The daily average minimum and maximum RH observed for the duration of the experiment was 37.6% and 46.4% respectively.

Mean PAR on May 25, 2012 at the start of the experiment approximately 5 cm above the soil was  $144 \mu\text{E cm}^{-2} \text{ s}^{-1}$  and ranged from 102 to  $182 \mu\text{E cm}^{-2} \text{ s}^{-1}$ . Mean PAR at the end of the experiment on August 16 to August 19 was  $228.6 \mu\text{E cm}^{-2} \text{ s}^{-1}$  light measured at HTTL and ranged from 123 to  $328 \mu\text{E cm}^{-2} \text{ s}^{-1}$ .

In general, a rank of 6 to 8 (average wet) was consistently observed throughout the study at approximately 2 cm below the soil surface. Readings below a rank 5, or average moisture conditions, were rarely observed. Similarly, the mean rank of 9 to 10, an indication of wet sediment conditions according to manufacturer specifications, was consistently observed throughout the study at a depth of approximately 5.5 cm with the tip of the probe in contact with the vessel bottom.

### **3.1.2 Viable Seed Identification**

Of the 82 seeds identified as nonviable seeds on May 25, three seeds had germinated by June 01 for a total percent germination of 3.4%. The three seeds that germinated were not seen within the plumage and remnant flower parts attached to clusters of two or three carpodia and were the result of observational error rather than incorrect identification.

100% germination was observed in three of the six Petri dishes into which seeds classified as viable were placed (Table 8). The overall percent germination for seeds identified as viable was 93.1% indicating that seeds added to each of the 40 microcosms were potentially viable seeds and that any differences between would not be due to experimental error. Three of the 57 germinated seeds identified as viable produced short white radicles and were considered unlikely to have survived to take root.

**Table 7. Percent germination of seeds classified as nonviable.**

<b>Classified as Nonviable Seed</b>			
<b>Petri Dish</b>	<b>Total number of seeds added May 25</b>	<b>Number germinated by June 01</b>	<b>Percent Germination</b>
1	13	1	7.7
2	12	0	0.0
3	10	0	0.0
4	16	2	12.5
5	15	0	0.0
6	16	0	0.0
<b>Total</b>	<b>82</b>	<b>3</b>	<b>3.4</b>

**Table 8. Percent germination of *Typha latifolia* seeds classified as viable.**

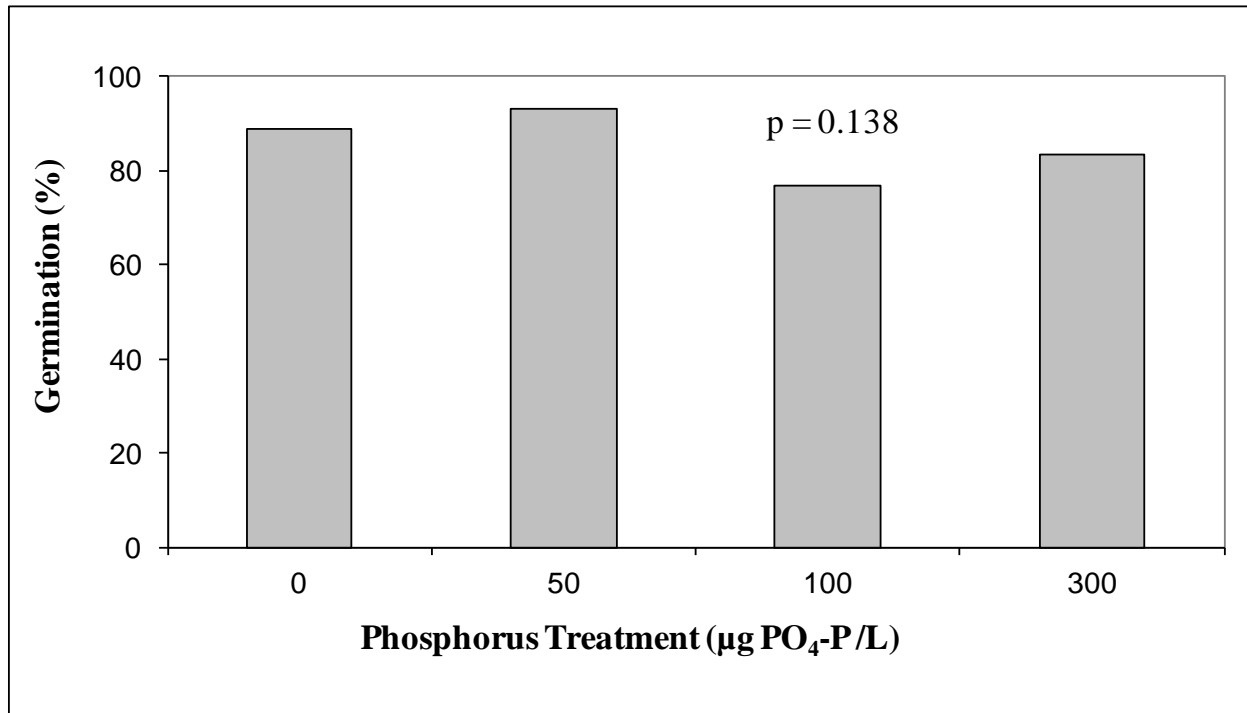
<b>Classified as viable Seed</b>			
<b>Petri Dish</b>	<b>Total number of seeds added May 25</b>	<b>Number germinated by June 01</b>	<b>Percent Germination</b>
1	11	11	100.0
2	10	10	100.0
3	10	8	80.0
4	11	10	90.9
5	11	11	100.0
6	8	7	87.5
<b>Total</b>	<b>61</b>	<b>57</b>	<b>93.1</b>

PAR ranged from 50 to 64  $\mu\text{E m}^{-2} \text{s}^{-1}$  with mean PAR being identical for the nonviable and viable seed groups at 62.2  $\mu\text{E m}^{-2} \text{s}^{-1}$ . Ambient air temperatures during the May 25 to June 01 germination period ranged from 21.8°C to 32.8°C.

### 3.1.3 Percent Germination

The total percent germination determined from May 27 to June 01 was similar between treatments (Figure 28) with the 0 and 50 $\mu\text{g PO}_4\text{-P}$  treatments having the highest percent germination at 88.9 and 93.3% respectively. No newly germinated seeds were observed after

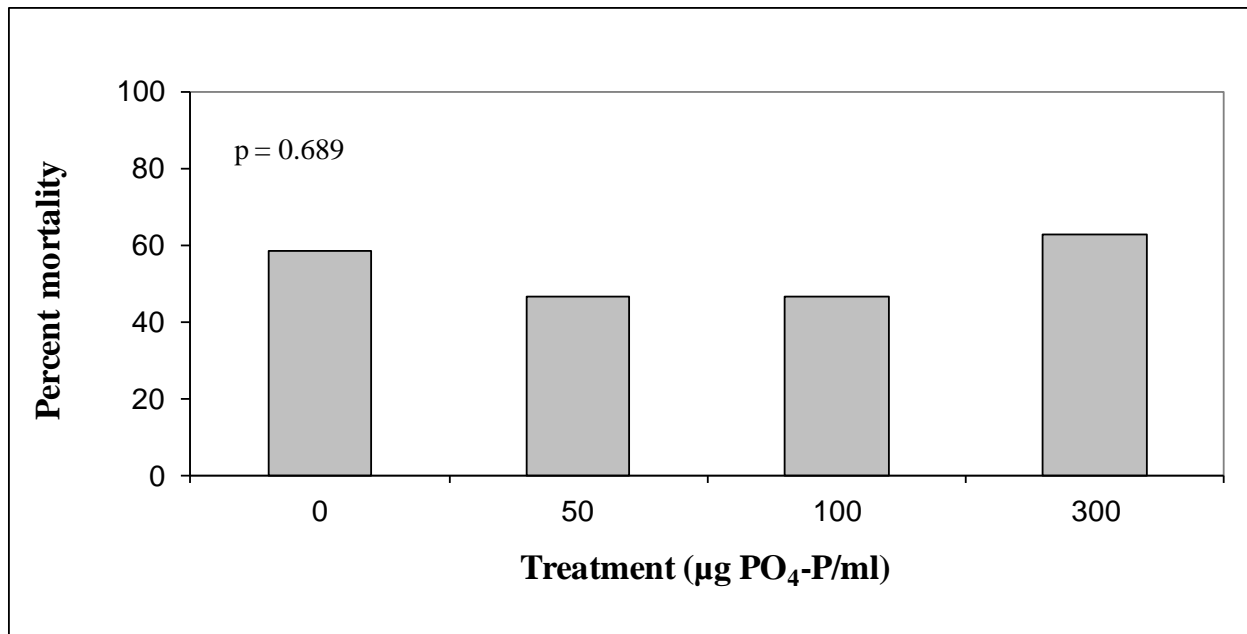
June 01. Overall percent germination for all four treatments combined was 85%. Differences in percent germination between treatments were not statistically significant ( $p = 0.138$ )



**Figure 28. Total percent germination over a six day period for *Typha latifolia* seeds observed in each of the four  $\text{PO}_4\text{-P}$  treatment microcosms.**

### 3.1.4 Percent Mortality

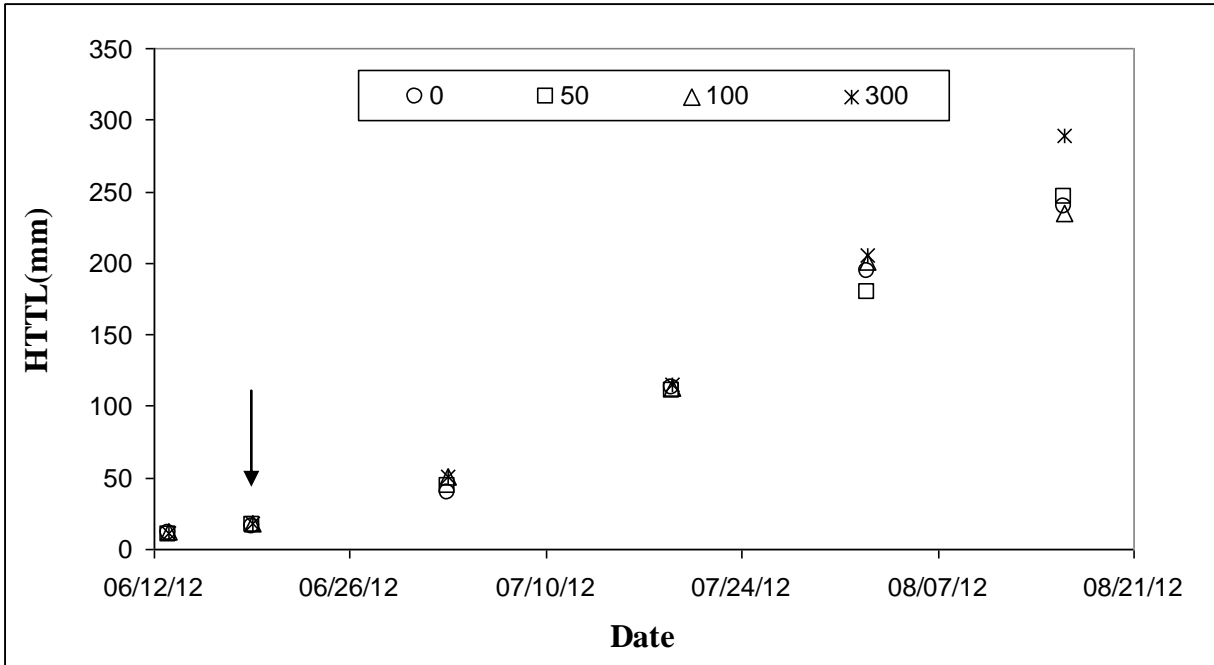
Overall mortality of post germinated seedlings was 53.5% from May 27 to August 16. The 300  $\mu\text{g PO}_4\text{-P mg/L}$  treatment experienced the greatest mean percent mortality from June 01 to August 16 at 62.5 % with the 0, 50 and 100  $\text{PO}_4\text{-P mg/L}$  treatments experiencing 58.3, 47.7 and 41.4 % mortality respectively. The differences in percent mortality were not statistically significant (Kruskal Wallis,  $p = 0.689$ ), Figure 29. Each treatment group included microcosms that experienced either 0 to 100% mortality.



**Figure 29.** Post germination percent *Typha latifolia* mortality from June 01 to August 16, 2012.

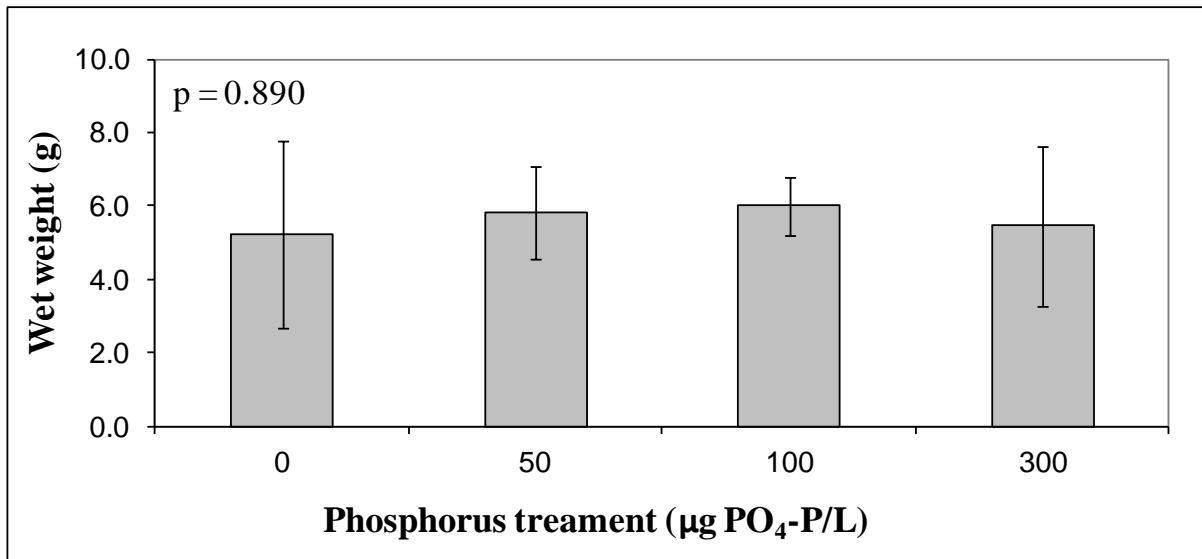
### 3.1.5 Growth and Biomass

Mean HTTL between treatments by the date the study was terminated were not statistically significant (ANOVA,  $p = 0.485$ ). The 300  $\mu\text{g PO}_4\text{-P/L}$  mean HTTL was between 43 to 55mm greater than the other treatments (Figure 30); however, greater mean average height did not translate into biomass (wet weight) where the 300  $\mu\text{g PO}_4\text{-P}$  treatment was less than the 50 and 100  $\mu\text{g PO}_4\text{-P}$  treatment (Figure 31). Mean microcosm biomass for each of treatments 0, 50, 100 and 300  $\text{PO}_4\text{-P/L}$  differed by  $< 1.0$  g and were 5.22, 5.81, 5.99 and 5.46 g respectively. The difference in biomass was not statistically significant (Kruskal Wallis,  $p = 0.890$ ). As for all figures presented in this thesis representing normally distributed data, error bars indicate standard deviations.



**Figure 30.** *Typha latifolia* mean height to tallest leaf over a twelve week period from the time of germination.

*The arrow indicates the start of additional phosphate treatment on June 19, ending July 21.*



**Figure 31.** *Typha latifolia* biomass (wet weight) after 12 weeks from germination.



### 3.1.6 Pore water

Due to an apparent lack of difference between treatments in HTTL by August 02, pore water sampling was conducted to determine whether the P additions had resulted in a difference in bioavailable P and whether P may be at concentrations potentially limiting to *T. latifolia* growth. The pore water sampling conducted August 9 and 10 incorporated an error due to approximately 2.0ml of millipore water remaining in the filter cartridges following filter flushing. On August 13, flushed filters were transferred to a separate filter cartridge and did not include a dilution error.

The mean pore water PO<sub>4</sub>-P concentrations for treatments 0, 50, 100 and 300 µg PO<sub>4</sub>-P/L treatments were 19.8, 19.8, 19.0 and 22.2 µg PO<sub>4</sub>-P/L respectively (**Figure 32**). There was no statistically significant difference between treatment pore water P (ANOVA, p = 0.841). However, the mean concentration of 37.2 µg PO<sub>4</sub>-P/L, observed in five microcosms without seedlings, one microcosm subjected to each of treatments 0, 50 and 100 µg PO<sub>4</sub>-P/L and two microcosms having received the 300 µg PO<sub>4</sub>-P/L treatment, was significantly different from fifteen microcosms from each of the four treatments (Mann Whitney, p = 0.01).

A similar result was observed on August 13 when no error due to distilled water in the filter cartridge affected results. The mean PO<sub>4</sub>-P concentrations for the microcosms containing shoots from treatments 0, 50, 100 and 300 µg PO<sub>4</sub>-P/L not sampled on August 09 or 10 was 27.3, 28.7, 30.9 and 31.3 µg PO<sub>4</sub>-P/L respectively. The PO<sub>4</sub> concentrations did not differ between the treatment groups (ANOVA, p = 0.942) (Figure 33). Whereas, the pooled PO<sub>4</sub> concentrations for eleven microcosms with plants (all treatment groups combined) were significantly different to four microcosms without plants subjected to the 0, 50 and 100 µg PO<sub>4</sub>-P/L treatments (Mann Whitney, p = 0.013). These results indicated that the addition of grow media had had no effect on pore water bioavailable PO<sub>4</sub>-P which was reflected in the similarities between treatments in biomass and HTTL.

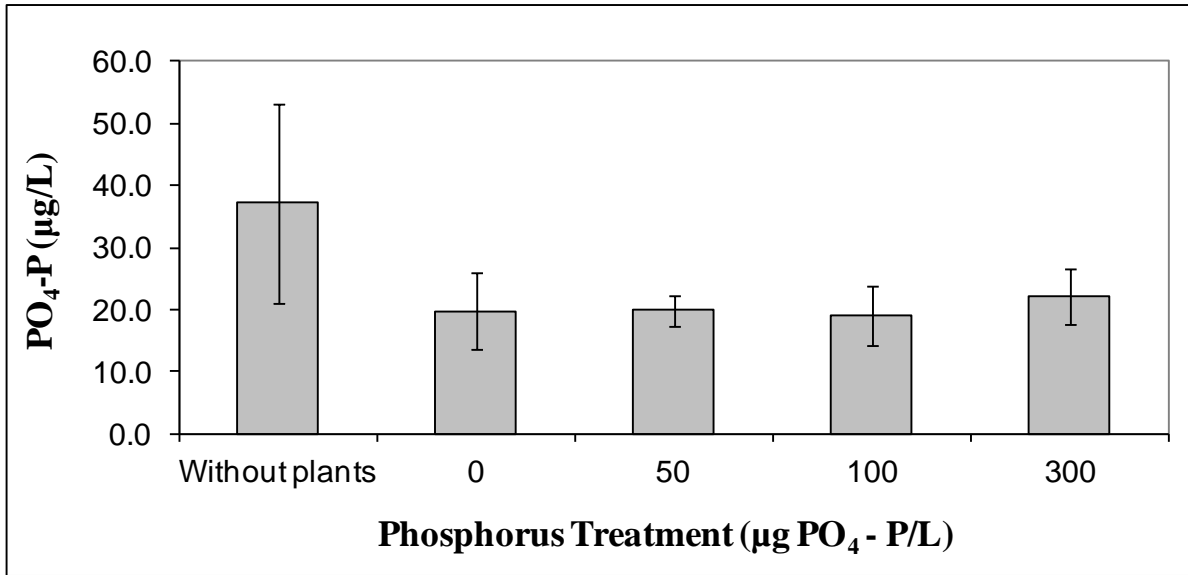


Figure 32. Microcosm pore water  $\text{PO}_4$  concentrations observed on August 09 and 10.

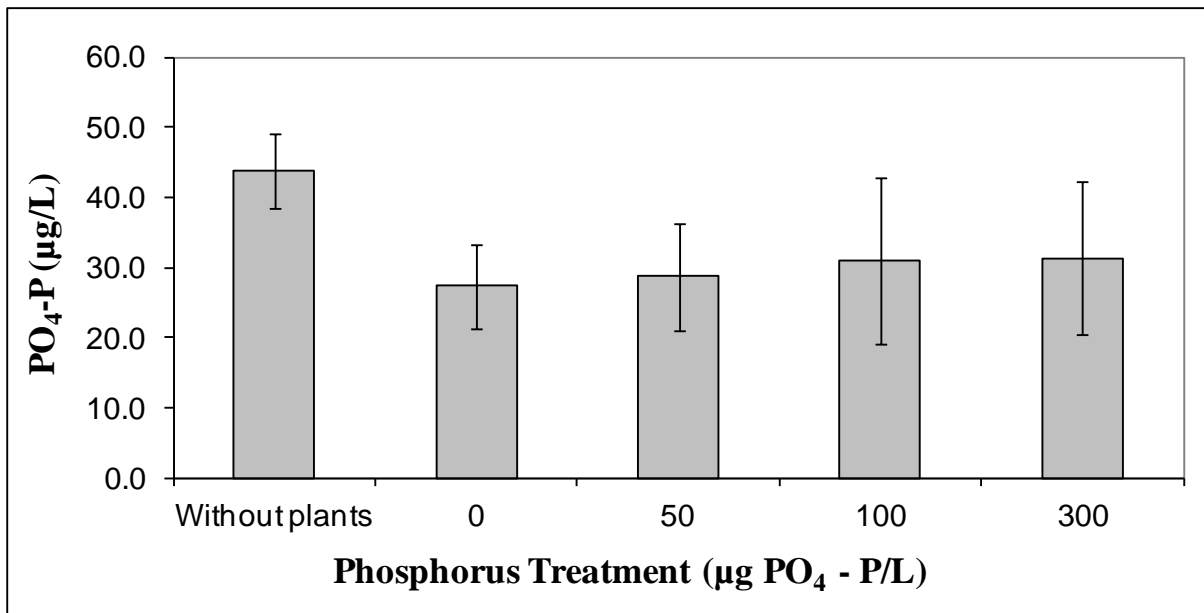


Figure 33. Pore water concentrations from microcosms sampled on August 13.

## 3.2 *Typha latifolia* Response to Oligotrophic and Eutrophic N and P Loading Rates.

### 3.2.1 Environmental Conditions

The overall median temperature throughout the November 08, 2012 to May 16, 2013 was 22.4°C with a range of 19.0°C to 27.5°C. The Accutemp sensor consistently provided very low instantaneous RH readings compared to that of the Noma RH sensor and were treated as erroneous. Mean PAR on November 09 was 67  $\mu\text{E cm}^{-2} \text{s}^{-1} \pm 20$  measured at HTTL and ranged from 23  $\mu\text{E cm}^{-2} \text{s}^{-1}$  measured at soil height to 90  $\mu\text{E cm}^{-2} \text{s}^{-1}$  measured at HTTL. Mean PAR on May 08 was 80  $\mu\text{E cm}^{-2} \text{s}^{-1} \pm 9$  and ranged from 7  $\mu\text{E cm}^{-2} \text{s}^{-1}$  measured at soil height to 118  $\mu\text{E cm}^{-2} \text{s}^{-1}$  measured at HTTL.

### 3.2.2 Basic Water Quality Parameters

The basic water quality parameters monitored monthly are provided in

Table 9. Water temperatures generally remained slightly above 20°C for the majority of the experiment with water temperatures increasing to up to 25.2°C in April and May 2012. The loading rates for the low and high N treatments from November 08 to May 10 were 2.53 g TN/m<sup>2</sup>/yr and 7.50 g TN/m<sup>2</sup>/yr respectively. The loading rates for the low and high P treatments from November 08 to May 10 were 0.03 g TP/m<sup>2</sup>/yr and 0.74 g TP/m<sup>2</sup>/yr respectively.

**Table 9. Basic water quality parameters monitored monthly for the duration of the experiment.**

Parameter	Water temperature (°C)	Dissolved oxygen (mg/L)	pH	Conductivity (μS/cm)	Salinity (mg/L)
Min	19.9	3.5	6.1	93.0	49.4
Max	25.2	11.5	8.0	895.0	453.0
Mean	22.2	8.5	6.7	508.3	248.8
S.D.	1.8	1.8	0.3	195.2	95.2

*S.D.* = standard deviation

Surface water dissolved oxygen (DO) concentrations indicated that microcosm surface water remained oxygenated throughout the experiment. Surface water pH was generally slightly acidic (mean = 6.7), a pH characteristic of natural marsh wetlands and surface flow constructed wetlands (Kadlec and Wallace, 2009). A wide range in conductivities and salinity levels were observed. The salinity level in one microcosm reached 453 mg/L which slightly exceeded the 450mg/L (0.45‰) salinity concentration defined as brackish by the EPA (EPA, 2013).

### **3.2.3 Pre-treatment Shoot Height, Total Leaf Length and Shoot Density Comparisons.**

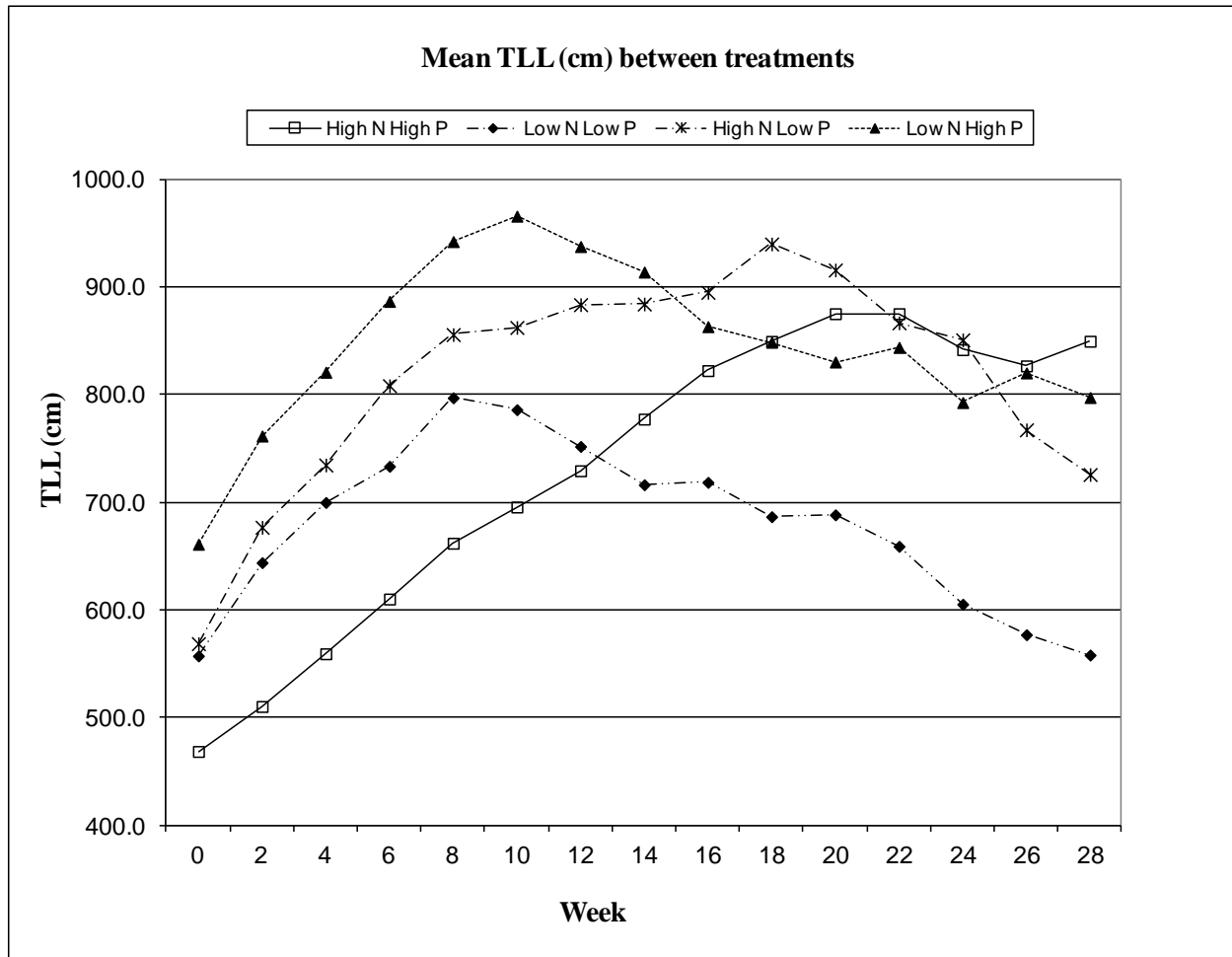
TLL, HTTL and shoot density data collected on November 07 and November 08 were used to determine whether treatments differed in TLL, HHTL or shoot densities after non-reference microcosms were randomly assigned to one of the four treatment groups. There were no statistically significant differences between mean TLL (ANOVA,  $p = 0.167$ ) or HTTL (ANOVA,  $p = 0.315$ ). Mean shoot density observed in the Low N-Low P was significantly higher than the mean shoot density observed in the High N-High P treatment (ANOVA,  $p = 0.04$ , Tukey,  $p = 0.028$ ).

### **3.2.4 Growth Expressed as Total Leaf Length in Response to N and P Treatments.**

Growth is defined here as change in TLL. All microcosms grew at a similar rate for the first eight weeks after which the Low N-Low P treatment entered into a steady decline from 796.9 cm until the end of the experiment on week 28 (Figure 34) by which time TLL was 558.0 cm, resulting in an overall negative growth of -85.8 cm from the 643.8 cm TLL observed at the start of week 2. Increase in TLL in the High N-Low P treatment was gradual after week 8 after which TLL increased to 939.5 cm over the following 10 weeks.

From week 18 onwards, the High N-Low P treatment entered into a steady decline with TLL decreasing to 725.7 cm by the week 28. After reaching a maximum TLL of 981.6 cm the Low N- High P Mean TLL entered into a decline after week 10 to 797.2 cm at which point a plateau may have been reached. The mean TLL for microcosms allocated to the High N-High P treatment increased steadily from the beginning of the experiment until week 20 when a mean

maximum TLL of 875.1cm was reached. TLL declined to 826.6 cm by April 25 followed by an increase to 849.5 cm suggesting that a possible plateau in growth had been reached. The difference in TLL between treatments was nearly statistically significant (ANOVA,  $p = 0.089$ )



**Figure 34 . Growth in response to N and P treatments expressed as change in TLL over time. N and P treatment additions began at the start of week 2.**

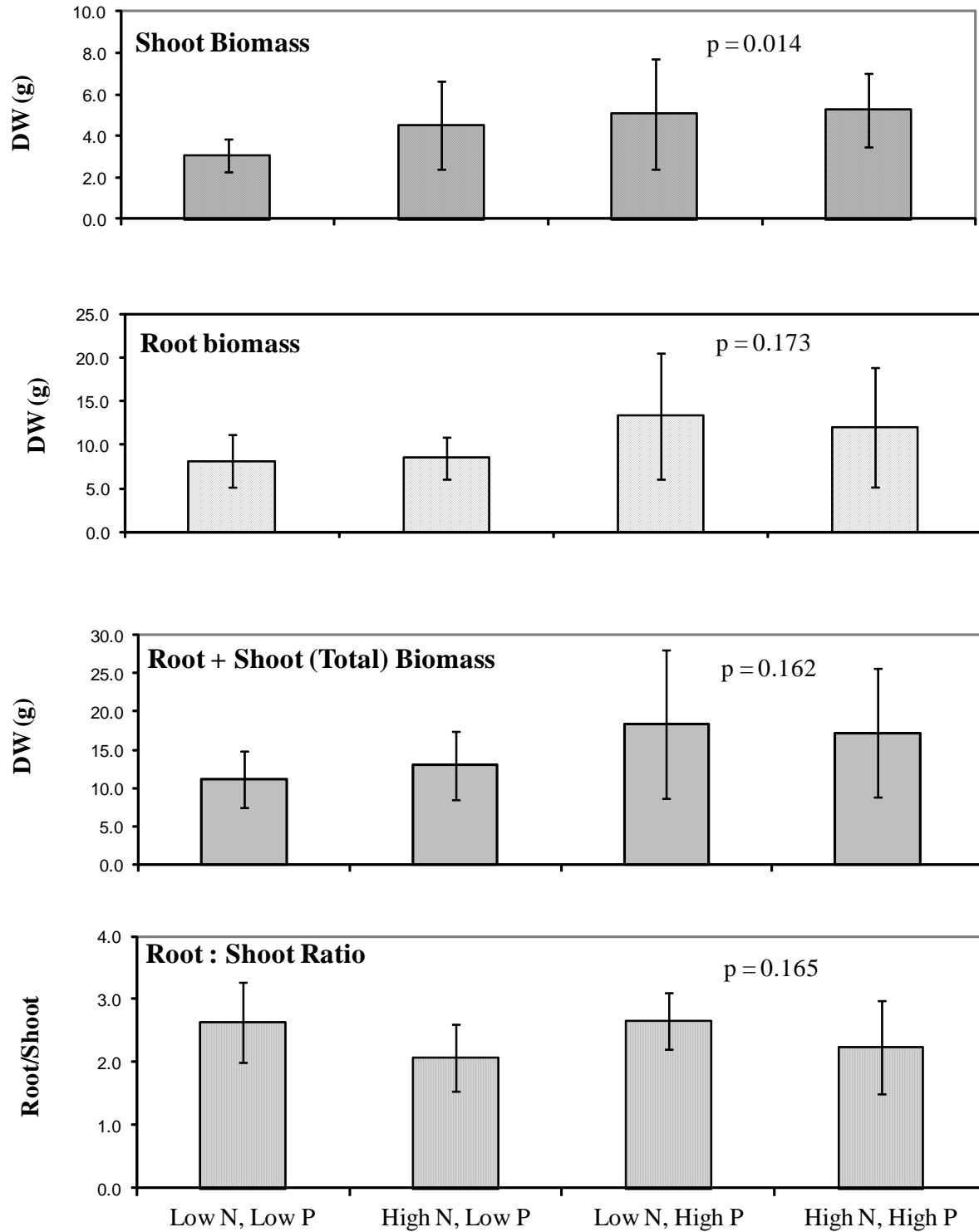
The High N-High P treatment had the greatest mean weekly growth rate at 13.0 cm TLL/week compared to the other treatments (Table 10), and 10 times the weekly growth rate observed for the Low N- High P treatment. The Low N-Low P treatment experienced a mean weekly growth rate of -3.3cm/week due to week 28 TLL being less than Week 2 TLL.

**Table 10. Mean weekly growth rate expressed as change in TLL over the course of the experiment.**

Treatment	Mean growth/week	
	TLL (cm)	S.D. (cm)
Low N, Low P	-3.3	34.1
High N, Low P	1.9	42.6
Low N, High P	1.3	38.3
High N, High P	13.0	31.3

### 3.2.5 Biomass

The Low N-Low P treatment had significantly less mean shoot biomass compared to the the Low N-High P treatment (Mann Whitney,  $p = 0.014$ ) and the High N-High P treatment (Mann Whitney,  $p = 0.001$ ) (Figure 35). Mean root biomass, mean root plus shoot (total) biomass and root to shoot ratio did not differ significantly (Figure 35). The Low N-Low P treatment mean HTTL was also significantly less than the Low N-High P treatment (ANOVA,  $p = 0.002$ ; Tukey,  $p = 0.003$ ) and the High N-High P treatment (ANOVA,  $p = 0.002$ ; Tukey,  $p = 0.015$ ). Based on the Kruskal Wallis test, there were no statistically significant differences between root biomass ( $p = 0.173$ ), root plus shoot biomass ( $p = 0.162$ ) and root : shoot ration ( $p = 0.165$ ) (Figure 35). These results indicate that *T. latifolia* growth and biomass production were primarily limited by phosphorus under the experimental conditions provided.



**Figure 35. Shoot biomass, root biomass, total biomass and root to shoot ratio observed for low and high N and P treatments.**

## 4 DISCUSSION

The primary objectives of the following thesis were to (1) establish a protocol for propagating *T. latifolia* from seed for effective hypothesis testing where individuals of equal age, fitness and genetic characteristics could be used to avoid experimental error associated with using plants of unknown origin, history, variable fitness and which could potentially introduce pests that could jeopardize study results. The majority of research reviewed involving *Typha spp.* has produced experimental units grown from rhizome (e.g. Weng *et al.*, 2006) collected from locations which may be subjected to stressors, including anthropogenic contaminants, that may influence study results. Ye *et al.* (1997), Wetzel and van der Valk (1998) and Kercher and Zedler (2004) propagated *T. latifolia* from seed for research purposes; however, complete details on watering and feeding regimens, lighting conditions such as photoperiod, light intensity; soil type if used, or soil moisture content were lacking.

The second primary objective was to (2) propagate *T. latifolia* in low nutrient sediment for use as a reference treatment to effectively test the hypothesis that *T. latifolia* survival, growth and biomass is enhanced as a result of elevated but environmentally relevant concentrations of N and P. Previous studies that examined *Typha spp.* response to elevated nutrients (e.g. Shipley and Keddy, 1988; Wetzel and van der Valk, 1998; Keddy *et al.*, 2000; Woo and Zedler, 2002) used concentrations that may be considered environmentally irrelevant and therefore are limited in terms of explaining *T. latifolia* invasions in areas where surface water and sediment nutrients may be far lower.

### 4.1 Germination

The overall percent germination of 85% observed in microcosms containing submerged or saturated sediment provided with various concentrations of PO<sub>4</sub> and the 93% germination observed in Petri dishes containing MDW were similar to the percent germination observed in preliminary trials (77 to 82%) and those reported by other authors which examined factors such as temperature, light and dissolved oxygen on *T. latifolia* germination (e.g. Morinaga, 1926;



Sifton, 1959; Yeo, 1964; Bonnewell *et al.*, 1983). There was no effect of P treatment on percent germination as was observed by Stewart *et al.* (1997).

Steps were not taken to maximize percent germination other than to minimize handling and maintain seeds in an unaltered condition to achieve as close to natural conditions as is possible in a laboratory setting. The documented optimum temperature for germinating *T. latifolia* seed under laboratory conditions differs between authors suggesting that preferred temperature may be population-dependant. Sifton (1959) reported the optimum temperature to range from 25 - 30°C whereas Bonnewell *et al.*, (1983) reported an optimum temperature of 30°C. McNaughton (1966) reported the highest percent germination occurred between 24 - 30°C and varied between seeds collected from populations ranging from the southern tip of Texas to the Canadian border. The widely fluctuating temperatures, where temperatures ranged between 22 - 32°C during the germination phase of the seedling propagation experiment involving P treatments, were within the optimum temperature range indicated by the aforementioned authors. Sifton (1959) also observed enhanced germination by alternating temperatures between 15 to 30°C. The literature therefore suggests that near optimum temperature conditions were present at the time seeds were imbibed at the beginning of the experiment and the seed viability investigation which likely contributed to the high percent germination observed.

Sifton (1959) reported a high percent germination > 80% of *T. latifolia* seeds under reduced light conditions subjected to alternating temperatures. Mean PAR measured in each of the 40 microcosms was 144  $\mu\text{E cm}^{-2} \text{s}^{-1}$  and 62  $\mu\text{E cm}^{-2} \text{s}^{-1}$  for the seed viability with overall percent germination being similar between the two groups of seeds suggesting that high germination can be achieved under comparatively low light conditions. In short, the seeds obtained from the single *T. latifolia* flower used for this thesis provided a similar germination rate reported in the literature indicating that the environment provided for seed germination was effective in achieving a percent germination sufficient for hypothesis testing.

## 4.2 A Protocol for Examining *Typha latifolia* Response to Phosphorus Loading.

The purpose of the following experiment was to test the hypothesis that *T. latifolia* post-germinated seedling growth and survival is increased by P loading at eutrophic and hypereutrophic conditions as defined by Wetzel (1983), Environment Canada (2004) and the EPA (Wikipedia, 2013a), considered here as environmentally relevant concentrations that are typically found in natural aquatic ecosystems impacted by non-point sources of P. The low N and P treatment loading rates added to microcosms for this experiment were 2.53 g TN/m<sup>2</sup>/yr and 0.03 g TPm<sup>2</sup>/yr respectively and 7.50 g TN/m<sup>2</sup>/yr and 0.74 g TP/m<sup>2</sup>/yr loading rates for the high N and P treatments respectively. The TN and TP loading rates documented by Reddy *et al.* (1993) where no impact (*T. domingensis* invasion) of nutrient loading were evident in the Florida Everglades ranged from 4.7 – 6.4 g TN/m<sup>2</sup>/yr and 0.06 TP/m<sup>2</sup>/yr whereas the loading rates of 13.6 – 16.6 g TN/m<sup>2</sup>/yr and 0.4 – 0.46 TP/m<sup>2</sup>/yr resulted in observable impact. Similarly, Reddy *et al.* (1993) observed dominance of *Cladium jamaicense* in areas of the Florida Everglades where the P loading rates ranged from 0.11 – 0.25 g TP/m<sup>2</sup>/yr whereas *T. domingensis* dominated areas where loading rates ranged from 0.54 – 1.14 g TP/m<sup>2</sup>/yr. The observed *T. latifolia* response to both the experimental high and low N and P loading rates are in close agreement to the observed distribution of *T. domingensis* in the Florida Everglades. Stewart et al (1997) noted that *T. latifolia*, believed to be a recent invader to the Florida Everglades, is restricted to areas impacted by nutrient loading. The environmentally relevant loading rates used by this experiment is therefore likely appropriate for conducting competitive experiments that more closely resemble non-point source N and P loading to determine *T. latifolia* competitiveness and potential invasiveness.

Study results did not observe an effect of PO<sub>4</sub>-P treatments added at 0, 50, 100 and 300 µg PO<sub>4</sub>-P/L with total volume added being 1300 ml on pore water P over a 10 week period. The maximum difference in pore water PO<sub>4</sub>-P between treatment mean concentrations sampled on August 09 and 10 was 3.2 µg PO<sub>4</sub>-P/L. The maximum difference between treatment mean concentrations collected on August 13 was 4.1 µg PO<sub>4</sub>-P/L. Similarly, the maximum difference between treatment mean seedling biomass was 0.77 g WW.

These results indicated that the sediment, low in P, adsorbed sufficient quantities of P to nullify the effects of each treatment. The similar pore water PO<sub>4</sub>-P concentrations in the 0 P added compared to the 50, 100 and 300 µg PO<sub>4</sub>-P/L indicated that sufficient amounts of bioavailable P were being generated internally through microbial decomposition of organic matter which comprised 80% of the 200 g of top soil added (Presidents Choice, personal communication). The total weight of P added to the 50, 100 and 300 µg PO<sub>4</sub>-P/L treatments in the total 1300 ml of grow media added was equivalent to 0, 65, 130 and 390 µg P/kg of sediment. P adsorption to sand and other types of sediments has been well documented (*e.g.* Richardson, 1985; Danen-Louwerse *et al.*, 1993; White *et al.*, 2000; Vymazal, 2007; Vohla *et al.*, 2011) and increasing the frequency of grow media addition from alternating days to daily additions from the start of the experiment may have resulted in a treatment effect. However, considering that sediment TP in Cootes Paradise can reach nearly 2.0 g TP/kg of sediment (Mayer *et al.*, 2008) with a mean surface water concentration averaging 210 µg TP/L (Court and Bowman, 2010), the amount P needed in order to have an observable effect would have been too high to be considered environmentally relevant and therefore counter to study objectives.

Furthermore, it cannot be determined whether the *T. latifolia* seedlings were limited by P. The significantly higher PO<sub>4</sub> levels in the pore water of microcosms without seedlings compared to those with seedlings indicates that seedlings were removing P at a sufficiently high enough rate to have had a measurable effect on sediment P concentrations. Given that > 20 µg PO<sub>4</sub>-P/L was detected in the pore water sediment affected by plant uptake may suggest that P was not limiting. Furthermore, some Low N – Low P *T. latifolia* genets grew to 1351.8cm TLL over a three month period from September 15 through to December 20. The addition of greater amounts of P may therefore not have resulted in greater post-germination seedling growth. By surface water quality standards, the 300 µg PO<sub>4</sub>-P treatment is considered very high based on criteria presented in Wetzel (1983) and three times the concentration classified as hypereutrophic by Environment Canada (2004). The P requirements of *T. latifolia* seedlings may be sufficiently low to enable early survival and possibly recruit to adult stages in P- poor sediment. *T. latifolia* can occur in isolated stands in what are generally considered to be nutrient-poor habitats (*e.g.*, City of Toronto and the Upper Thames Conservation Authority, 2009) as can other *Typha spp* where they may be restricted to isolated stands (Newman *et al.*, 1996). The presence of *T.*

*latifolia* in pristine wetlands within the circumpolar boreal regions (Bourgeois *et al.* 2012) suggests that *T. latifolia* can establish in low nutrient wetland habitats. However, the presence of mesotrophic to eutrophic levels of P in porewater observed for the seedling response to P experiment including treatments that did not receive P additions suggests that sediment organic matter content and its rate of mineralization by microorganisms is an important factor in *T. latifolia* recruitment, establishment, distribution and invasiveness. Further research is needed to the significance of sediment and pore water P on *T. latifolia* distribution and invasiveness.

The ability to propagate *T. latifolia* from seed in sediment low in P would enable future competition studies that could compare the competitiveness of *T. latifolia* against other native species at various P concentrations to determine a P threshold above which *T. latifolia* can become invasive. The broadly distributed but often rare species from the Class Isoetes are adapted to low nutrient conditions (Smolders *et al.*, 2002) and may be considered as a candidate species in competition experiments. *Schoenoplectus spp.* are not considered to be adapted to low nutrient conditions but have proven to be effective in constructed wetlands (*e.g.* Tanner, 2001; Nelson and Thullen, 2008; Kadlec and Wallace 2009; Lee *et al.* 2009; Daniels *et al.*, 2010; Pollard, 2010). Neil and Graham (1989) observed greater N and P removal rates by wetland microcosms planted with *S. tabernaemontani* compared to wetlands planted with *T. angustifolia*. Efforts to propagate sufficient numbers of *Schoenoplectus acutus* and *Schoenoplectus tabernaemontani* from seed were unsuccessful. Continued research into developing methods for germinating noninvasive native species is recommended.

No data was found in the literature with which to compare the observed overall 53.5 % seedling mortality rate, the cause of which is unknown. Qualitative observations of the surficial sediments which at times appeared dry despite registering a moist to average wet indicated a possible water limitation especially for newly germinated seedlings with limited root penetration. Wetzel and van der Valk (1998) stated that only a few seedlings died after two weeks from the time of germination. Also, the observed mortality rate was higher than compared to one preliminary propagation trial (Propagation Trial 2, Appendix E) which observed a mortality rate of 20%. Propagation Trial 2 applied previously germinated seeds onto high nutrient Scots Garden Essentials topsoil in seed trays where soil was kept moist by placing seed trays into a reservoir containing up to 1.0cm of MDW.

The MDW used throughout the first experiment contained chloramine which ranged from 0.29mg/L to 0.93mg/L (Jorge Loyo, personal communication). MDW analysis carried out by a private lab indicated elevated Cu. Toxicity trials of unfiltered MDW VS activated carbon filtered MDW resulted in increased mortality for *Daphnia magna* and *Hyaella azteca* (M. Raby, personal communication) in unfiltered water to which chloramine and Cu may have been contributing factors. *T. latifolia* has been shown have constitutional tolerance (inherent tolerance regardless of population or exposure history) to zinc (Zn), cadmium (Cd) and lead (Pb) at high concentrations (McNaughton *et al.*, 1974; Ye *et al.*, 1997). Tolerance of *T. latifolia* adults and clones to Cu has also been observed by Taylor and Crowder (1984) and Maddison *et al.* (2005). The effect of elevated Cu levels on newly germinated *T. latifolia* seedlings is unknown and may have contributed to seedling mortality.

Data on post-germination seedling growth rates were also not found in the literature. Wetzel and van der Valk (1998) began measuring seedling growth determined by HTTL and number of leaves three weeks after germination in interspecific competition experiments in soil containing 5 to 20mg of bioavailable P/g of soil and 7mg/g nitrate and at similar light intensities of up to  $420\mu\text{E m}^{-2} \text{ s}^{-1}$ . The overall mean HTTL observed in this experiment was 25.2cm after twelve weeks which were highly comparable to the HTTL illustrated in the graphs presented in Wetzel and van der Valk (1998) where heights of approximately 30cm or less were observed depending on the degree of competition with other species. If competition had a negligible effect on seedlings, it would suggest that nutrient limitation may not have occurred over the first twelve week period. Wetzel and van der Valk (1998) did however observe a noticeable difference in dry weight biomass at 18 weeks. The very high production of biomass observed by Yeo (1964) where 34 aerial shoots ranging in 46 to 122 cm was observed further suggests that seedlings were nutrient limited.

#### **4.3 *Typha latifolia* Response to Oligotrophic and Eutrophic N and P Loading Rates.**

The results of the experiment conducted to determine the relative effects of N and P at oligotrophic and/or eutrophic to hypereutrophic loading rates observed significantly higher shoot

biomass and HTTL in the Low N- High P treatment and the High N – High P treatment compared to the oligotrophic Low N - Low P supporting the hypothesis that *T. latifolia* biomass increases significantly under elevated yet environmentally relevant P loading rates. The High N – Low P loading rate treatment was intermediate in biomass and did not differ significantly between either. N was therefore a contributor to biomass suggesting N and P co-limitation at the time the experiment was terminated. These results are in agreement with Woo and Zedler (2002) who observed N and P co-limitation in *T. x glauca* reared under green house conditions. However, the High N – low P treatment (Figure 35) had exhibited a downward linear trend since week 18 at a rate more rapid than that of the Low N – low P treatment. Had the experiment ran longer and this trend continued, the biomass would have decreased to levels significantly different to that of the high P treatments, assuming that the high P treatments had stabilized. Based on these results, *T. latifolia* appears to have been primarily P-limited as was observed by Boyd and Hess (1970).

Due to the 16L: 8D photoperiod, *T. latifolia* growth over nearly a twelve month period was achieved since the late summer translocation of carbohydrates to the rhizome and the subsequent overwintering dormancy period that occurs under natural conditions (Garver *et al.*, 1988) did not occur. The decline in the Low N –Low P and the High N – Low P treatment raises the question as to whether *T. latifolia* could have persisted under low N and P conditions given that the majority of carbohydrate storage is consumed during the overwintering period with the remainder being used to generate new shoots the following spring. How the lack of a wintering period affected results is unknown but it could be assumed that a greater amount of biomass and carbohydrate storage was able to accumulate under experimental controlled conditions than would have occurred under natural conditions.

*T. latifolia* biomass production and shoot height under artificial light relative to natural light is unknown. The mean shoot-to- root ratios for all treatments (similar to root: shoot ratios provided in the Results section) was 0.45 and ranged from 0.22 – 0.77. Reddy and Portier (1987) considered shoot: root ratios between 0.72 – 0.82 to be low and, following N fertilization and increased temperatures to 25°C, observed shoot: root ratios of 1.83 – 3.04. Shoot production under experimental conditions was therefore relatively low and may be a reflection of reduced light intensities. However, Grace and Wetzel (1981a) observed greater heights and leaf volume

in a natural *T. latifolia* stand growing under a canopy compared a *T. latifolia* stand occurring in an open marsh. The genets growing under canopy did not produce flowers, as was observed in this experiment, whereas genets growing in an open marsh habitat produced flowers leading Grace and Wetzel (1981a) to hypothesize that the shoots, growing under canopy, were light – limited and therefore allocated resources entirely to shoot production to achieve maximum height. *T. latifolia* are considered to be shade tolerant (Grace and Wetzel, 1981a) and the use of artificial light may not significantly reduce biomass production but may inhibit the production of culms and inflorescence. Newly emerging ramets were observed throughout the experiment under light conditions as low as  $7 \mu\text{E cm}^{-2} \text{s}^{-1}$ . The high germination rate observed at a mean PAR of  $62 \mu\text{E cm}^{-2} \text{s}^{-1}$  suggests that experiments carried out under artificial light can produce environmentally relevant results.

The observed shoot densities, which ranged from 53 to 481 shoots /m<sup>2</sup>, was considerably higher than 20 – 40 shoot/m<sup>2</sup> shoot densities typically reported in the literature (*e.g.* Walton *et al.*, 1990). This difference is likely due to methodology where stem density counted under field conditions is likely based on the number of mature or visible shoots with newly emerging shoots either being discounted or, if submerged or hidden by leaf litter, not detected. However, one shoot placed into one microcosm (area = 0.01873/m<sup>2</sup>) is equivalent to 53 shoots/m<sup>2</sup>. Vessel size may therefore have restricted growth and therefore the ability to fully take advantage of the available N and P. Increasing vessel size may produce a greater response to N and P loading but at the cost of the number of replicates and therefore reduced statistical power.

#### **4.4 Recommendations for Further Study**

For future development of *T. latifolia* and other wetland species propagation protocols where high survival rate is required, a rigorous evaluation of newly germinated seedling mortality and sediment moisture conditions is recommended. Given that nutrient additions were unlikely limiting, the drying of surficial sediments may contribute to seedling desiccation . The soil moisture probe was not effective in sediments above 2.0 cm and so sediment moisture levels may not have been an accurate reflection of sediment conditions for newly germinated seedlings.

It is recommended that competition studies be conducted to determine the N and P levels at which N and P loading rates result in *T. latifolia* dominance. The protocol used a low P

formulated sediment which is necessary to establish a low nutrient, pre-impact reference. To provide a suitable high N and/or P comparison, the same sediment should be treated with sufficient levels of P prior to the addition of seedlings given that added N and P at environmentally relevant levels once seedlings are planted would not be possible due to the risk mortality caused by drowning.

The inability to germinate *Schoenoplectus acutus* and *S. tabernaemontani* indicates that established protocols for germinating a variety of native species are needed. Species that are light sensitive and which may require natural light could be germinated in cold frames (wooden boxes with a glass lid) which can easily be built to a size according to available space to establish germination protocols that will provide a sufficient number of individuals required for effective hypothesis testing.

It is further recommended that the presence and distribution of *T. latifolia* and other *Typha* spp. occurring in low nutrient wetlands and how their distributions relate to spacial differences in sediment organic matter and nutrient content to determine of isolated *Typha* stands in low nutrient wetlands are restricted to microhabitats of higher nutrient availability. How *Typha* distribution can be predicted by more easily determined surface water nutrient levels as compared to sediment and pore water nutrient levels would (i) determine the usefulness of surface water nutrients in predicting *T. latifolia* invasion and success of recovery efforts and (ii) whether pore water as observed by Angeloni et al. (2006) and sediment P may be better predictors of *T. latifolia* invasion.

## 4.5 Summary

The objectives of the thesis were as follows:

1. Develop a protocol for rearing *Typha latifolia* from seed to adult stages under laboratory conditions to enable hypothesis testing based on data collected from plants of known history, uniform age and comparable fitness;



2. Conduct controlled, rigorous experiments with sufficient replication to further existing knowledge on the effects of environmentally relevant N and P loading rates on *T. latifolia* survival, growth and biomass;

The sub-objectives were as follows:

1. Determine the feasibility of conducting future competition and nutrient removal experiments between *T. latifolia* and the native non-invasive species such as *Schoenoplectus acutus* and *Schoenoplectus tabernaemontani* under laboratory conditions;
2. Based on literature review, present supporting evidence for the hypothesis that native, non-invasive wetland macrophytes would provide N and P removal rates comparable to currently used invasive species and raise the issue of invasive species use in constructed wetlands to encourage research to determine the feasibility of utilizing non-invasive species in constructed wetlands for N and P removal.

#### **4.5.1 *Typha latifolia* Nitrogen and Phosphorus Limitation and Constructed Wetland Use for Ecological Protection and Restoration.**

The significant difference between the shoot biomass and maximum leaf height of microcosms exposed to oligotrophic P loading rates compared to eutrophic to hypereutrophic P loading rates supports the hypothesis that environmentally relevant levels of P, far lower than have previously been assessed by previous studies, is supported. Increased biomass and leaf height may therefore enable *T. latifolia* to outcompete less competitive species or species adapted to low nutrient conditions under nutrient loading rates commonly observed in nonpoint source agricultural and stormwater runoff

Other authors have identified increased and stabilized water levels as a cause for *T. latifolia* invasions (*e.g.* Kercher and Zedler, 2004; Wilcox *et al.*, 1985; Wilcox *et al.*, 2008). Smolders *et al.* (2006) indicated that anoxic sediments created as a result of inundation can lead to the

reduction of Fe-bound P and the release of bioavailable P. Artificially high water levels and P availability may therefore be inextricably linked with a potentially synergistic effect that contributes to *Typha* invasions.

The effect of eutrophic N loading rates on biomass was not significant. However, other researchers have observed N limitation in *T. latifolia* (e.g. Grace and Wetzel, 1981a; Reddy and Portier, 1987; Grace, 1988). The management of *T. latifolia* and other *Typha* species must therefore consider both N and P.

#### **4.5.2 The Use of Constructed Wetlands for the Protection and Restoration of Aquatic Ecosystems**

The effectiveness of constructed wetlands for the restoration of eutrophicated aquatic ecosystems appears highly limited based on the available literature. A careful mass balance estimate should be conducted, including N and P burial rate and denitrification, in order to evaluate whether the amount of N and P removed would have a beneficial biological response in terms of reduced algal biomass and increased water clarity. A risk assessment should also be conducted to determine the potential impacts of new developments involving the release of nutrients and subsequent treatment by constructed wetlands to determine (1) if a proposed constructed wetland will result in a biologically significant reduction in N and P based on changes in chlorophyll *a* and algal biomass and (2) whether significant impact downstream as a result of ecologically impacting levels of N and P in effluent will occur. The use of *T. latifolia* and other species known to be invasive and commonly used in constructed wetlands under high nutrient loading rates would impose a significant environmental risk in watersheds that have not been impacted by invasive species. Furthermore, the false notion that wetlands are highly effective at removing N and P with the currently available constructed wetland technology needs to be addressed through a greater transfer of knowledge and appreciation of natural ecosystems and the ecological and socioeconomic impacts invasive species can incur. An objective evaluation of existing nutrient management programs and strategies should be carried out to identify potential opportunities for improving nutrient management before considering constructed wetlands for nutrient removal.



## 5 APPENDIX

*Raw data not included in this Appendix Section is available upon special request, pending purpose and approval from the author of this thesis and the authors' supervisor.*

### 5.1 Appendix A

#### Batch 1 Sediment

**Table A1**

Date in	Time in	Date out	Time out
May 02,12	1650	May 03,12	1700

Note: tests conducted on May 03, 12 at 1300 and 1900 indicated constant weight (total moisture removal) at 1300 hrs

#### **May 02 and May 03, 2012. PCBE seived.**

Container Weight (g)	Container Weight + PCBE (g)	Weight + PCBE (g)	Container Weight + PCBE (g)	PCBE DW (g)	Moisture Content (g)	Moisture (%)
58.58	68.94	10.36	62.64	4.06	6.30	60.81
65.64	76.01	10.37	69.74	4.10	6.27	60.46
60.31	70.35	10.04	64.29	3.98	6.06	60.36
58.56	68.49	9.93	62.50	3.94	5.99	60.32
58.71	69.28	10.57	62.95	4.24	6.33	59.89
60.03	70.31	10.28	64.15	4.12	6.16	59.92
62.49	72.46	9.97	66.46	3.97	6.00	60.18
69.07	79.12	10.05	73.09	4.02	6.03	60.00
63.44	73.41	9.97	67.42	3.98	5.99	60.08
68.57	78.57	10.00	72.56	3.99	6.01	60.10
Mean		10.15		4.04		60.21
Std dev		0.22		0.09		0.28

## Batch 2 Sediment

**Table A3**

Date in	Time in	Date out	Time out (1st measurement)	Date out	Time out (2nd measurement)
May 17,12	2258	May 18,12	1301	May 18,12	1530

**May 17 and May 18, 2012. PCBE unseived. Sediment mixed somewhat in bag by rotating**

Glass Beaker Weight (g)	Glass Beaker Weight + sand/PCBE WW (g)	Beaker Weight + PCBE WW (g)	Beaker Weight + PCBE DW (g)		PCBE DW (g)	Moisture Content (g)	Moisture (%)
			(1st measurement)	(2nd measurement)			
27.7	33.32	5.62	29.84	29.85	2.14	3.48	61.92
28.54	33.51	4.97	30.39	30.39	1.85	3.12	62.78
26.79	32.51	5.72	28.92	28.92	2.13	3.59	62.76
30.53	35.44	4.91	32.37	32.37	1.84	3.07	62.53
28.5	33.77	5.27	30.46	30.45	1.96	3.31	62.81
<b>Mean</b>		<b>5.30</b>	<b>30.40</b>	<b>30.40</b>	<b>1.98</b>	<b>3.31</b>	<b>62.56</b>
<b>Stdev</b>		<b>0.37</b>	<b>1.26</b>	<b>1.26</b>	<b>0.15</b>	<b>0.22</b>	<b>0.37</b>

## Batch 2, 3 Sediment

**Table A4**

Date in	Time in	Date out	Time out (1st measurement)	Date out	Time out (2nd measurement)
May 18,12	1615	May 18,12	2200	May 19,12	1230

**May 18 and May 19, 2012. PCBE seived. Topsoil mixed in mixed bucket prior to sampling.**

Glass Beaker #	Glass Beaker Weight (g)	Beaker Weight + sand/PCBE WW (g)	PCBE WW (g)	Beaker Weight + PCBE DW (g)		PCBE DW (g)	Moisture Content (g)	Moisture (%)
				(1st measurement)	(2nd measurement)			
1	27.7	32.46	4.76	29.54	29.49	1.79	2.97	62.39
2	28.54	33.25	4.71	30.36	30.33	1.79	2.92	62.00
3	26.79	32.12	5.33	29.19	28.79	2	3.33	62.48
7	30.53	35.6	5.07	32.62	32.44	1.91	3.16	62.33
9	28.5	34.11	5.61	30.95	30.62	2.12	3.49	62.21
	<b>Mean</b>		<b>5.10</b>	<b>30.53</b>	<b>30.33</b>	<b>1.92</b>	<b>3.17</b>	<b>62.28</b>
	<b>Stdev</b>		<b>0.38</b>	<b>1.36</b>	<b>1.38</b>	<b>0.14</b>	<b>0.24</b>	<b>0.19</b>

## Batch 4

**Table A5**

Date in	Time in	Date out	Time out (1st measurement)	Date out	Time out (2nd measurement)
May 19,12	2138	May 20,12	1123	May 20,12	1421

**May 19 and May 20, 2012. PCBE sieved. Topsoil mixed in mixed bucket prior to sampling.**

Beaker #	Glass Beaker Weight (g)	Beaker Weight + mixed sand/PCBE WW (g)	PCBE WW (g)	Beaker Weight + PCBE DW (g) (1st measurement)	Beaker Weight + PCBE DW (g) (2nd measurement)	PCBE DW (g)	Moisture Content (g)	Moisture (%)
1	27.7	32.74	5.04	29.62	29.62	1.92	3.12	61.90
2	28.55	34.05	5.5	30.62	30.63	2.08	3.42	62.18
3	26.79	33.45	6.66	29.33	29.33	2.54	4.12	61.86
7	30.53	36.74	6.21	29.9 <sup>a</sup>	32.91	2.38	3.83	61.67
9	28.5	34.3	5.8	30.68	30.69	2.19	3.61	62.24
<b>Mean</b>			<b>5.84</b>			<b>2.22</b>	<b>3.62</b>	<b>61.97</b>
<b>Stdev</b>			<b>0.63</b>			<b>0.24</b>	<b>0.38</b>	<b>0.24</b>

**Batch 5 (1st Bag of soil sieved May 03 and stored in bucket. Original moisture was 60.21%)**

**Table A6**

Date in	Time in	Date out	Time out (1st measurement)	Date out	Time out (2nd measurement)
May 20,12	1200	May 20,12	1918	May 20,12	2040

**May 20, 2012. PCBE sieved. Topsoil mixed in mixed bucket prior to sampling. Data below from stored in a bucket soil sieved May 03**

Beaker #	Glass Beaker Weight (g)	Beaker Weight + mixed sand/PCBE WW (g)	PCBE WW (g)	Beaker Weight + PCBE DW (g) (1st measurement)	Beaker Weight + PCBE DW (g) (2nd measurement)	Beaker Weight + PCBE DW (g) (3rd measurement)	PCBE DW (g) (4th measurement)	Moisture Content (g)	Moisture (%)	Moisture Content (2nd obs.)(g)	Moisture (%) (2nd obs.)	Moisture Content (3rd obs.)(g)	Moisture (%) (3rd obs.)	Moisture Content (4th rd obs.)(g)	Moisture (%) (4th obs.)
10	28.87	34.36	5.49	31.14	31.11	31.09	31.07	2.27	3.22	58.65	3.25	59.20	0.55	3.27	59.56
11	37.01	42.21	5.2	39.14	39.11	39.10	39.09	2.13	3.07	59.04	3.1	59.62	0.58	3.11	59.81
12	29.65	35.68	6.03	32.25	32.12	32.07	32.07	2.6	3.43	56.88	3.56	59.04	2.16	3.61	59.87
15	30.51	36.54	6.03	33.09	33	32.96	32.95	2.58	3.45	57.21	3.54	58.71	1.49	3.58	59.37
17	31.45	37.41	5.96	34.01	33.9	33.86	33.85	2.56	3.4	57.05	3.51	58.89	1.85	3.55	59.56
<b>Mean</b>			<b>5.74</b>					<b>2.43</b>	<b>3.31</b>	<b>57.7667</b>	<b>3.39</b>	<b>59.0902</b>	<b>1.3235</b>	<b>3.42</b>	<b>59.6343</b>
<b>Stdev</b>			<b>0.38</b>					<b>0.21</b>	<b>0.16</b>	<b>1.0009</b>	<b>0.21</b>	<b>0.3452</b>	<b>0.22</b>	<b>0.2027</b>	<b>0.22</b>

## Batch 6 ( Bag 2 soil only).

**Table A7**

Date in	Time in	Date out	Time out (1st measurement)	Date out	Time out (2nd measurement)
May 20,12	1200	May 20,12		May 22,12	1058hrs

May 20, 2012. PCBE sieved. Topsoil mixed in mixed bucket prior to sampling. Data below from stored in a bucket soil sieved May 03

Beaker #	Glass Beaker Weight (g)	Beaker Weight + mixed sand/PCBE WW (g)	PCBE WW (g)	Beaker Weight + PCBE DW (g) (1st measurement)	Beaker Weight + PCBE DW (g) (2nd measurement)	PCBE DW (g)	Moisture Content (g)	Moisture (%)	Moisture Content (2nd obs.)(g)	Moisture (%) (2nd obs.)	Moisture (%) (2nd obs. -1st obs.)
1	27.7	32.68	4.98	29.5	29.51	1.8	3.18	63.86	3.17	63.65	-0.20
2	28.55	33.23	4.68	30.22	30.24	1.67	3.01	64.32	2.99	63.89	-0.43
3	26.79	32.31	5.52	28.78	28.79	1.99	3.53	63.95	3.52	63.77	-0.18
7	30.53	35.35	4.82	32.27	32.29	1.74	3.08	63.90	3.06	63.49	-0.41
9	28.5	34.14	5.64	30.52	30.54	2.02	3.62	64.18	3.6	63.83	-0.35
<b>Mean</b>			<b>5.13</b>			<b>1.84</b>	<b>3.28</b>	<b>64.0411</b>	<b>3.27</b>	<b>63.7254</b>	-0.3158
<b>Stdev</b>			<b>0.43</b>			<b>0.15</b>	<b>0.27</b>	<b>0.1993</b>	<b>0.28</b>	<b>0.1597</b>	

## Batch 7 ( Bag 3 soil only).

Date in	Time in	Date out	Time out (1st measurement)	Date out	Time out (2nd measurement)
May 23,12	105	May 23,12	1115	May 23,12	

**Table A8**

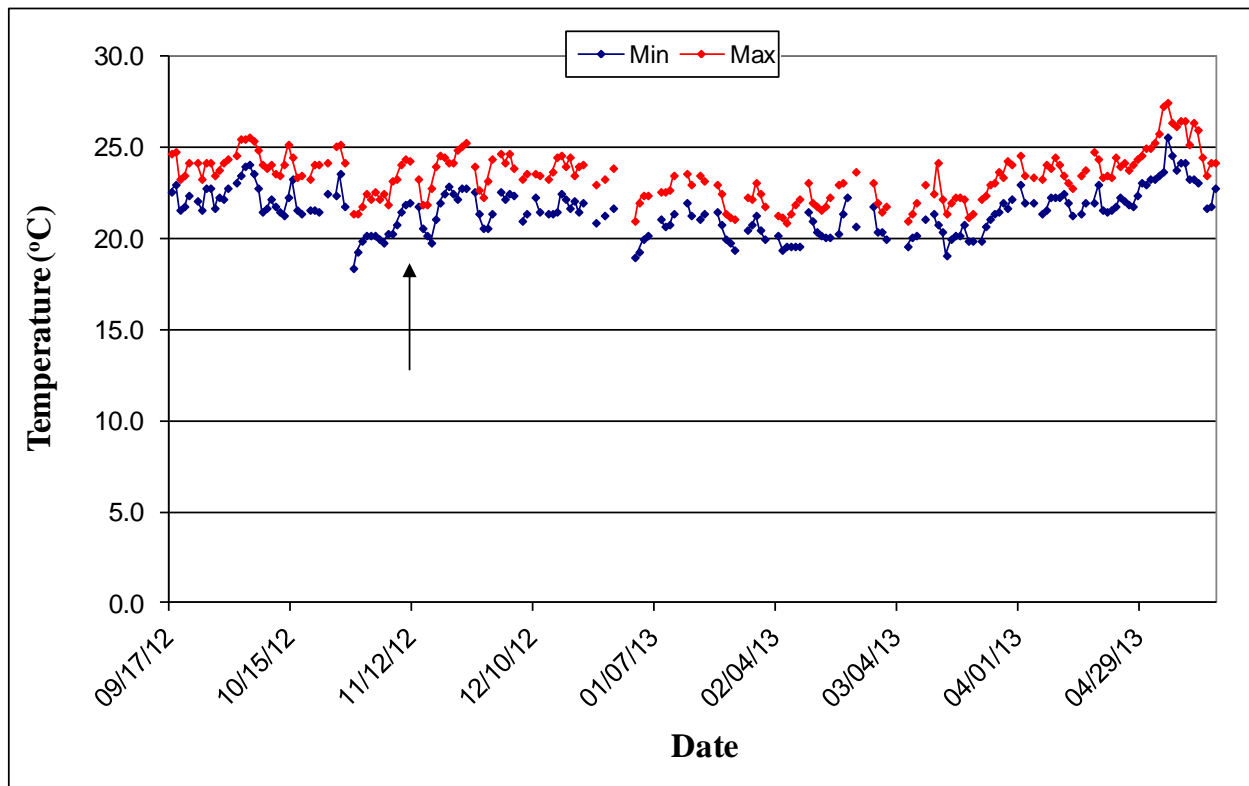
May 23, 2012. PCBE sieved. Topsoil mixed in mixed bucket prior to sampling. Data below from stored in a bucket soil sieved May 03

Beaker #	Glass Beaker Weight (g)	Beaker Weight + mixed sand/PCBE WW (g)	PCBE WW (g)	Beaker Weight + PCBE DW (g) (1st measurement)	Beaker Weight + PCBE DW (g) (2nd measurement)	PCBE DW (g)	Moisture Content (g)	Moisture (%)	Moisture Content (2nd obs.)(g)	Moisture (%) (2nd obs.)	Moisture (%) (2nd obs. -1st obs.)
1	27.7	31.99	4.29	29.28	29.27	1.58	2.71	63.17	2.71	63.17	0.00
2	28.55	33.74	5.19	30.48	30.47	1.93	3.26	62.81	3.26	62.81	0.00
3	26.79	31.73	4.94	28.6	28.6	1.81	3.13	63.36	3.13	63.36	0.00
7	30.53	34.48	3.95	31.99	31.99	1.46	2.49	63.04	2.49	63.04	0.00
9	28.5	33.68	5.18	30.42	31.41	1.92	3.26	62.93	3.26	62.93	0.00
<b>Mean</b>			<b>4.71</b>		<b>30.35</b>	<b>1.74</b>	<b>2.97</b>	<b>63.0632</b>	<b>2.97</b>	<b>63.0632</b>	0.0000
<b>Stdev</b>			<b>0.56</b>		<b>1.42</b>	<b>0.21</b>	<b>0.35</b>	<b>0.2118</b>	<b>0.35</b>	<b>0.2118</b>	

## Table A9. Sediment Percent Moisture Summary Table

Sediment moisture content for Batches 1 to 7

Batch	Mean moisture content (%)	n	s
1	60.21	10.00	0.28
2	62.56	5.00	0.37
2,3	62.28	5.00	0.19
4	61.97	5.00	0.24
5	59.81	5.00	0.18
6	63.73	5.00	0.16
7	63.06	5.00	0.21
<b>Mean</b>	61.95		0.23



**Figure B1.** Temperature data from light bank 3 for the acclimation and entire study period from November 08, 2012, indicated by the arrow, to May 16, 2013.



## 5.2 Appendix B

*Typha latifolia* protocol development and Seedling Response To Phosphorus

Table B.1 Total P added (approximate)

Total Votal volume	P Treatment (µg/L)			
1.3	0	50	100	300
Total P added	0	65	130	390

Table B2. *Typha latifolia* grow media reagents modified from Weng *et al.*, 2006.

**Table B2. Reagent molar mass for Chlorella and Typha grow medium culture**

Molar mass obtained from: Mortimer. CE. 1971. Chemistry, a conceptual approach. 2nd edition. Litton Educational Publishing Inc.

Element	g/mol	Reagent	Media/Source	Molar mass (g)	Weight needed (g)	Molarity (mol/L)	% active ingredient	Weight of Active Ingredient (g)
H	1.008	MnSO <sub>4</sub>		150.9956				
N	14.0067	MnSO <sub>4</sub> *H <sub>2</sub> O	Alternative	169.011	0.098110505	0.000580498	0.893406938	0.087652605
O	15.9994	MnSO <sub>4</sub> *6H <sub>2</sub> O	Weng et al.	259.088	0.1504	0.000580498	0.582796579	0.087652605
P	30.9738	CaCl <sub>2</sub>	Alternative	110.986	0.193259655	0.001741298	1	0.193259655
S	32.06	CaCl <sub>2</sub> *2H <sub>2</sub> O	Weng et al.	147.0168	0.256	0.001741298	0.754920526	0.193259655
Cl	35.453	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	Alternative	132.135	5.239497093	0.039652606	1	5.239497093
Ca	40.08	(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub> *7H <sub>2</sub> O	Weng et al.	258.2428	10.24	0.039652606	0.511669638	5.239497093
Fe	55.847	MgSO <sub>4</sub> *7H <sub>2</sub> O	Weng et al.	246.4704	2.56	0.010386643		
Mg	24.305	FeCl <sub>3</sub> *6H <sub>2</sub> O	Weng et al.	270.2984	0.0128	4.73551E-05		
Mn	54.938	KH <sub>2</sub> PO <sub>4</sub>	Weng et al.	136.0894	2.193	0.016114407		
K	39.102	P		30.9738	0.499124424	0.016114407	0.227598917	0.499124424
Na	22.9898							
H <sub>2</sub> O	18.0154							

22.76% of weight KH<sub>2</sub>PO<sub>4</sub> is P or 0.499g of 2.193g KH<sub>2</sub>PO<sub>4</sub> is P

**Table B3. Typha stock solution per 1000ml of Distilled water**

Reagent	Weight (g)	Molarity (Mol/L)	Volume (ml)	µg/L	Dilution volume to 200ug/L (L)	Comments
MnSO <sub>4</sub> *H <sub>2</sub> O	0.098	0.000580498	1000			Alternative
CaCl <sub>2</sub>	0.193	0.001741298	1000			Alternative
CaCl <sub>2</sub> *2H <sub>2</sub> O	0.256	0.001741298	1000			
(NH <sub>4</sub> ) <sub>2</sub> SO <sub>4</sub>	5.239	0.039652606	1000			Alternative
MgSO <sub>4</sub> *7H <sub>2</sub> O	2.56	0.010386643	1000			
FeCl <sub>3</sub> *6H <sub>2</sub> O	0.0128	4.73551E-05	1000			
KH <sub>2</sub> PO <sub>4</sub>	2.193	0.016114407	1000			0.499g is P
P in KH <sub>2</sub> PO <sub>4</sub>	0.499124424	0.016114407	1000	499124.4241	2495.62212	
Total weight	10.5518					
P to total ratio	0.047302301					

### 5.3 Appendix C

Table C. 1. *T. latifolia* Germination lighting Conditions

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Date	Mesocosm	Old meter Light Intensity ( $\mu\text{E}$ )	New meter Light Intensity ( $\mu\text{E}$ )
05/25/12	1		116
05/25/12	2		131
05/25/12	3		140
05/25/12	4		127
05/25/12	5		120
05/25/12	6		166
05/25/12	7		164
05/25/12	8		156
05/25/12	9		156
05/25/12	10		182
05/25/12	11		180
05/25/12	12		156
05/25/12	13		147
05/25/12	14		169
05/25/12	15		169
05/25/12	16		143
05/25/12	17		134
05/25/12	18		161
05/25/12	19		155
05/25/12	20		143
05/25/12	21		138
05/25/12	22		149
05/25/12	23		144
05/25/12	24		128
05/25/12	25		140
05/25/12	26		153
05/25/12	27		146
05/25/12	28		129
05/25/12	29		142
05/25/12	30		155
05/25/12	31		148
05/25/12	32		144
05/25/12	33		135
05/25/12	34		153

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05/25/12	35	154
05/25/12	36	132
05/25/12	37	121
05/25/12	38	127
05/25/12	39	123
05/25/12	40	102
Mean		144.45
Stddev		17.44287
Min		102
Max		182

**Table C2.** Light intensity observed within microcosms at different locations and conditions under light bank 1.

Date	Microcosm	Light Bank Location	Transect location	New meter Light Intensity ( $\mu\text{E}$ )	Comments
06/28/12	9	19	1	139	Surrounded by other mesocosms
06/28/12	9	19	2	142	Surrounded by other mesocosms
06/28/12	9	19	3	135	Surrounded by other mesocosms
06/28/12	9	19	4	146	Surrounded by other mesocosms
06/28/12	9	19	5	150	Surrounded by other mesocosms
06/28/12	9	19	6	140	Surrounded by other mesocosms
06/28/12	9	19	7	141	Surrounded by other mesocosms
06/28/12	9	19	8	144	Surrounded by other mesocosms
06/28/12	9	19	9	138	Surrounded by other mesocosms
06/28/12	10	19	1	141	Surrounded by other mesocosms
06/28/12	10	19	2	136	Surrounded by other mesocosms
06/28/12	10	19	3	134	Surrounded by other mesocosms
06/28/12	10	19	4	145	Surrounded by other mesocosms
06/28/12	10	19	5	150	Surrounded by other mesocosms
06/28/12	10	19	6	141	Surrounded by other mesocosms
06/28/12	10	19	7	142	Surrounded by other mesocosms
06/28/12	10	19	8	144	Surrounded by other mesocosms
06/28/12	10	19	9	137	Surrounded by other mesocosms

06/28/12	3	19	1	135	Surrounded by other mesocosms
06/28/12	3	19	2	138	Surrounded by other mesocosms
06/28/12	3	19	3	133	Surrounded by other mesocosms
06/28/12	3	19	4	144	Surrounded by other mesocosms
06/28/12	3	19	5	150	Surrounded by other mesocosms
06/28/12	3	19	6	142	Surrounded by other mesocosms
06/28/12	3	19	7	138	Surrounded by other mesocosms
06/28/12	3	19	8	142	Surrounded by other mesocosms
06/28/12	3	19	9	141	Surrounded by other mesocosms
					Surrounded by other mesocosms
06/28/12	21	19	1	136	Surrounded by other mesocosms
06/28/12	21	19	2	135	Surrounded by other mesocosms
06/28/12	21	19	3	132	Surrounded by other mesocosms
06/28/12	21	19	4	144	Surrounded by other mesocosms
06/28/12	21	19	5	149	Surrounded by other mesocosms
06/28/12	21	19	6	142	Surrounded by other mesocosms
06/28/12	21	19	7	135	Surrounded by other mesocosms
06/28/12	21	19	8	138	Surrounded by other mesocosms
06/28/12	21	19	9	137	Surrounded by other mesocosms
06/28/12	22	19	1	140	Surrounded by other mesocosms
06/28/12	22	19	2	137	Surrounded by other mesocosms
06/28/12	22	19	3	138	Surrounded by other mesocosms
06/28/12	22	19	4	146	Surrounded by other mesocosms
06/28/12	22	19	5	151	Surrounded by other mesocosms
06/28/12	22	19	6	144	Surrounded by other mesocosms
06/28/12	22	19	7	141	Surrounded by other mesocosms
06/28/12	22	19	8	140	Surrounded by other mesocosms
06/28/12	22	19	9	136	Surrounded by other mesocosms
06/28/12	9	19	1	143	Not Surrounded
06/28/12	9	19	2	143	Difference_ light btwn surrouned and not surrouned = 2.6 $\mu$ E
06/28/12	9	19	3	139	Not Surrounded
06/28/12	9	19	4	150	Not Surrounded
06/28/12	9	19	5	153	Not Surrounded
06/28/12	9	19	6	145	Not Surrounded
06/28/12	9	19	7	143	Not Surrounded
06/28/12	9	19	8	142	Not Surrounded
06/28/12	9	19	9	140	Not Surrounded
06/28/12	13	15	1	159	

06/28/12	13	15	2	156
06/28/12	13	15	3	159
06/28/12	13	15	4	163
06/28/12	13	15	5	168
06/28/12	13	15	6	163
06/28/12	13	15	7	154
06/28/12	13	15	8	147
06/28/12	13	15	9	152
06/28/12	33	31	1	145
06/28/12	33	31	2	143
06/28/12	33	31	3	140
06/28/12	33	31	4	149
06/28/12	33	31	5	154
06/28/12	33	31	6	157
06/28/12	33	31	7	137
06/28/12	33	31	8	140
06/28/12	33	31	9	135
07/05/12	3	12	1	122
07/05/12	3	12	2	125
07/05/12	3	12	3	120
07/05/12	3	12	4	137
07/05/12	3	12	5	147
07/05/12	3	12	6	141
07/05/12	3	12	7	157
07/05/12	3	12	8	160
07/05/12	3	12	9	156
07/05/12	9	23	1	122
07/05/12	9	23	2	123
07/05/12	9	23	3	124
07/05/12	9	23	4	133
07/05/12	9	23	5	138
07/05/12	9	23	6	132
07/05/12	9	23	7	123
07/05/12	9	23	8	130
07/05/12	9	23	9	124
07/05/12	10	8	1	115
07/05/12	10	8	2	116
07/05/12	10	8	3	117
07/05/12	10	8	4	135

07/05/12	10	8	5	137
07/05/12	10	8	6	132
07/05/12	10	8	7	143
07/05/12	10	8	8	142
07/05/12	10	8	9	143

*Typha latifolia* response to oligotrophic and eutrophic N and P loading rates under laboratory conditions

Table D.1. Light intensity (PAR) characteristics for experimental unit microcosms observed on 11/09/12 and 05/08/13.

Each cart has two rows for four microcosm locations. Locations are in order of W to E for locations 1 -4, 5 - 8, 9 -12 ect.  
 HTTL = height to tallest leaf  
 Soil Level measured outside of microcosm adjacent to vessel height of black duct tape which indicated soil height  
 Measurement at microcosm center collected by resting extension on vessel lip, perpendicular to soil level, as close to the middle of the microcosm as possible without physically contacting ramets (shoots)

Date (mm/dd/yy)	Location	Cart	Microcosm	Measurement Location		
				Soil Level ( $\mu\text{E m}^{-2} \text{s}^{-1}$ )	HHTL ( $\mu\text{E m}^{-2} \text{s}^{-1}$ )	Approx. Microcosm Center ( $\mu\text{E m}^{-2} \text{s}^{-1}$ )
11/09/12	1	1	11	23	52	33
11/09/12	2	1	32	24	57	23
11/09/12	3	1	20	26	55	32
11/09/12	4	1	2	24	53	31
11/09/12	5	1	22	30	59	28
11/09/12	6	1	12	31	60	30
11/09/12	7	1	10	34	55	40
11/09/12	8	1	4	32	64	33
11/09/12	9	2	8	30	74	32
11/09/12	10	2	24	31	69	38
11/09/12	11	2	31	34	74	40
11/09/12	12	2	34	32	62	40
11/09/12	13	2	27	31	77	37
11/09/12	14	2	15	27	75	42
11/09/12	15	2	7	29	75	38
11/09/12	16	2	29	28	76	38
11/09/12	17	3	6	31	70	41
11/09/12	18	3	37	27	81	42
11/09/12	19	3	28	29	90	37

11/09/12	20	3	23	28	68	37
11/09/12	21	3	18	35	69	39
11/09/12	22	3	17	37	75	43
11/09/12	23	3	9	37	78	42
11/09/12	24	3	16	36	71	42
11/09/12	25	4	35	35	78	43
11/09/12	26	4	40	37	58	44
11/09/12	27	4	1	37	72	47
11/09/12	28	4	19	36	75	42
11/09/12	29	4	13	28	66	43
11/09/12	30	4	25	29	61	44
11/09/12	31	4	38	33	73	38
11/09/12	32	4	36	31	69	39
11/09/12	33	5	26	28	66	39
11/09/12	34	5	5	29	64	36
11/09/12	35	5	3	33	59	40
11/09/12	36	5	14	31	61	28
11/09/12	37	5	21	26	57	34
11/09/12	38	5	30	26	51	36
11/09/12	39	5	39	23	57	34
11/09/12	40	5	33	27	68	28
05/08/13	1	1	4	7	59	26
05/08/13	2	1	11	10	88	32
05/08/13	3	1	32	8	32	31
05/08/13	4	1	2	7	54	29
05/08/13	5	1	20	18	72	34
05/08/13	6	1	22	22	59	36
05/08/13	7	1	10	26	70	36
05/08/13	8	1	12	21	56	30
05/08/13	9	2	15	12	83	28
05/08/13	10	2	8	15	72	36
05/08/13	11	2	31	15	101	37
05/08/13	12	2	24	14	83	34
05/08/13	13	2	27	23	108	35
05/08/13	14	2	7	22	112	35
05/08/13	15	2	29	20	110	31
05/08/13	16	2	34	21	75	38
05/08/13	17	3	23	22	86	38
05/08/13	18	3	6	20	84	38
05/08/13	19	3	16	19	108	34
05/08/13	20	3	9	19	41	40
05/08/13	21	3	28	20	118	34
05/08/13	22	3	37	22	89	40
05/08/13	23	3	18	20	93	38
05/08/13	24	3	17	22	76	29
05/08/13	25	4	36	19	86	36
05/08/13	26	4	1	22	96	36
05/08/13	27	4	19	22	89	34
05/08/13	28	4	13	22	84	35
05/08/13	29	4	38	18	78	36



05/08/13	30	4	40	14	94	37
05/08/13	31	4	35	19	70	30
05/08/13	32	4	25	18	84	37
05/08/13	33	5	21	22	71	32
05/08/13	34	5	3	19	77	31
05/08/13	35	5	14	19	77	33
05/08/13	36	5	33	22	112	28
05/08/13	37	5	5	10	65	28
05/08/13	38	5	26	10	77	28
05/08/13	39	5	30	11	47	30
05/08/13	40	5	39	10	64	26

## 5.4 Appendix D

**Table D1. *Typha Latifolia* Seed Germination**

Three seeds were added to each microcosm

Microcosm	Treat.	Date	Number Germinated	Volume of water added post assessment (ml)	Standing water (% surface area)	Wet sediment (% surface area)	Moist sediment (% surface area)	Germ. Rate
4	1	05/27/12	1	15	100			0.333
5	1	05/27/12	1	15	100			0.333
8	1	05/27/12	1	15	40	60		0.333
12	1	05/27/12	none obser.	15	25	75		0.000
13	1	05/27/12	1	15	5	95		0.333
19	1	05/27/12	2	15		100		0.667
31	1	05/27/12	3	15			100	1.000
33	1	05/27/12	1	15		100		0.333
35	1	05/27/12	1	15			100	0.333
37	1	05/27/12	none obser.	15		100		0.000
4	1	05/28/12	2	15	60	40		0.667
5	1	05/28/12	3	15		100		1.000
8	1	05/28/12	2	15		100		0.667
12	1	05/28/12	2	15		100		0.667
13	1	05/28/12	2	15		100		0.667
19	1	05/28/12	3	15			100	1.000
31	1	05/28/12	3	15			100	1.000
33	1	05/28/12	2	15		100		0.667
35	1	05/28/12	3	15			100	1.000
37	1	05/28/12	none obser.	15		100		0.000

4	1	05/29/12	2	15		100		0.667
5	1	05/29/12	3	15		50	50	1.000
8	1	05/29/12	2	15		100		0.667
12	1	05/29/12	3	15		100		1.000
13	1	05/29/12	2	14.5		50	50	0.667
19	1	05/29/12	3	15			100	1.000
31	1	05/29/12	3	15			100	1.000
33	1	05/29/12	3	15.5		100		1.000
35	1	05/29/12	3	15			100	1.000
37	1	05/29/12	none obser.	15		100		0.000
13	1	06/01/12	2					0.667
8	1	06/01/12	2					0.667
4	1	06/01/12	2					0.667
35	1	06/01/12	3					1.000
33	1	06/01/12	3					1.000
31	1	06/01/12	3					1.000
19	1	06/01/12	3					1.000
12	1	06/01/12	3					1.000
5	1	06/01/12	3					1.000
37	1	06/01/12	none obser.					
2	2	05/27/12	1	15		100		0.333
3	2	05/27/12	3	15		100		1.000
7	2	05/27/12	1	15	100			0.333
14	2	05/27/12	2	15	40	60		0.667
17	2	05/27/12	1	15	30	70		0.333
21	2	05/27/12	none obser.	15	95	5		0.000
23	2	05/27/12	1	15	95	5		0.333
34	2	05/27/12	none obser.	15	100			0.000
36	2	05/27/12	1	15	70	30		0.333
39	2	05/27/12	3	15			100	1.000
2	2	05/28/12	3	15			100	1.000
3	2	05/28/12	3	15			100	1.000
7	2	05/28/12	2	15	25	100		0.667
14	2	05/28/12	3	15		100		1.000
17	2	05/28/12	2	15		100		0.667
21	2	05/28/12	3	15		100		1.000
23	2	05/28/12	none obser.	15		100		0.000
34	2	05/28/12	2	15	100			0.667
36	2	05/28/12	2	15	10	90		0.667
39	2	05/28/12	3	15		100		1.000

2	2	05/29/12	3	15			100	1.000
3	2	05/29/12	3	15			100	1.000
7	2	05/29/12	3	15.5		50	50	1.000
14	2	05/29/12	3	15			100	1.000
17	2	05/29/12	3	15		100		1.000
21	2	05/29/12	2	15		100		0.667
23	2	05/29/12	3	15		100		0.000
34	2	05/29/12	3	15	30	70		1.000
36	2	05/29/12	2	15		100		0.667
39	2	05/29/12	3	15			100	1.000
36	2	06/01/12	2					0.667
21	2	06/01/12	2					0.667
39	2	06/01/12	3					1.000
34	2	06/01/12	3					1.000
23	2	06/01/12	3					1.000
17	2	06/01/12	3					1.000
14	2	06/01/12	3					1.000
7	2	06/01/12	3					1.000
3	2	06/01/12	3					1.000
2	2	06/01/12	3					1.000
1	3	05/27/12	1	15	100			0.333
6	3	05/27/12	1	15	100			0.333
9	3	05/27/12	none obser.	15	15	85		0.000
10	3	05/27/12	2	15			100	0.667
11	3	05/27/12	none obser.	15	25	75		0.000
20	3	05/27/12	none obser.	15		100		0.000
25	3	05/27/12	3	15	50	50		1.000
29	3	05/27/12	none obser.	15	95	5		0.000
30	3	05/27/12	1	15	60	40		0.333
40	3	05/27/12	2	15	60	40		0.667
1	3	05/28/12	2	15	95	5		0.667
6	3	05/28/12	3	15		100		1.000
9	3	05/28/12	3	15		100		1.000
10	3	05/28/12	2	15			100	0.667
11	3	05/28/12	2	15		100		0.667
20	3	05/28/12	2	15	50	50		0.667
25	3	05/28/12	3	15		100		1.000
29	3	05/28/12	3	15		100		1.000
30	3	05/28/12	2	15		100		0.667

40	3	05/28/12	2	15		100		0.667
1	3	05/29/12	2	15	100			0.667
6	3	05/29/12	3	15			100	1.000
9	3	05/29/12	3	15			100	1.000
10	3	05/29/12	2	15			100	0.667
11	3	05/29/12	2	15			100	0.667
20	3	05/29/12	3	15		100		1.000
25	3	05/29/12	3	15		100		1.000
29	3	05/29/12	2	15		100		0.667
30	3	05/29/12	2	15		100		0.667
40	3	05/29/12	2	15		100		0.667
40	3	06/01/12	2					0.667
30	3	06/01/12	2					0.667
29	3	06/01/12	2					0.667
25	3	06/01/12	2					0.667
11	3	06/01/12	2					0.667
10	3	06/01/12	2					0.667
1	3	06/01/12	2					0.667
20	3	06/01/12	3					1.000
9	3	06/01/12	3					1.000
6	3	06/01/12	3					1.000
15	4	05/27/12	1	15		100		0.333
16	4	05/27/12	none obser.	15	40	60		0.000
18	4	05/27/12	1	15	5	95		0.333
22	4	05/27/12	1	15		100		0.333
24	4	05/27/12	1	15			100	0.333
26	4	05/27/12	none obser.	15	10	90		0.000
27	4	05/27/12	1	15	20	80		0.333
28	4	05/27/12	2	15	40	60		0.667
32	4	05/27/12	2	15	10	90		0.667
38	4	05/27/12	3	15		100		1.000
15	4	05/28/12	2	15			100	0.667
16	4	05/28/12	3	15		100		1.000
18	4	05/28/12	3	15			100	1.000
22	4	05/28/12	2	15		100		0.667
24	4	05/28/12	2	15			100	0.667
26	4	05/28/12	2	15			100	0.667
27	4	05/28/12	3	15		100		1.000
28	4	05/28/12	2	15		100		0.667

32	4	05/28/12	2	15	100		0.667
38	4	05/28/12	4	15	100		0.667
15	4	05/29/12	2	15		100	0.667
16	4	05/29/12	3	15	50	50	1.000
18	4	05/29/12	3	15		100	1.000
22	4	05/29/12	3	15		100	1.000
24	4	05/29/12	2	30		100	0.667
26	4	05/29/12	2	15		100	0.667
27	4	05/29/12	3	15		100	1.000
28	4	05/29/12	3	15	100		1.000
32	4	05/29/12	2	15	100		0.667
38	4	05/29/12	4	15		100	0.667
32	4	06/01/12	2	3			0.667
26	4	06/01/12	2	3			0.667
24	4	06/01/12	2	3			0.667
15	4	06/01/12	2	3			0.667
28	4	06/01/12	3	3			1.000
27	4	06/01/12	3	3			1.000
22	4	06/01/12	3	3			1.000
18	4	06/01/12	3	3			1.000
16	4	06/01/12	3	3			1.000
38	4	06/01/12	4	6			0.667

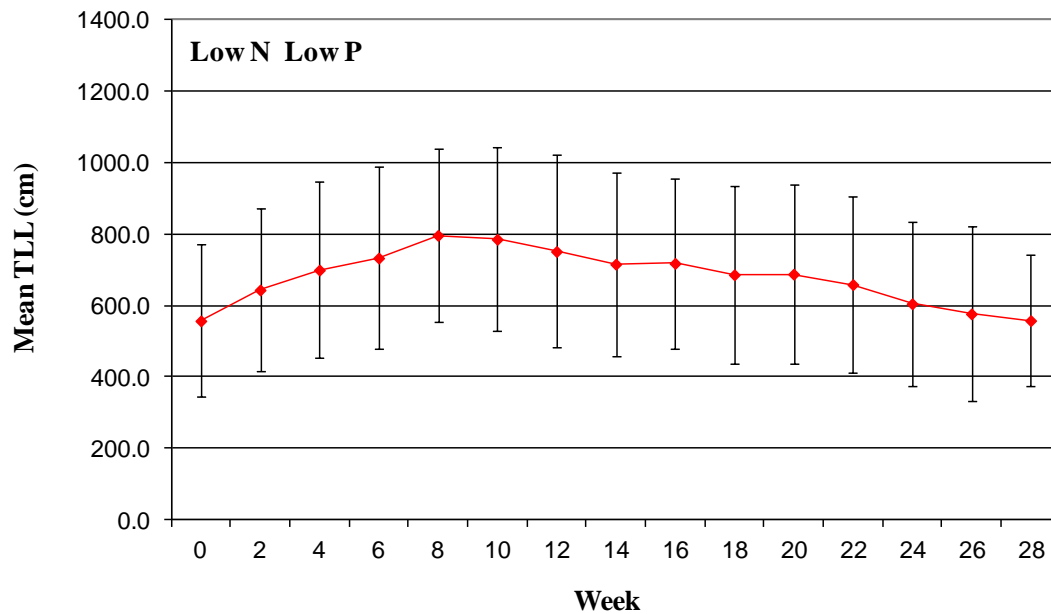
**Table D2. *Typha latifolia* Seedling Biomass**

Micro.	N	Treat.	Sample Date	P1 biomass (g)	P2 biomass (g)	P3 biomass (g)	Root Biomass (g)	Total biomass (g)
4		1	08/18/12	7.3				7.3
5		1	08/18/12	6.11			1.82	7.93
8		1	08/18/12	3.83	0.07		0.42	4.32
12		1	08/18/12	0.41				0.41
19		1	08/18/12	2.12	1.1		1.58	4.8
31		1	08/16/12	4.93	0.44		1.46	6.83
37	7	1	08/18/12	0.34	3.71		0.9	4.95
2		2	08/18/12	2.73	1.07		1.29	5.09
7		2	08/16/12	1.03	1.47	1.67	2.29	6.46
14		2	08/18/12	2.95	0.1		1.09	4.14
17		2	08/18/12	2.34	0.31	2.43	1.83	6.91
34		2	08/16/12	1.56	0.27	3.23	2.57	7.63

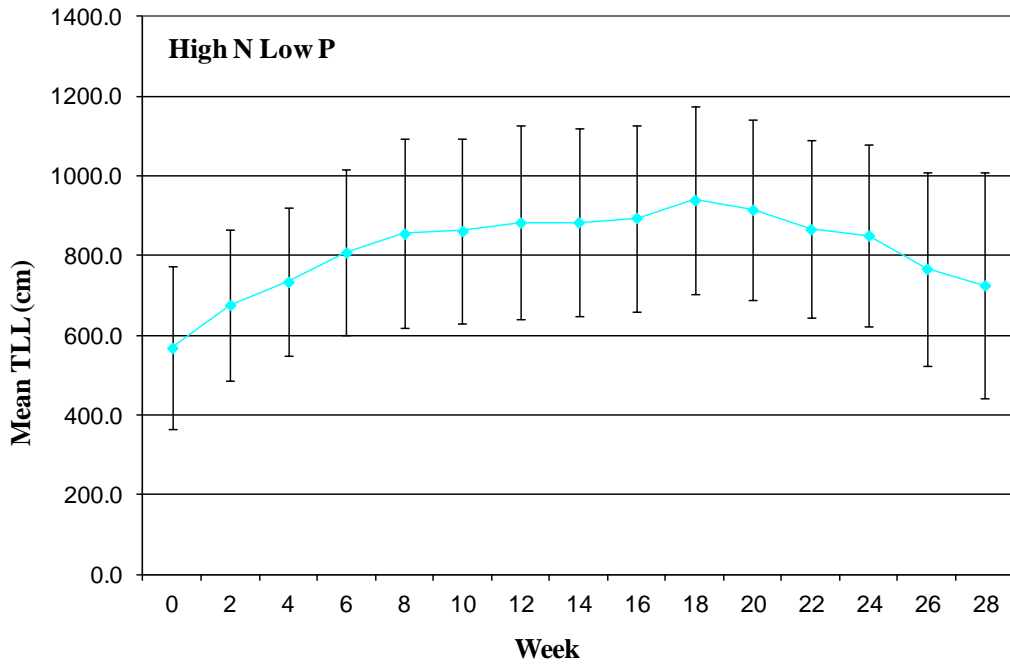
36		2	08/18/12	3.42	1.24		1.05	5.71
39	7	2	08/18/12		4.12		0.62	4.74
1		3	08/16/12	2.61	2.53		0.52	5.66
6		3	08/18/12	0.33	3.5		1.15	4.98
11		3	08/19/12	0.54	6.08		0.2	6.82
20		3	08/19/12	0.54	6.08		0.2	6.82
29		3	08/16/12	0	4.15		1.56	5.71
30		3	08/19/12	5.1			0.11	5.21
40	7	3	08/19/12	5.63	0.54		0.58	6.75
15		4	08/19/12	5.36	1.02			6.38
16		4	08/19/12	0.29	1.73		0.06	2.08
24		4	08/19/12	3.95			0.79	4.74
26		4	08/19/12	4.27			0.89	5.16
27		4	08/19/12	0.42	5.47	0.8	0.19	6.88
28		4	08/19/12	6.75			0.74	7.49
38	7	4	08/16/12	2.2				2.2

## *Typha latifolia* response to N and P

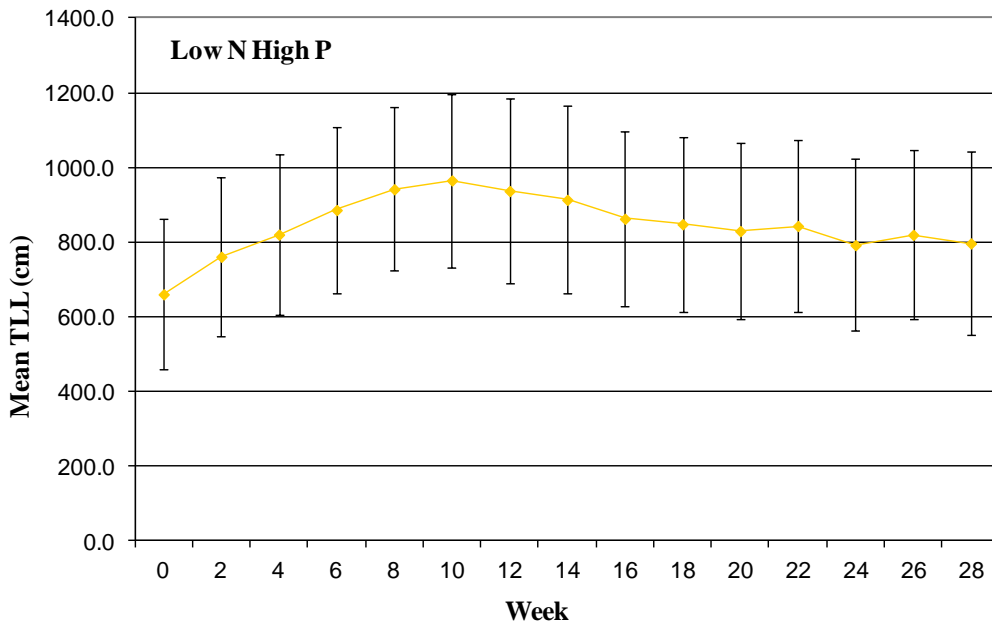
### Standard Deviations for mean TLL biweekly measurements



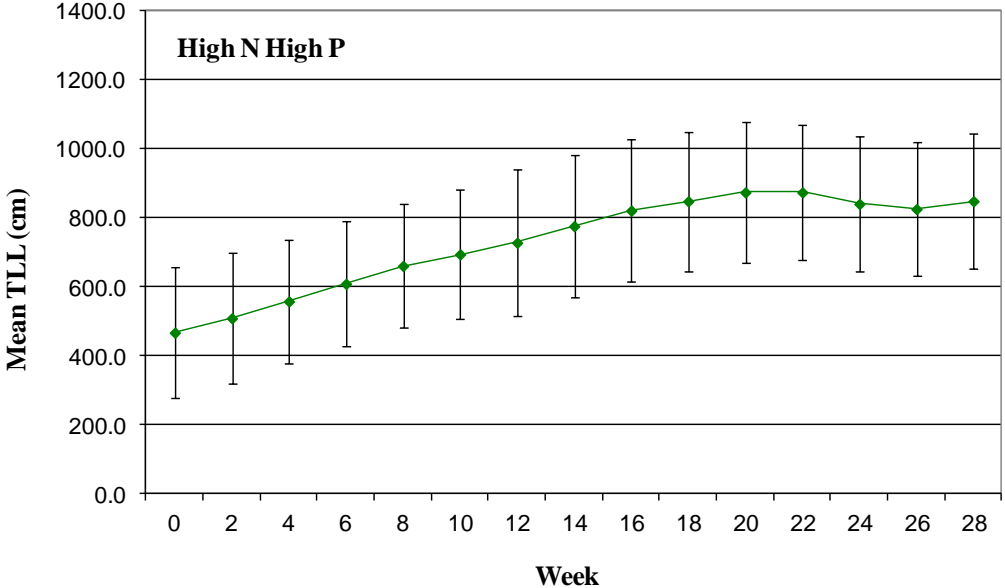
**Figure D1. Mean TLL and standard deviations *Typha latifolia* reared in Low N and Low P.**



**Figure D2. Mean TLL and standard deviations *Typha latifolia* reared in High N and Low P.**



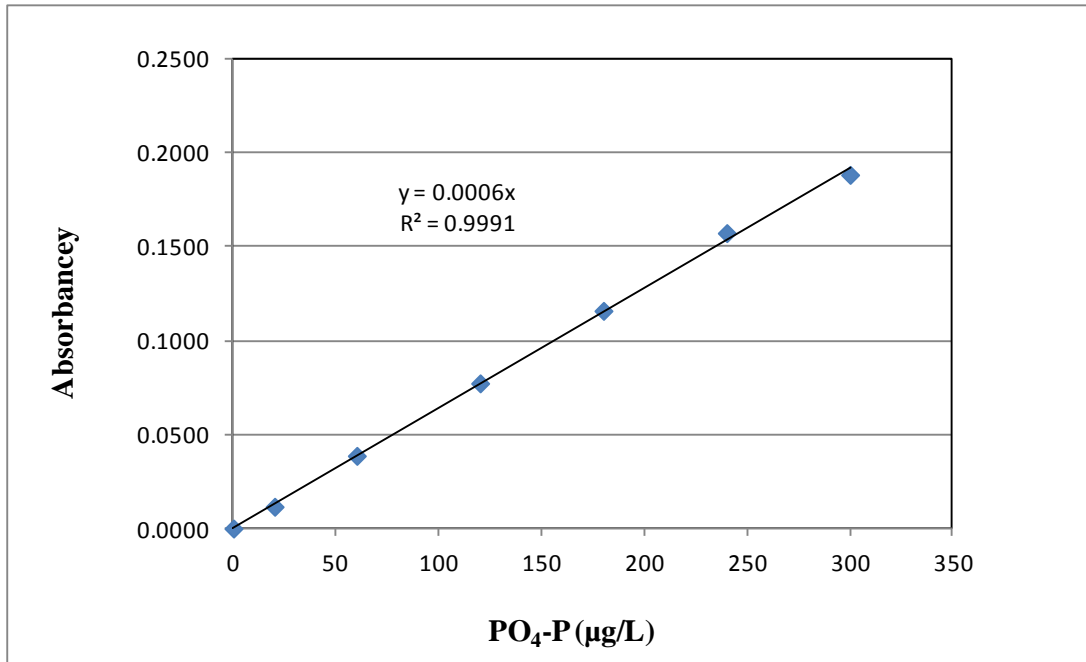
**Figure D3. Mean TLL and standard deviations *Typha latifolia* reared in Low N and High P.**



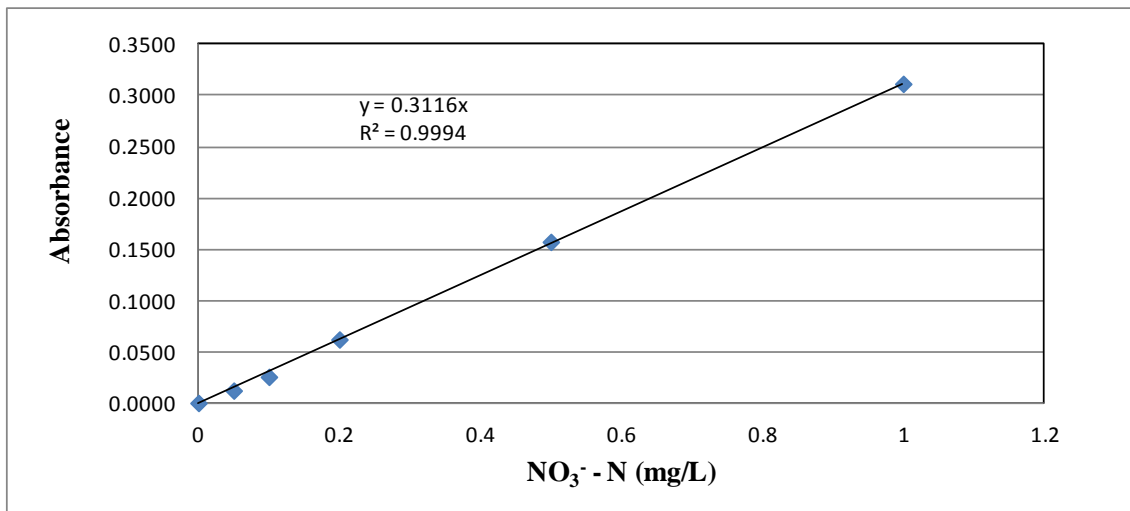
**Figure D4. Mean TLL and standard deviations *Typha latifolia* reared in High N and High P.**



## 5.5 Appendix E



**Figure E1.** Phosphate standard curve used to estimate the porewater phosphate concentrations in sediments used to propagate *Typha latifolia* seedling.



**Figure E2.** Nitrate standard curve used to estimate the nitrate concentration in City of Toronto Municipal drinking water provided to microcosms for the *Typha latifolia* seedling propagation and response to N and P for the acclimation period and first month of the *Typha latifolia* response to N and P loading.

## Water Quality: *Typha latifolia* response to oligotrophic and eutrophic N and P loading rates under laboratory conditions

All water samples collected before watering/grow media additions except in some cases salinity was measured before and after watering/grow media additions.

**Table E. 1. Supporting water quality data for *Typha latifolia* response to N and P treatments.**

Wk = week; M. = microcosm; T. = treatment; DO = dissolved oxygen; PSC = water conductivity meter; cond = conductivity

N	Date (mm/dd/yy)	Wk	M.	T.	DO meter (°C)	PSC meter (°C)	DO (mg/L)	pH	Cond. (µS/cm)	Salinity (mg/L)	TDS (mg/L)
1	12/06/12	6	3	2	22.5	22.4	6.89	6.49	698.0	332.0	495
2	12/06/12	6	4	1	22.4	22.1	6.92	6.69	445.0	200.0	306
3	12/06/12	6	20	4	21.3	22.8	5.66	6.26	677.0	325.0	486
4	12/06/12	6	2	4	22.4	22.8	9.02	7.10	342.0	161.0	243
5	12/06/12	6	32	1	22.2	22.5	10.54	7.36	662.0	315.0	471
6	12/06/12	6	11	3	22.3	22.4	8.99	7.05	429.0	201.0	305
7	12/06/12	6	22	2	23.3	22.4	11.29	7.81	580.0	267.0	393
8	12/06/12	6	12	1	22.4	21.1	7.44	6.62	332.0	158.0	245
9	12/06/12	6	31	3	22.3	22.6	7.74	6.84	690.0	331.0	493
10	12/06/12	6	9	3	22.2	22.5	7.47	6.80	441.0	207.0	313
11	12/06/12	6	36	4	22.2	22.5	10.51	7.62	684.0	324.0	488
12	12/06/12	6	38	2	21.9	22.2	11.03	8.02	462.0	218.0	328
1	01/10/13	11	3	2	20.4	21.2	7.47	6.40	607.0	292.0	430
2	01/10/13	11	4	1	20.4	21.2	9.51	6.45	401.0	192.0	285
3	01/10/13	11	20	4	20.5	21.4	7.86	6.25	661.0	317.0	471
4	01/10/13	11	2	4	20.3	21.1	10.15	6.78	429.0	203.0	305
5	01/10/13	11	32	1	20.5	21.2	11.38	7.35	713.0	321.0	506
6	01/10/13	11	11	3	20.4	21.2	11.28	7.11	450.0	214.0	315
7	01/10/13	11	22	2	20.3	21.2	10.16	7.01	585.0	283.0	418
8	01/10/13	11	12	1	20.4	21.1	9.33	7.23	301.0	144.0	213
9	01/10/13	11	31	3	20.4	21.2	10.84	6.99	588.0	286.0	404
10	01/10/13	11	33	3	20.4	21.1	7.16	6.38	258.0	123.0	183
11	01/10/13	11	36	4	20.4	21.2	6.94	6.65	646.0	310.0	458
12	01/10/13	11	38	1	20.6	21.3	10.73	7.09	447.0	214.0	318
1	01/11/13	15	1	2		22.0				342.0	
2	01/11/13	15	2	4		22.1				216.0	
3	01/11/13	15	3	2		22.1				316.0	
4	01/11/13	15	4	1		22.0				206.0	
5	01/11/13	15	5	1		21.9				332.0	
6	01/11/13	15	6	3		21.9				250.0	
7	01/11/13	15	7	4		22.2				186.0	
8	01/11/13	15	8	1		22.1				341.0	

9	01/11/13	15	9	3								
10	01/11/13	15	10	4		22.3					316.0	
11	01/11/13	15	11	3		21.9					238.0	
12	01/11/13	15	12	1		21.9					157.0	
13	01/11/13	15	13	1		21.5					272.0	
14	01/11/13	15	14	3		22.0					343.0	
15	01/11/13	15	15	2		22.1					215.0	
16	01/11/13	15	16	2		21.8					181.0	
17	01/11/13	15	17	2								
18	01/11/13	15	18	4		22.1					220.0	
19	01/11/13	15	19	1		21.6					220.0	
20	01/11/13	15	20	4		22.0					287.0	
21	01/11/13	15	21	3		21.9					79.5	
22	01/11/13	15	22	2		22.0					295.0	
23	01/11/13	15	23	3		21.9					175.0	
24	01/11/13	15	24	1		22.0					265.0	
25	01/11/13	15	25	1		21.8					385.0	
26	01/11/13	15	26	4		21.9					101.0	
27	01/11/13	15	27	2		20.0					172.0	
28	01/11/13	15	28	3		21.8					59.3	
29	01/11/13	15	29	4		22.1					80.6	
30	01/11/13	15	30	1		21.9					315.0	
31	01/11/13	15	31	3		22.0					300.0	
32	01/11/13	15	32	1		22.0					359.0	
33	01/11/13	15	33	3		21.8					144.0	
34	01/11/13	15	34	2		22.1					337.0	
35	01/11/13	15	35	4		22.1					381.0	
36	01/11/13	15	36	4		22.0					327.0	
37	01/11/13	15	37	2		22.1					201.0	
38	01/11/13	15	38	2		21.9					230.0	
39	01/11/13	15	39	3		21.9					123.0	
40	01/11/13	15	40	4		22.1					404.0	
1	02/12/13	15	1	2	20.3	21.0	10.02	6.93	694.0	332.0		Not done
2	02/12/13	15	2	4	20.4	21.3	9.22	6.99	543.0	259.0		Not done
3	02/12/13	15	3	2	21.1	20.4	7.94	6.32	623.0	301.0		Not done
4	02/12/13	15	4	1	20.4	21.0	9.27	6.69	545.0	288.0		Not done
5	02/12/13	15	5	1	20.6	20.7	10.09	6.47	704.0	361.0		Not done
6	02/12/13	15	6	3	20.2	20.7	10.42	6.70	583.0	272.0		Not done
7	02/12/13	15	7	4	20.4	21.3	7.67	6.55	326.0	161.0		Not done
8	02/12/13	15	8	1	20.3	21.0	9.48	6.60	837.0	400.0		Not done
9	02/12/13	15	9	3								
10	02/12/13	15	10	4	20.4	20.9	10.35	6.67	570.0	273.0		Not done
11	02/12/13	15	11	3	20.6	20.9	11.07	6.94	484.0	245.0		Not done

12	02/12/13	15	12	1	20.4	21.0	9.95	6.77	373.0	180.0	done
13	02/12/13	15	13	1	20.4	20.6	9.34	6.75	656.0	304.0	Not done
14	02/12/13	15	14	3	20.7	20.6	8.97	6.52	748.0	365.0	Not done
15	02/12/13	15	15	2	20.2	21.0	9.27	6.89	500.0	237.0	Not done
16	02/12/13	15	16	2	20.4	20.7		6.09	151.5	91.6	Not done
17	02/12/13	15	17	2	20.4	20.8	8.25	6.14	361.0	179.0	Not done
18	02/12/13	15	18	4	20.4	21.0	10.27	6.66	497.0	237.0	Not done
19	02/12/13	15	19	1	20.4	21.1	9.07	6.66	508.0	248.0	Not done
20	02/12/13	15	20	4	20.6	21.0	6.74	6.12	678.0	350.0	Not done
21	02/12/13	15	21	3	20.5	20.9	9.41	6.41	205.0	99.4	Not done
22	02/12/13	15	22	2	20.6	21.2	8.48	6.77	717.0	345.0	Not done
23	02/12/13	15	23	3	20.2	20.8	10.39	6.44	395.0	185.0	Not done
24	02/12/13	15	24	1	20.4	21.3	10.64	6.82	681.0	333.0	Not done
25	02/12/13	15	25	1	20.4	20.8	7.96	6.70	749.0	361.0	Not done
26	02/12/13	15	26	4	20.7	20.8	9.87	6.54	190.0	94.0	Not done
27	02/12/13	15	27	2	20.3	20.8	7.83	6.39	439.0	211.0	Not done
28	02/12/13	15	28	3	20.8	20.1	6.51	6.30	187.0	87.2	Not done
29	02/12/13	15	29	4	20.4	20.8	7.60	6.57	176.3	78.9	Not done
30	02/12/13	15	30	1	20.7	21.2	7.55	6.46	702.0	338.0	Not done
31	02/12/13	15	31	3	20.0	21.0	10.88	6.59	710.0	348.0	Not done
32	02/12/13	15	32	1	20.6	21.1	8.72	6.90	763.0	371.0	Not done
33	02/12/13	15	33	3	20.6	20.9	6.15	6.24	198.0	96.9	Not done
34	02/12/13	15	34	2	20.1	21.1	9.80	6.87	810.0	395.0	Not done
35	02/12/13	15	35	4	20.4	20.9	10.24	6.76	804.0	384.0	Not done
36	02/12/13	15	36	4	19.9	20.6	9.51	6.67	544.0	294.0	Not done
37	02/12/13	15	37	2	20.3	21.0	9.73	6.62	491.0	238.0	Not done
38	02/12/13	15	38	2	20.2	20.7	7.47	6.70	519.0	247.0	Not done
39	02/12/13	15	39	3	20.6	20.6	6.38	6.22	290.0	138.0	Not done

												done
												Not done
40	02/12/13	15	40	4	20.4	20.8	8.18	6.81	801.0	389.0		
1	03/12/13	19	3	2	20.7	20.9	9.54	6.46	588.0	296.0		418
2	03/12/13	19	4	1	20.5	20.9	9.27	7.35	651.0	342.0		471
3	03/12/13	19	20	4	21.0	21.1	5.45	6.14	554.0	286.0		393
4	03/12/13	19	2	4	20.9	21.1	9.28	6.46	433.0	221.0		306
5	03/12/13	19	32	1	20.5	20.9	8.58	6.63	741.0	384.0		522
6	03/12/13	19	11	3	20.4	20.5	11.09	6.63	460.0	233.0		330
7	03/12/13	19	22	2	21.0	21.0	9.89	6.58	741.0	380.0		527
8	03/12/13	19	12	1	20.6	20.6	10.36	6.73	371.0	201.0		283
9	03/12/13	19	31	3	20.9	20.8	10.32	6.75	701.0	361.0		457
10	03/12/13	19	33	3	21.0	21.2	4.60	6.42	219.0	112.0		156
11	03/12/13	19	36	4	20.9	21.1	11.50	7.21	474.0	245.0		289
12	03/12/13	19	38	2	20.9	21.2	8.14	6.65	462.0	241.0		334
1	04/16/13	24	1	2		22.4				222.0		
2	04/16/13	24	2	4		22.4				210.0		
3	04/16/13	24	3	2		22.4				308.0		
4	04/16/13	24	4	1		22.3				324.0		
5	04/16/13	24	5	1		22.1				349.0		
6	04/16/13	24	6	3		22.4				341.0		
7	04/16/13	24	7	4		22.0				102.0		
8	04/16/13	24	8	1		22.2				419.0		
9	04/16/13	24	9	3								
10	04/16/13	24	10	4		22.3				302.0		
11	04/16/13	24	11	3		22.5				191.0		
12	04/16/13	24	12	1		21.9				247.0		
13	04/16/13	24	13	1		22.4				367.0		
14	04/16/13	24	14	3		22.3				323.0		
15	04/16/13	24	15	2		21.8				300.0		
16	04/16/13	24	16	2		22.1				132.0		
17	04/16/13	24	17	2		22.1				207.0		
18	04/16/13	24	18	4		22.2				246.0		
19	04/16/13	24	19	1		22.6				282.0		
20	04/16/13	24	20	4		22.4				224.0		
21	04/16/13	24	21	3		22.5				122.0		
22	04/16/13	24	22	2		22.4				453.0		
23	04/16/13	24	23	3		22.3				229.0		
24	04/16/13	24	24	1		22.2				323.0		
25	04/16/13	24	25	1		22.3				311.0		
26	04/16/13	24	26	4		22.4				70.1		
27	04/16/13	24	27	2		21.7				221.0		
28	04/16/13	24	28	3		22.5				83.8		
29	04/16/13	24	29	4		22.1				77.4		
30	04/16/13	24	30	1		22.3				355.0		
31	04/16/13	24	31	3		22.1				303.0		
32	04/16/13	24	32	1		22.6				401.0		
33	04/16/13	24	33	3								
34	04/16/13	24	34	2		22.3				448.0		
35	04/16/13	24	35	4		22.4				276.0		
36	04/16/13	24	36	4		22.5				215.0		

37	04/16/13	24	37	2		22.0					312.0	
38	04/16/13	24	38	2		22.3					242.0	
39	04/16/13	24	39	3		22.2					128.0	
40	04/16/13	24	40	4		22.4					282.0	
1	04/18/13	25	3	2	23.4	23.7	8.24	6.30	582.0	276.0	419	
2	04/18/13	25	4	1	23.0	23.5	8.55	6.57	623.0	289.0	441	
3	04/18/13	25	20	4	23.2	23.7	4.88	6.49	389.0	183.0	227	
4	04/18/13	25	2	4	23.3	23.6	6.68	6.56	354.0	168.0	254	
5	04/18/13	25	32	1	23.0	23.6	7.16	6.49	794.0	381.0	570	
6	04/18/13	25	11	3	23.1	23.4	10.45	6.71	410.0	194.0	293	
7	04/18/13	25	22	2	23.2	23.6	9.46	6.64	849.0	410.0	608	
8	04/18/13	25	12	1	23.6	23.7	8.81	6.38	483.0	229.0	347	
9	04/18/13	25	31	3	23.4	23.4	10.13	6.94	614.0	297.0	447	
10	04/18/13	25	33	3	23.4	23.7	3.45	6.36	129.6	65.2	94.9	
11	04/18/13	25	36	4	23.4	23.7	9.14	6.74	419.0	198.0	299	
12	04/18/13	25	38	2	23.5	23.7	8.77	6.53	437.0	208.0	311	
1	5/08//2013	28	1	2	23.9	24.6	9.75	6.64	353.0	166.0	252	
2	5/08//2013	28	2	4	24.4	24.6	3.94	6.45	349.0	162.0	247	
3	5/08//2013	28	3	2	24.1	24.8	8.30	6.62	565.0	267.0	404	
4	5/08//2013	28	4	1	24.7	25.0	7.70	6.78	399.0	198.0	295	
5	5/08//2013	28	5	1	24.4	24.6	7.68	6.54	555.0	259.0	388	
6	5/08//2013	28	6	3	24.8	25.0	9.84	6.71	691.0	329.0	497	
7	5/08//2013	28	7	4	24.6	24.9	6.04	6.49	206.0	96.2	131	
8	5/08//2013	28	8	1	24.5	24.9	7.79	6.63	801.0	378.0	570	
9	5/08//2013	28	9	3								
10	5/08//2013	28	10	4	24.4	24.8	9.92	6.68	483.0	229.0	345	
11	5/08//2013	28	11	3	24.3	24.6	8.27	6.40	445.0	197.0	318	
12	5/08//2013	28	12	1	23.9	24.3	7.52	6.56	604.0	284.0	420	
13	5/08//2013	28	13	1	24.6	25.0	8.38	6.56	680.0	330.0	510	
14	5/08//2013	28	14	3	23.3	24.7	8.77	6.70	526.0	260.0	406	
15	5/08//2013	28	15	2	24.9	24.9	7.90	6.78	576.0	271.0	407	
16	5/08//2013	28	16	2	24.6	24.7	5.89	6.31	157.9	74.7	115	
17	5/08//2013	28	17	2	24.6	24.6	7.01	6.60	431.0	204.0	309	
18	5/08//2013	28	18	4	24.7	25.0	7.70	6.78	399.0	198.0	295	
19	5/08//2013	28	19	1	23.7	25.0	8.26	6.65	541.0	316.0	389	
20	5/08//2013	28	20	4	25.2	24.8	4.54	6.33	357.0	164.0	260	
21	5/09//2013	28	21	3	24.4	24.7	7.79	6.92	252.0	118.0	178	
22	5/09//2013	28	22	2	24.1	24.6	8.37	6.66	895.0	426.0	890	
23	5/09//2013	28	23	3	24.8	24.8	7.59	6.45	531.0	247.0	379	
24	5/09//2013	28	24	1	24.9	24.9	8.92	6.75	575.0	273.0	414	
25	5/09//2013	28	25	1	23.4	24.7	9.57	6.54	507.0	240.0	360	
26	5/09//2013	28	26	4	23.7	24.7	7.13	7.14	99.4	49.4	67.2	
27	5/09//2013	28	27	2	24.7	25.2	7.31	6.76	340.0	159.0	236	
28	5/09//2013	28	28	3	24.2	24.5	6.26	6.62	109.2	53.7	74.6	
29	5/09//2013	28	29	4	23.9	24.8	6.63	6.11	93.0	50.0	70.5	
30	5/09//2013	28	30	1	24.8	24.9	7.23	6.45	706.0	339.0	496	
31	5/09//2013	28	31	3	24.8	25.1	10.30	6.85	590.0	277.0	415	
32	5/09//2013	28	32	1	24.7	24.9	7.37	6.20	799.0	383.0	569	
33	5/09//2013	28	33	3	24.9	24.7	3.47	6.42	119.1	58.3	85.9	
34	5/09//2013	28	34	2	24.7	25.0	9.07	6.68	864.0	367.0	611	

35	5/09//2013	28	35	4	24.9	24.9	7.97	6.47	541.0	263.0	395
36	5/09//2013	28	36	4	24.5	24.7	7.78	6.60	433.0	206.0	312
37	5/09//2013	28	37	2	24.9	24.8	8.92	6.50	545.0	255.0	390
38	5/09//2013	28	38	2	24.8	25.0	7.80	6.64	513.0	239.0	365
39	5/09//2013	28	39	3	24.8	25.0	6.11	6.55	227.0	107.0	159
40	5/09//2013	28	40	4	24.7	24.8	7.25	6.55	425.0	198.0	297
Min					19.9	20.0	3.5	6.1	93.0	49.4	67.2
Max					25.2	25.2	11.5	8.0	895.0	453.0	890.0
Mea n					22.2	22.4	8.5	6.7	508.3	248.8	356.2
S.D.					1.8	1.4	1.8	0.3	195.2	95.2	152.4

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